

Fe(II)-catalyzed transformation of Fe (hydr)oxides in particle-size soil organic matter from amended agricultural soils

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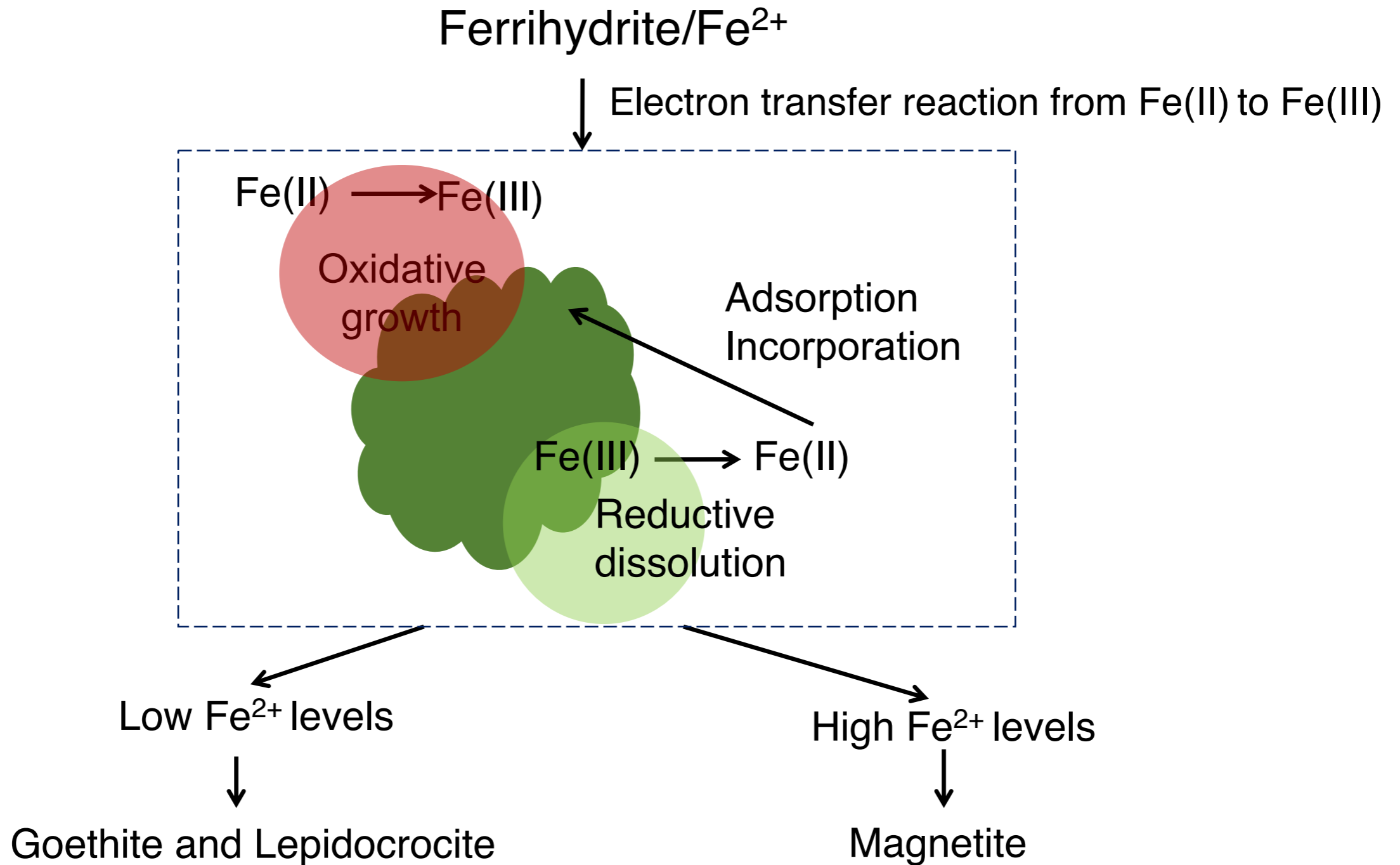
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01 AQUEOUS Fe(II)-CATALYZED FERRIHYDRITE TRANSFORMATION



- Which are the effects of **organic matter** on the extent and pathway of **Fe²⁺-catalyzed transformation**?
- The Fe²⁺-catalyzed transformation is an unexplored pathway for C mobilization or sequestration.

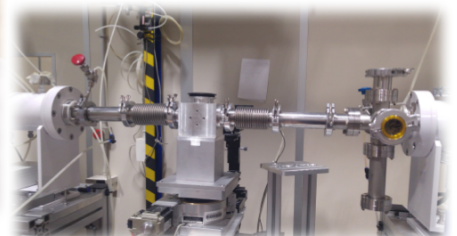
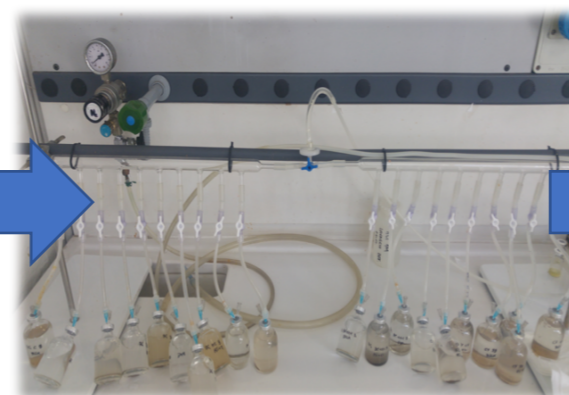
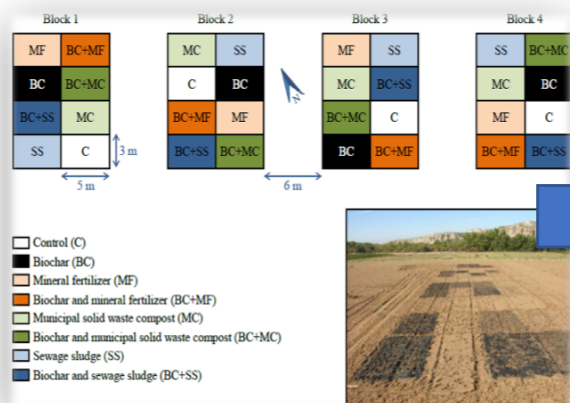
02 OBJECTIVES

HYPOTHESIS:

Soil organic matter (SOM) quantity, quality, and distribution between different pools can affect Fe(III) oxides transformation under reducing conditions by altering Fe atom exchange kinetics.

MAIN OBJECTIVES:

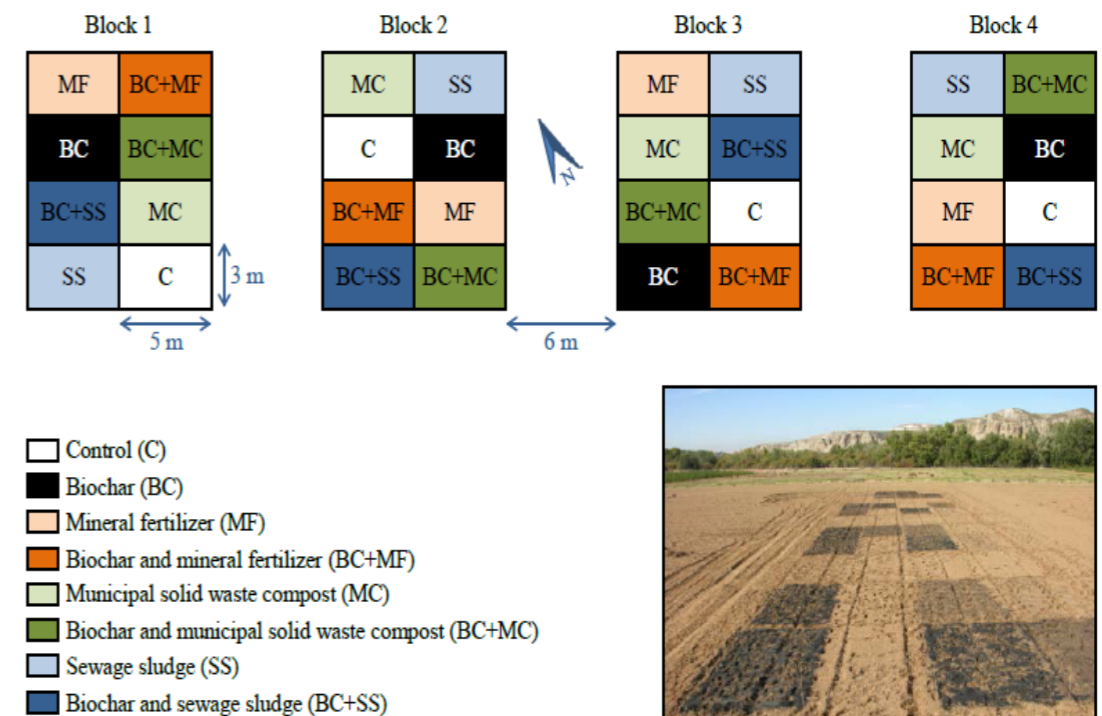
- To determine the effect of **organic amendments** on the **Fe(II)-induced abiotic transformation of Fe(III) minerals** in an **agricultural soil (simulated temporary anoxia)**.
- To investigate the influence of **Fe-OM associations** on mineral **transformations across particle-size SOM pools**.



03 EXPERIMENTAL DESIGN: SOIL SAMPLES

Fe speciation in SOM pools under agricultural soils subjected to biochar and organic fertilizers amendments.

- Unamended agricultural soil (**UN**).
- Municipal solid waste amended soil (**MC**).
- Biochar amended soil (**BC**).
- Biochar and municipal solid waste amended soil (**BC+MC**).
- Bulk, fine sand (**FSa**) and fine silt plus clay (**FSi+Cl**) fractions.



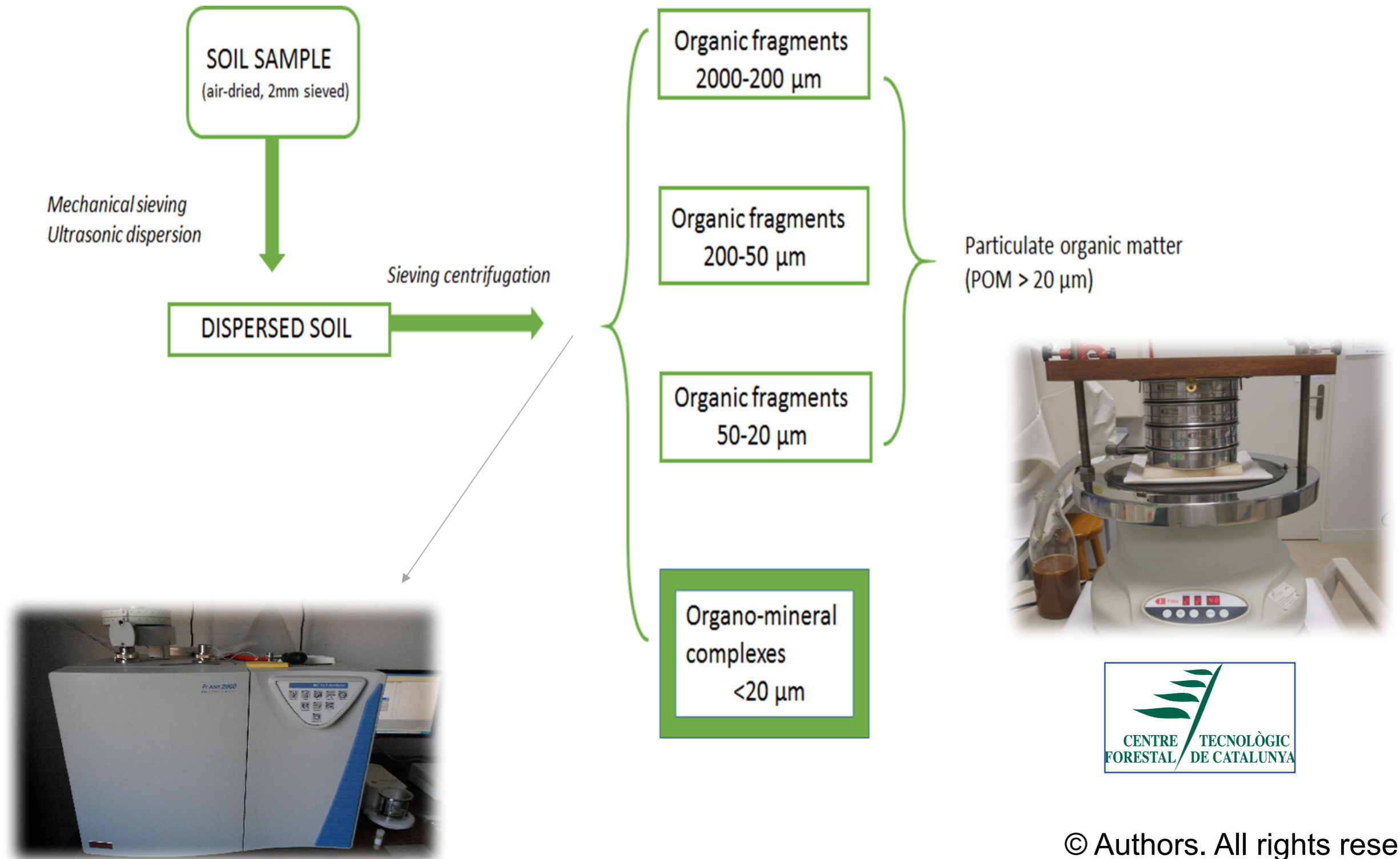
Plaza et al. 2016

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04 PHYSICAL FRACTIONATION



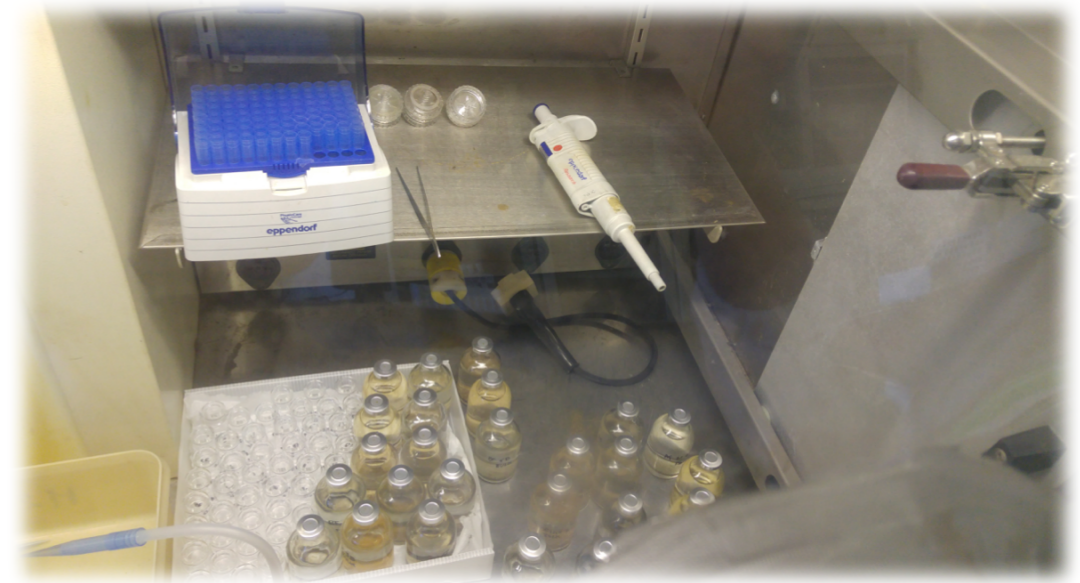
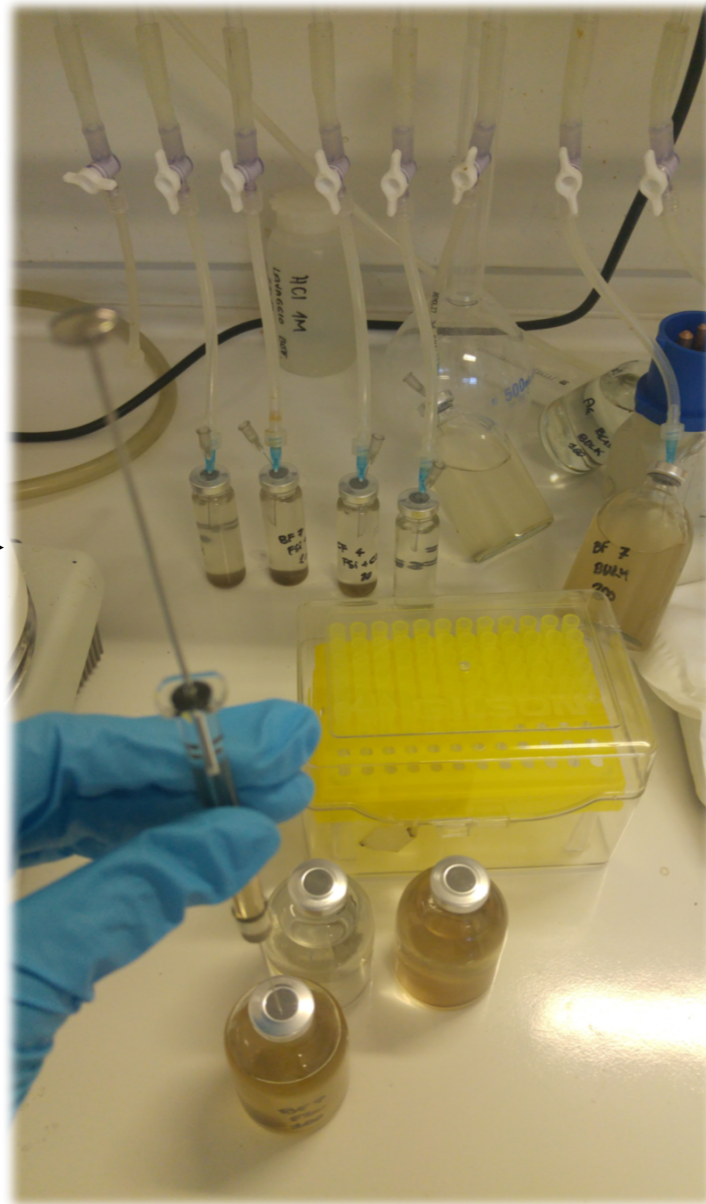
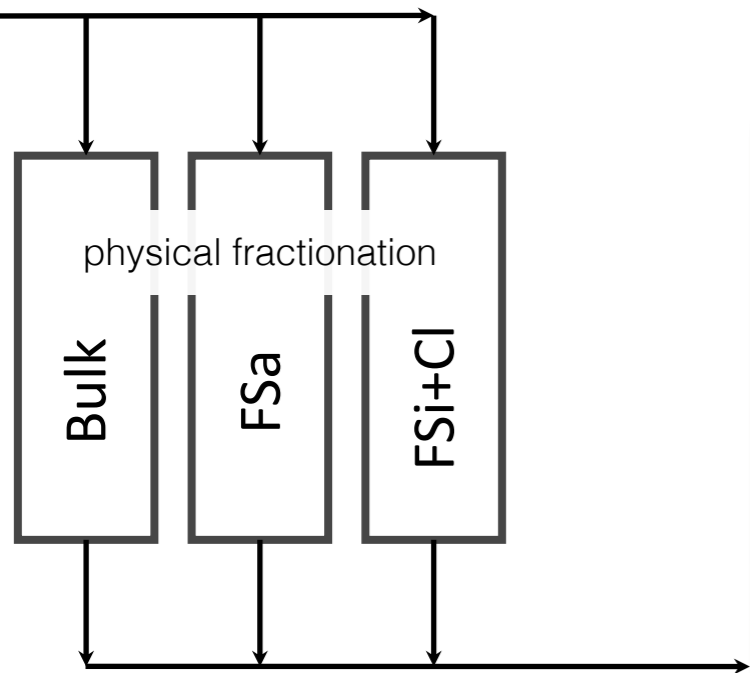
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Lopez-Sangil and Rovira, 2013

Giannetta et al. 2019



05 Fe(II) SPIKING

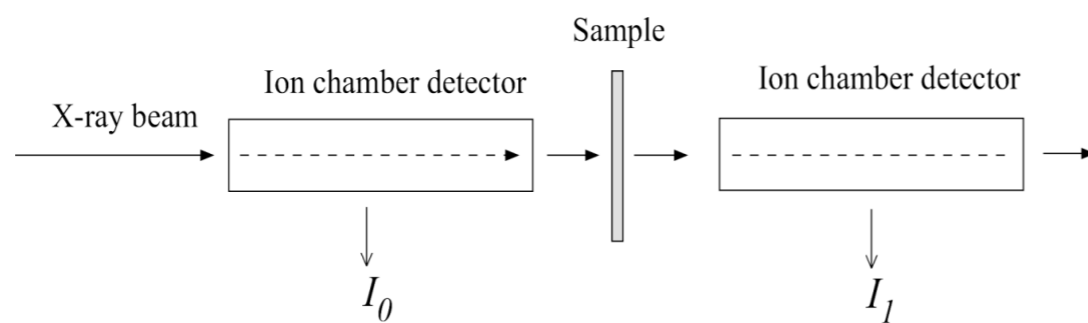
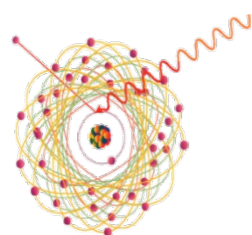


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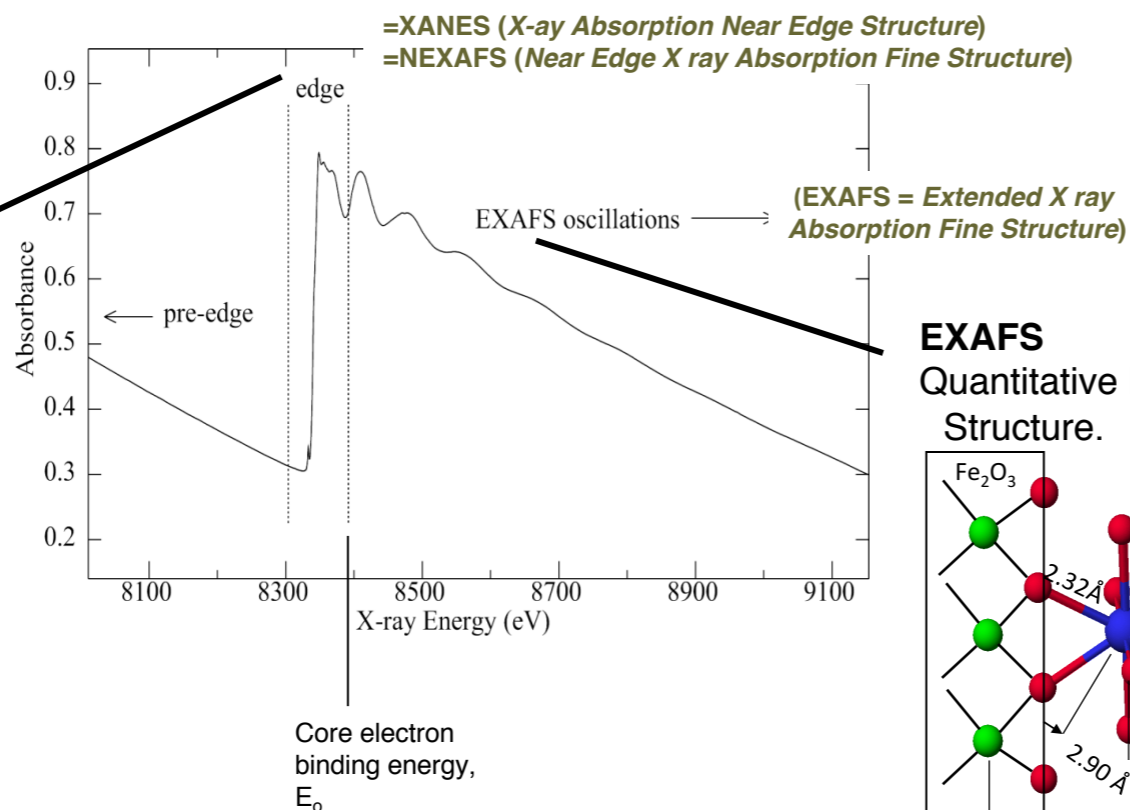


06 Fe SPECIATION: Fe EXAFS

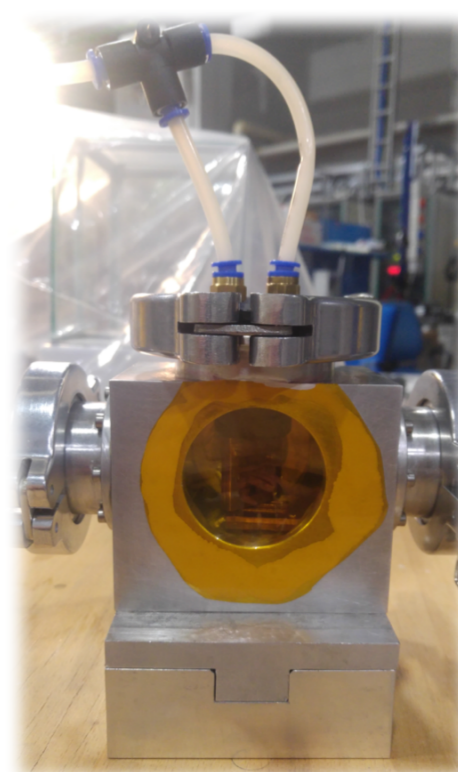
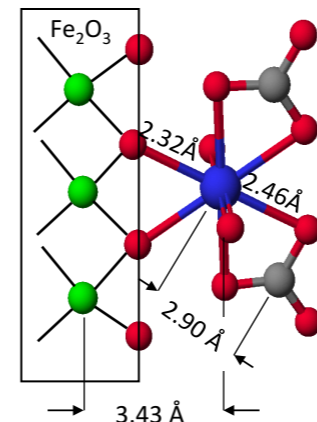
Basic Experiment :



XANES / NEXAFS
Oxidation state,
Molecular structure,
Electronic structure.



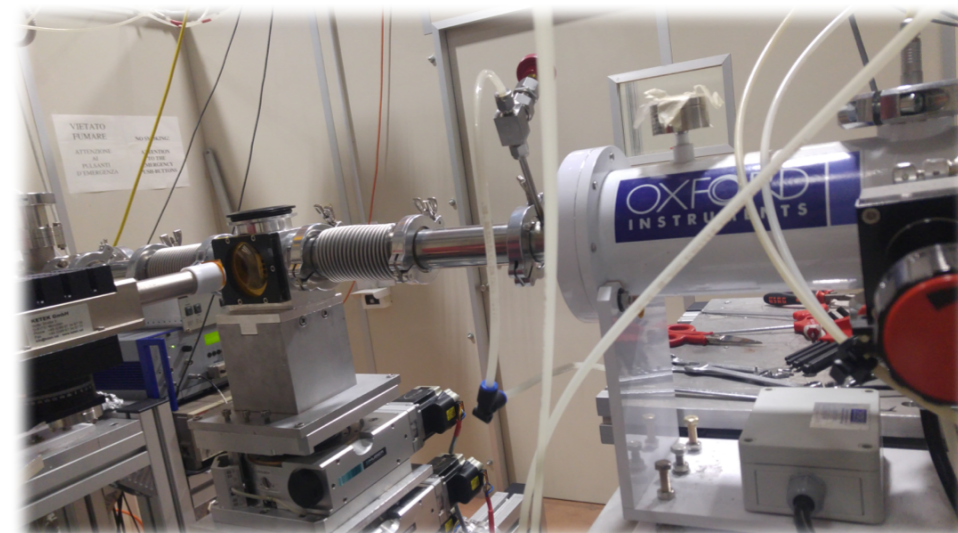
EXAFS
Quantitative Local
Structure.




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Experimental station: XAFS.

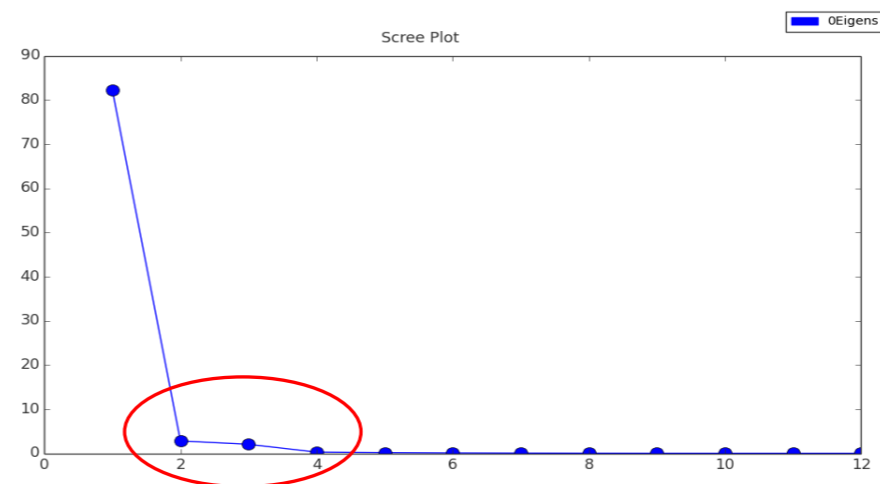


07 RESULTS: Fe EXAFS Bulk soils

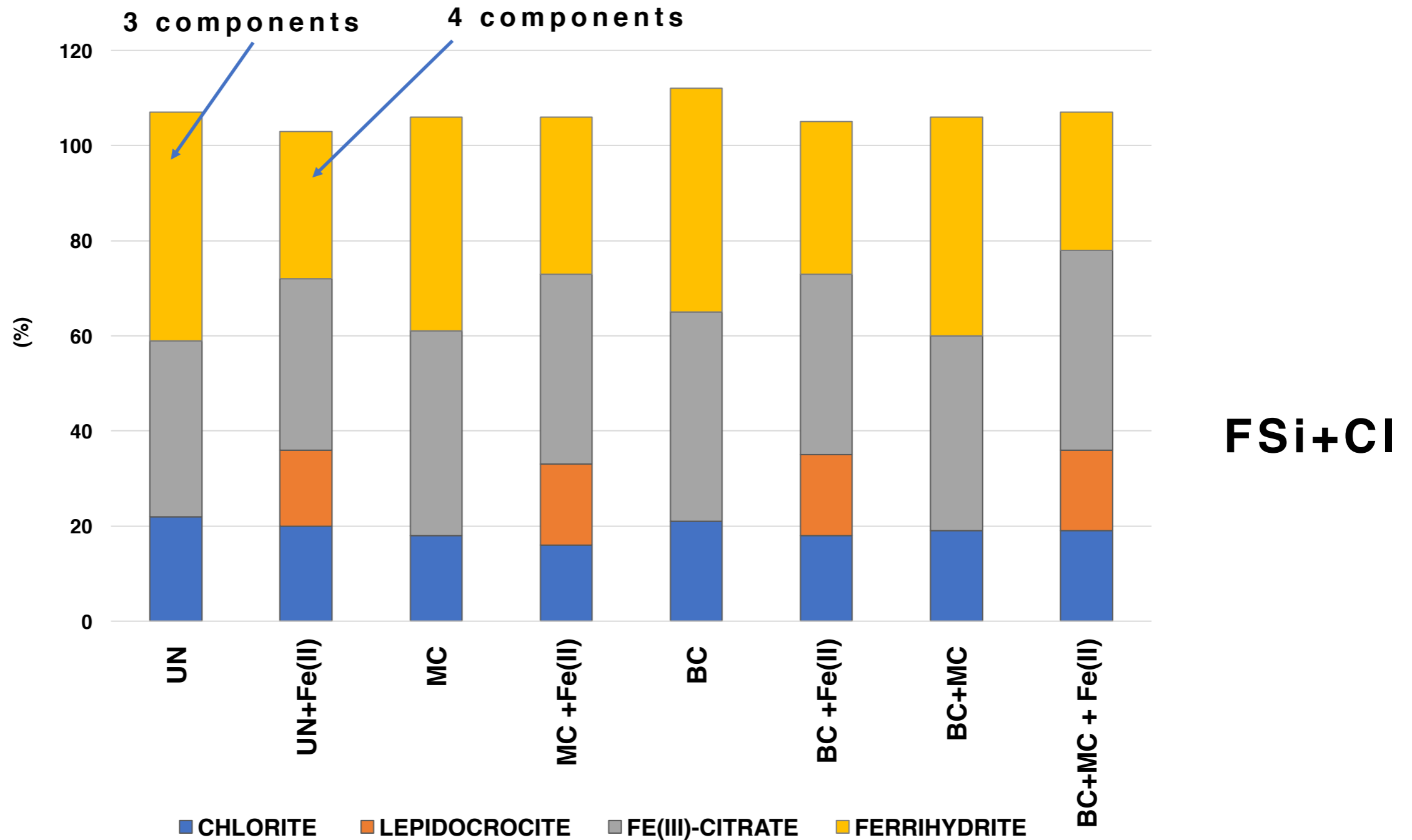


	n. components	COMPONENT 1	COMPONENT 2	COMPONENT 3	COMPONENT 4
UN	3	CHLORITE	LEPIDOCROCITE	FE(III)-CITRATE	
MC	3 and 4	CHLORITE	LEPIDOCROCITE	FE(III)-CITRATE	FERRIHYDRITE
BC	3	CHLORITE	LEPIDOCROCITE	FE(III)-CITRATE	
BC+MC	4	CHLORITE	LEPIDOCROCITE	FE(III)-CITRATE	FERRIHYDRITE

- **PCA**
- **TT**
- **SPOIL**
- **R-values**
- **F-test**
- **LCF**



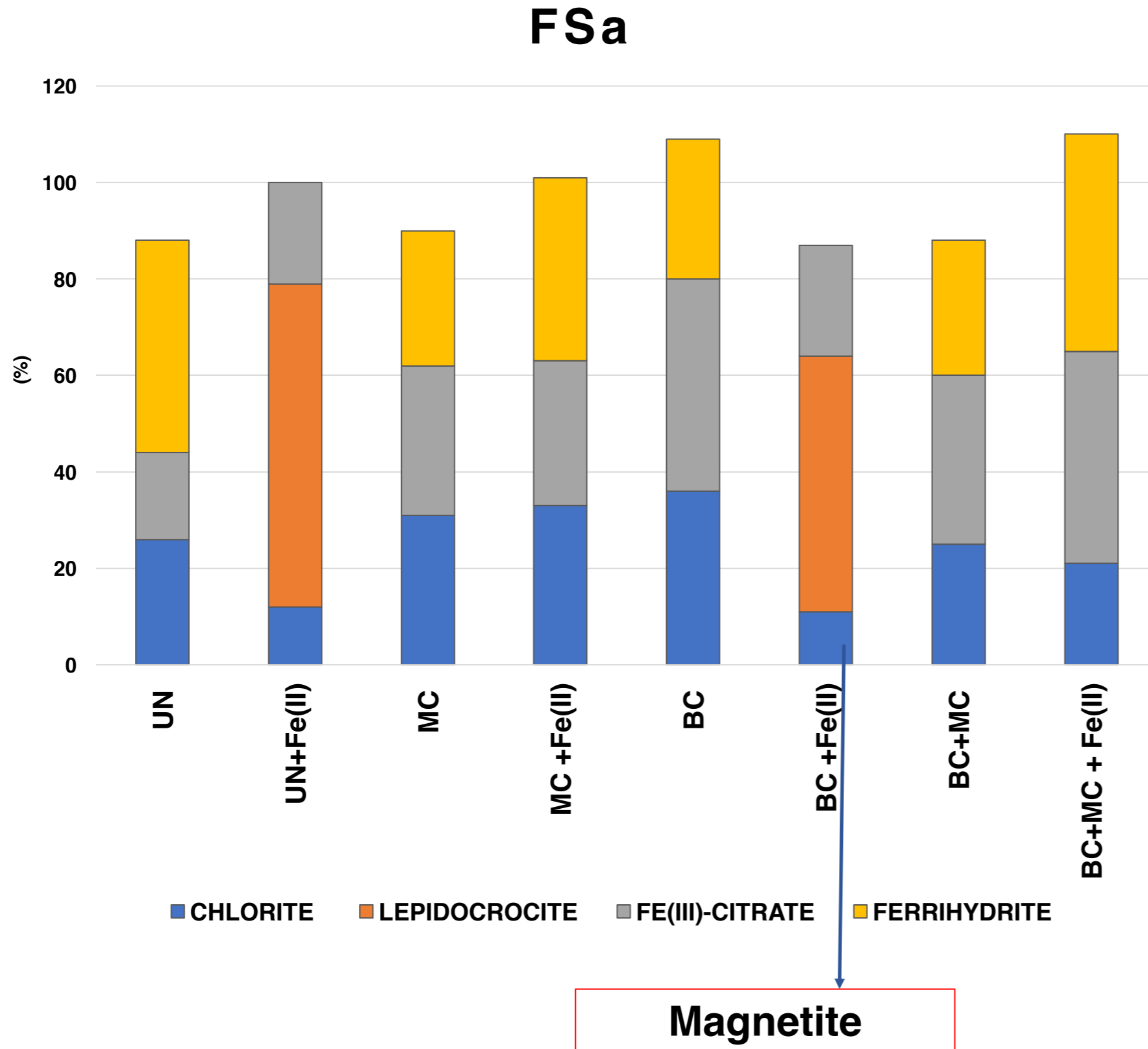
08 RESULTS: Fe(II)-CATALYZED OXIDE TRANSFORMATION IN THE FSi+Cl FRACTION



- FSi+Cl fraction before Fe(II) addition: 3 components.
- FSi+Cl fraction after Fe(II) addition: 4 components.

Ferrihydrite was transformed to lepidocrocite. The percentage of lepidocrocite remained stable in both the unamended and amended soils.

09 RESULTS: Fe(II)-CATALYZED OXIDE TRANSFORMATION IN THE FSa FRACTION



UN: lepidocrocite formation.

MC: lepidocrocite formation hindered.

BC: biochar functions as an electron shuttle, thus favoring the reduction of the Fe(III) oxyhydroxides.

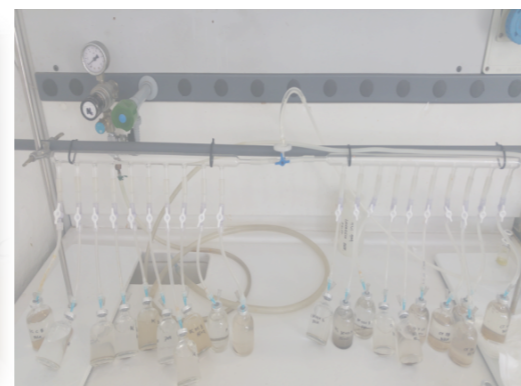
BC+MC: intermediate situation.

FSa fractions represent an understudied pool of SOM reactive to Fe mineral transformation.

10 CONCLUSIONS

SOM quantity, quality and distribution between different pools can affect Fe(III) oxides transformation under reducing conditions by altering Fe atom exchange kinetics.

- The increase in SOM due to organic amendments can contribute to limiting abiotic Fe(II)-catalyzed ferrihydrite transformation.
- This effect of amendment on Fe oxide transformation is less evident in fine with respect to coarse particle-size fractions.
- In this fraction, Fe(II) addition mainly lead to the transformation of ferrihydrite to lepidocrocite, however this depended on organic amendment type.
- With respect to compost, biochar addition favored the formation of both lepidocrocite and magnetite, possibly due to the role of aromatic constituents in electron shuttling.



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