

# Seasonal variability of submicron aerosol acidity at a coastal site in the Eastern Mediterranean

Anna Maria Neroladaki<sup>1,\*</sup>, Iasonas Stavroulas<sup>1,2</sup>, Irini Tsiodra<sup>1,3</sup>, Pavlos Zarmpas<sup>1</sup>, Stelios Myriokefalitakis<sup>2</sup>, Athanasios Nenes<sup>3,4</sup>, Nikos Mihalopoulos<sup>1,2</sup>, and Maria Kanakidou<sup>1,3,\*</sup>

<sup>1</sup>Environmental Chemical Processes Laboratory, Department of Chemistry, University of Crete, Heraklion, Greece <sup>2</sup>Institute for Environmental Research and Sustainable Development, National Observatory of Athens (NOA), GR-15236 Palea Penteli, Greece <sup>3</sup>Center for Studies of Air Quality and Climate Change, Institute for Chemical Engineering Sciences, Foundation for Research and Technology Hellas, Patras, Greece <sup>4</sup>Laboratory of Atmospheric Processes and their Impacts, School of Architecture, Civil and Environmental Engineering, Ecole Polytechnique Fédérale de Lausanne, CH-1015, Lausanne,

Switzerland \*mariak@uoc.gr; amwateroil2@gmail.com

## Introduction

Importance: Aerosol pH effects on:

- chemical composition of atmospheric particles, including aerosol water
- gas-particle partitioning of semi-volatile gases
- solubility and concentration of toxic forms of trace metals

• Climate

(Pye et al., ACP, 2020)



How is it determined?

- Direct measurements of particle pH: challenging
- Thermodynamic model calculations based on aerosol and air composition measurements: mostly used

#### Aim of this study:

• Calculate the submicron aerosol water and pH in the area (Finokalia atmospheric observatory in the eastern Mediterranean) • Examine the seasonality of the derived pH and its drivers

#### Measurement site:

Finokalia atmospheric observation station in Crete. Greece (35°20'N. 25°40'E; 250 m a.s.l.). http://finokalia.chemistry.uoc.gr/





#### Model results-Sensitivity to crustals and NH<sub>3</sub>



### Model results - Seasonality - Aerosol pH



#### Sensitivity of pH to crustal elements and NH<sub>3</sub> levels



# Conclusions

### **Submicron aerosol at Finokalia**

• highly acidic with an average pH value of  $1.09 \pm 0.71$ 

• Water associate with organics has a minor effect on the aerosol pH (average increase by 0.07 pH unit).

-0,5 Feb Mar Apr May June July Aug Sep Oct Nov Dec

pH\_Wi: calculated based on Wi pH\_Wi+Wo: calculated based on Wi+Wo

- Accounting for organic water (Wo) increases the pH by 0.05 to 0.1 unit.  $\checkmark$
- Aerosol pH shows summertime minimum and wintertime maximum (1.8 pH unit ✓ difference).

### Acknowledgements

We acknowledge support of this work by the project "PANhellenic infrastructure for Atmospheric Composition and climatE chAnge" (MIS 5021516) which is implemented under the Action "Reinforcement of the Research and Innovation Infrastructure", funded by the Operational Programme "Competitiveness, Entrepreneurship and Innovation" (NSRF 2014-2020) and cofinanced by Greece and the European Union (European Regional Development Fund). <u>https://panacea-ri.gr/</u>

• Aerosol pH shows clear seasonality with the highest values in winter and the lowest in summer.

• Interpreting the crustal species in the aerosol pH prediction: average increase by 0.83 pH units accompanied by an increase in particle phase nitrate.

**Environmental Chemical Processes Laboratory,** Department of Chemistry, University of Crete, Greece mariak@uoc.gr