

Characterizing lignin as an atmospherically long-lived ice nucleating biopolymer in immersion freezing

Sophie Bogler¹ ([@BoglerSophie](#)), Nadine Borduas-Dedekind^{1,2}

¹Institute for Biogeochemistry and Pollutant Dynamics, ETH Zürich, Zürich, Switzerland

²Institute for Atmospheric and Climate Science, ETH Zürich, Zürich, Switzerland

1 Introduction

Background:

- Understanding the ice formation pathways in mixed-phase clouds remains crucial for climate models.
- Heterogeneous ice formation ($> -38^{\circ}\text{C}$) relies on ice nucleating particles (INPs).
- Organic aerosols are ubiquitous and a source of INPs, but their complexity hinders detailed INA characterizations.
- The biopolymer lignin is a subcomponent of organic aerosols and serves as proxy for organic matter.

Research Questions:

By analyzing lignin as an INP we want to contribute to the characterization of INA in organic aerosols.

What is lignin's ability to act as an ice nucleating particle in mixed-phase clouds?

What are the effects of...

- A. typical IN laboratory treatments
 - B. atmospheric processing
- ... on lignin's INA?

A. sonication
heating
 H_2O_2

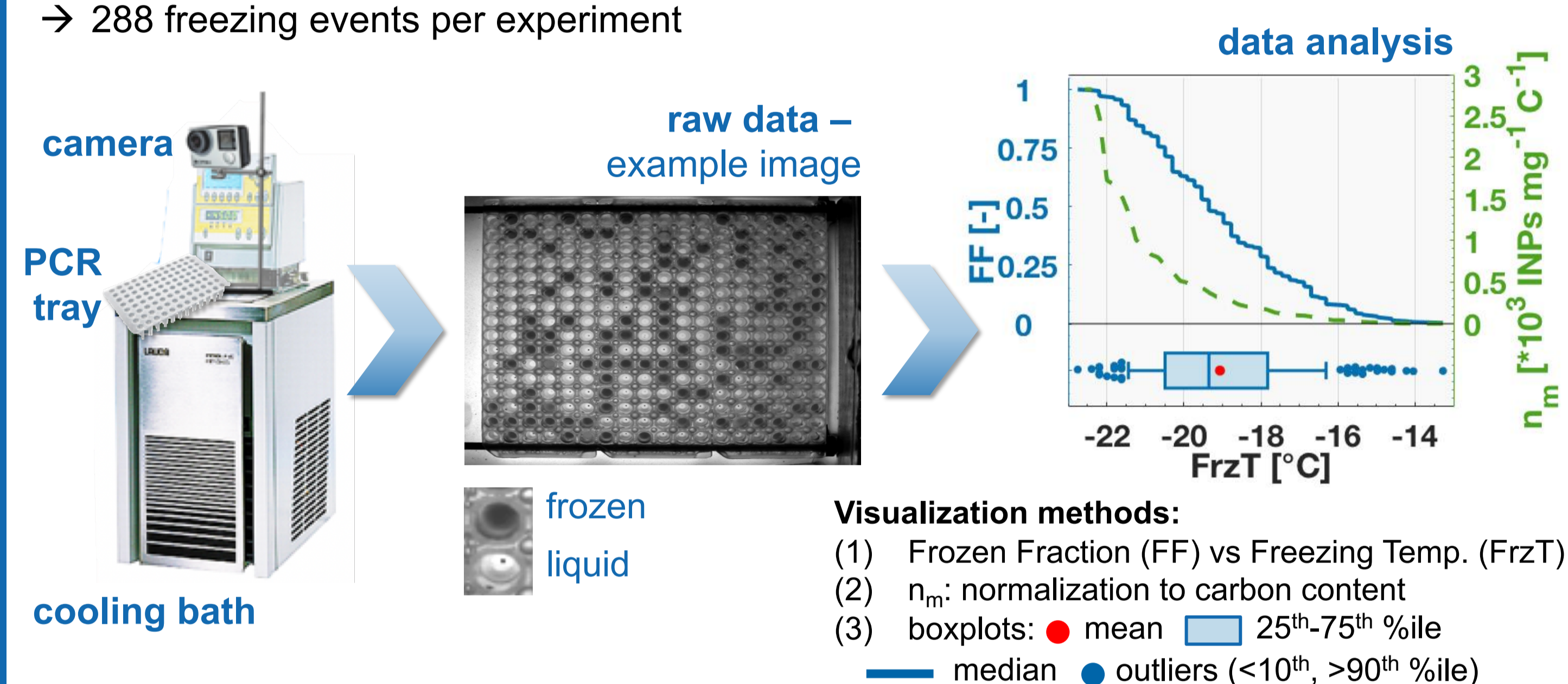
B. ozonation
photochemistry

How does lignin act as INP?

2 Methods – Ice Nucleation Setup

Freezing Ice Nuclei Counter (FINC) (Miller et al., 2020¹)

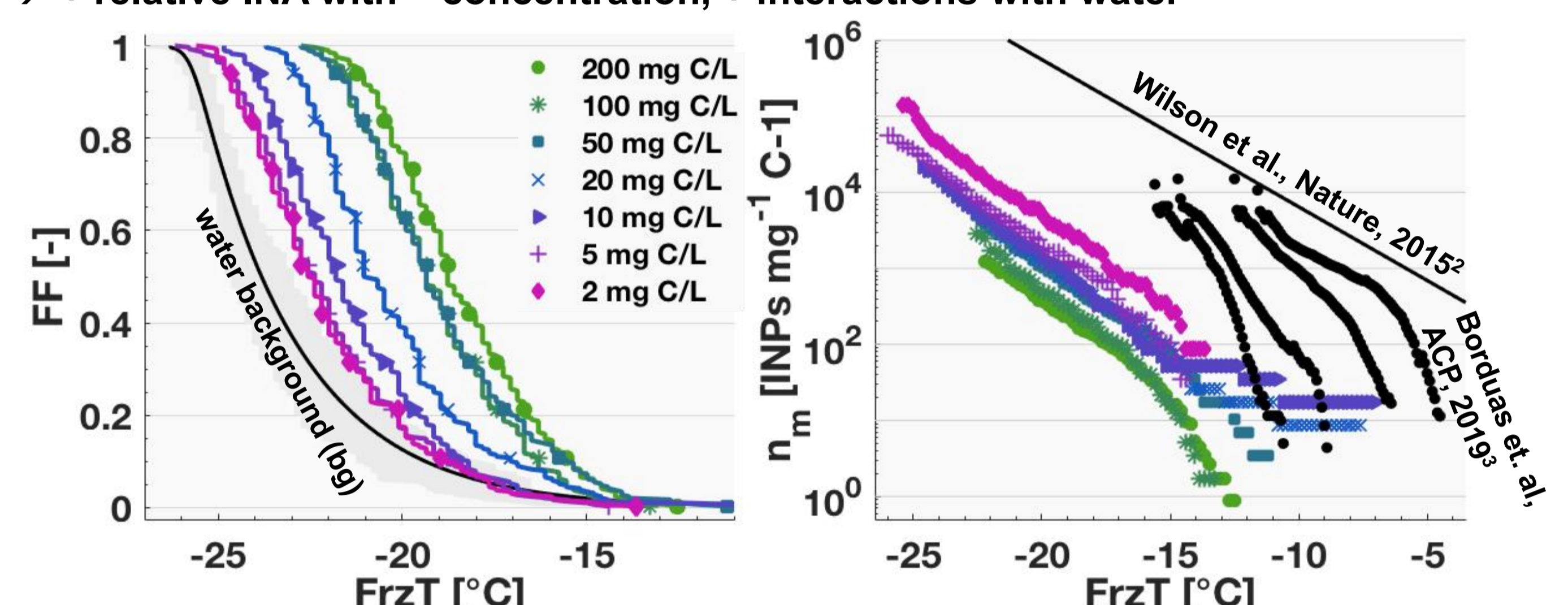
- droplet freezing (20 μL , range 0°C to -26°C)
- 288 freezing events per experiment



3 Lignin's Ice Nucleating Ability

Lignin's INA is relevant for mixed-phase clouds, but lower than INA of complex dissolved organic matter (DOM)

- n_m range from 1 to 10^5 INPs/mg C (-7°C to -26°C)
- 50% FF at -18.7°C for 200 mg C/L
- + relative INA with – concentration, + interactions with water



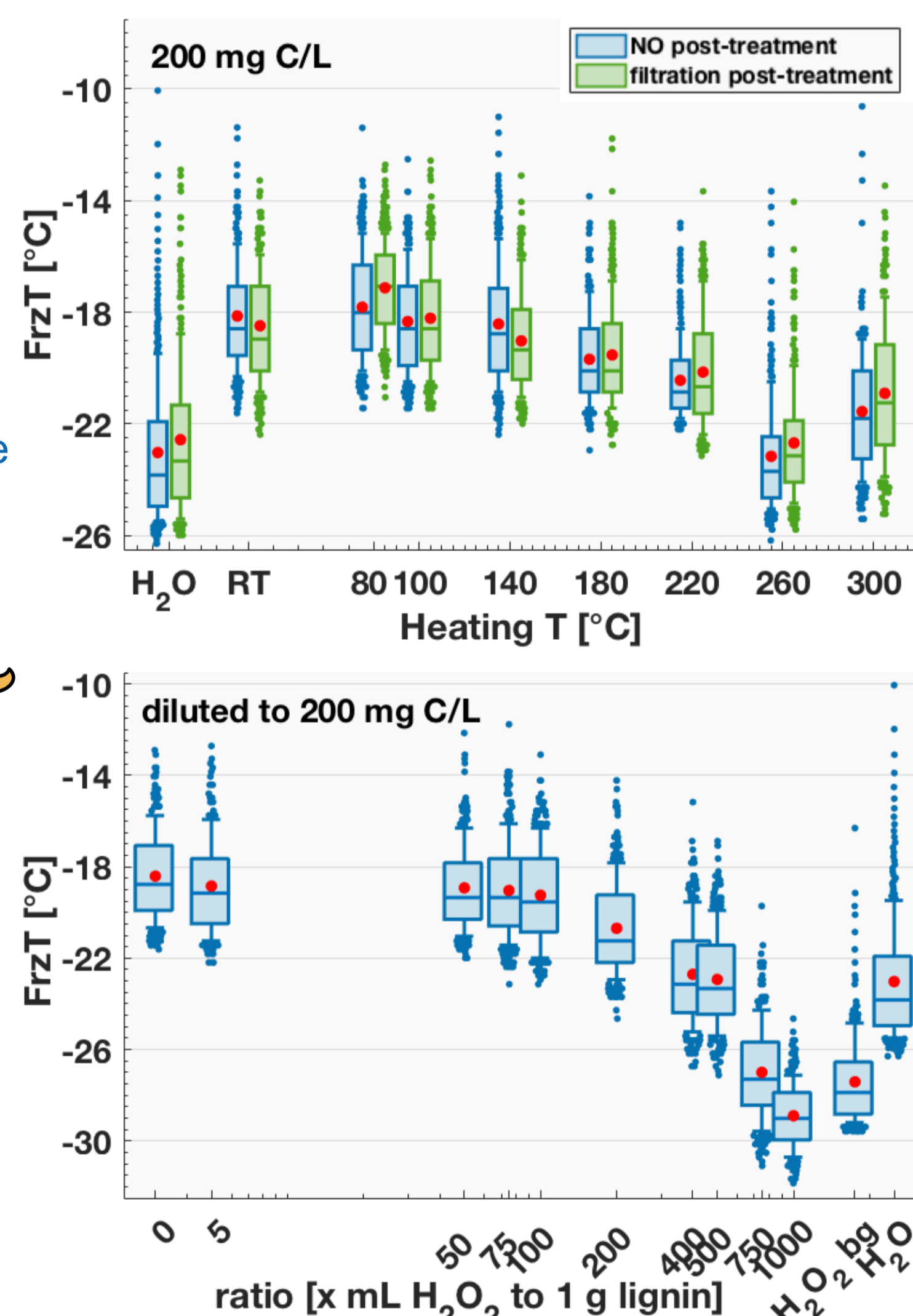
4 Laboratory Treatments

1. Sonication: no change in INA

- extraction tool for particulates on aerosol filters, radical reactions may affect chemical structure (Miljevic et al., AS&T, 2014⁴)

2. Heating: INA decrease

- volatilization of organic material by pyrolysis
- experiment: heating of lignin as powder, then H_2O addition to 200 mg C/L
- – INA with + Heating T, FrzT of H_2O reached after heating to 260°C
- + relative INA after removal of insoluble fraction, as mass loss



3. H_2O_2 : INA decrease

- oxidative removal of organic material
- experiment: reactions with varying ratios of mL H_2O_2 (35 w%) : g lignin, then dilution with H_2O to 200 mg C/L
- – INA with + ratio of H_2O_2 to lignin, FrzT of background with H_2O reached after ratio 400:1

5 Atmospheric Processing

1. Ozonation: no change in INA

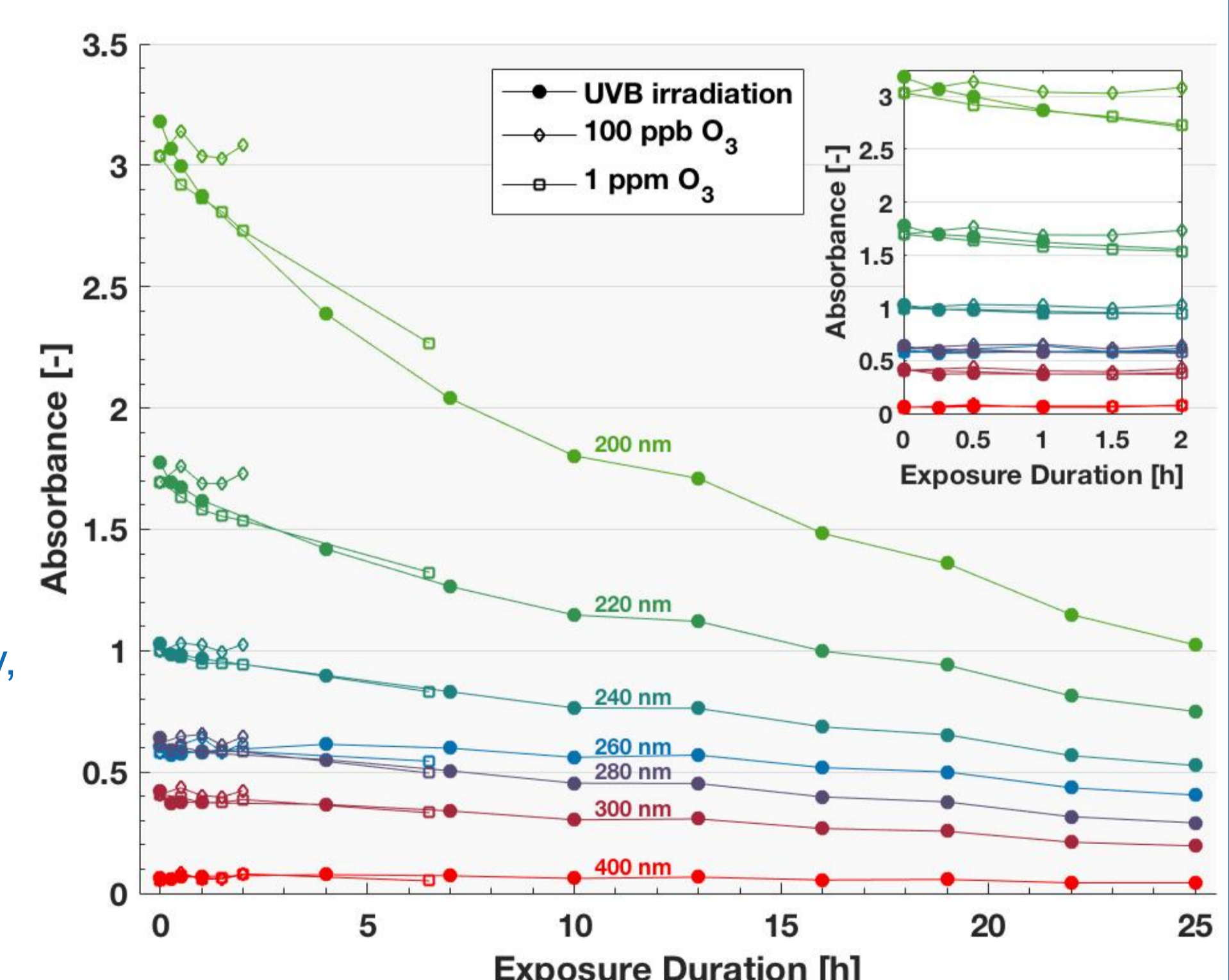
- experiment: bubbling of 0.1 L/min O_3 (100 ppb – 1 ppm) through bulk solution of 20 mg C/L up to 6.5 h

2. Photochemistry: minimal INA decrease

- experiment: up to 25 h (6.5 days atmospheric equivalent) of UVB irradiation on 20 mg C/L solution
- small – trend in lignin's INA through photochemistry, however less than observed for complex DOM (Borduas et al., ACP, 2019³)
- max. -1°C decrease at 50% FF after 22 h compared to 0 h

Measurement of UV/Vis spectra (200 – 400 nm)

- – absorbance after exposure to ozone or UVB irradiation
- decay of chromophoric components in lignin's structure
- the IN activity is mostly retained despite this decay, so these chromophores are not decisive for the biopolymer's INA



6 Conclusions

Lignin's IN ability:

1. is relevant for mixed-phase clouds

- n_m from 1 to 10^5 INPs/mg C (-7°C to -26°C)

2. is robust and lasts the aerosols' lifetime

- INA decrease to the water background requires harsh laboratory treatments ($\geq 260^{\circ}\text{C}$, [400:1] mL H_2O_2 : g lignin)

- minimal change through atmospheric processing (exposed to up to 1 ppm O_3 , to up to 25 h of UVB light with an atmospheric equivalent of 6.5 days sunlight)

3. is retained after structural changes in the biopolymer's chromophoric structure

- decrease of UV/Vis absorption (200 – 400 nm) after exposure to UVB light or ozone

References:

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