



**Barcelona  
Supercomputing  
Center**  
*Centro Nacional de Supercomputación*



# Modelling nitric acid uptake by mineral dust using parameterizations of different complexity

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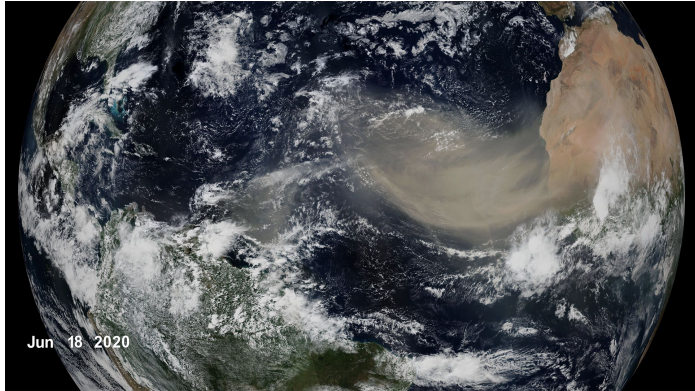
**11622**



**AXA  
Research Fund**

**May 24th 2022**

# (1/4) Introduction: Dust heterogeneous chemistry



Source: NASA's Goddard Space Flight Center (December 2021)



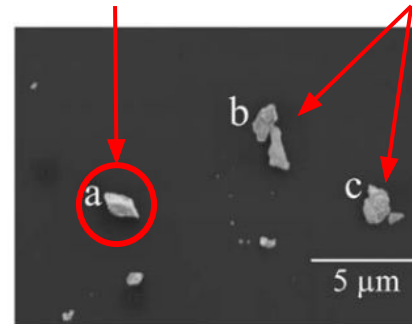
Source: oizom.com (May 2022)



## Airborne dust heterogeneous chemical reactions dependent on mineralogy

$\text{CaMg}(\text{CO}_3)_2$  (Dolomite)

AlSi

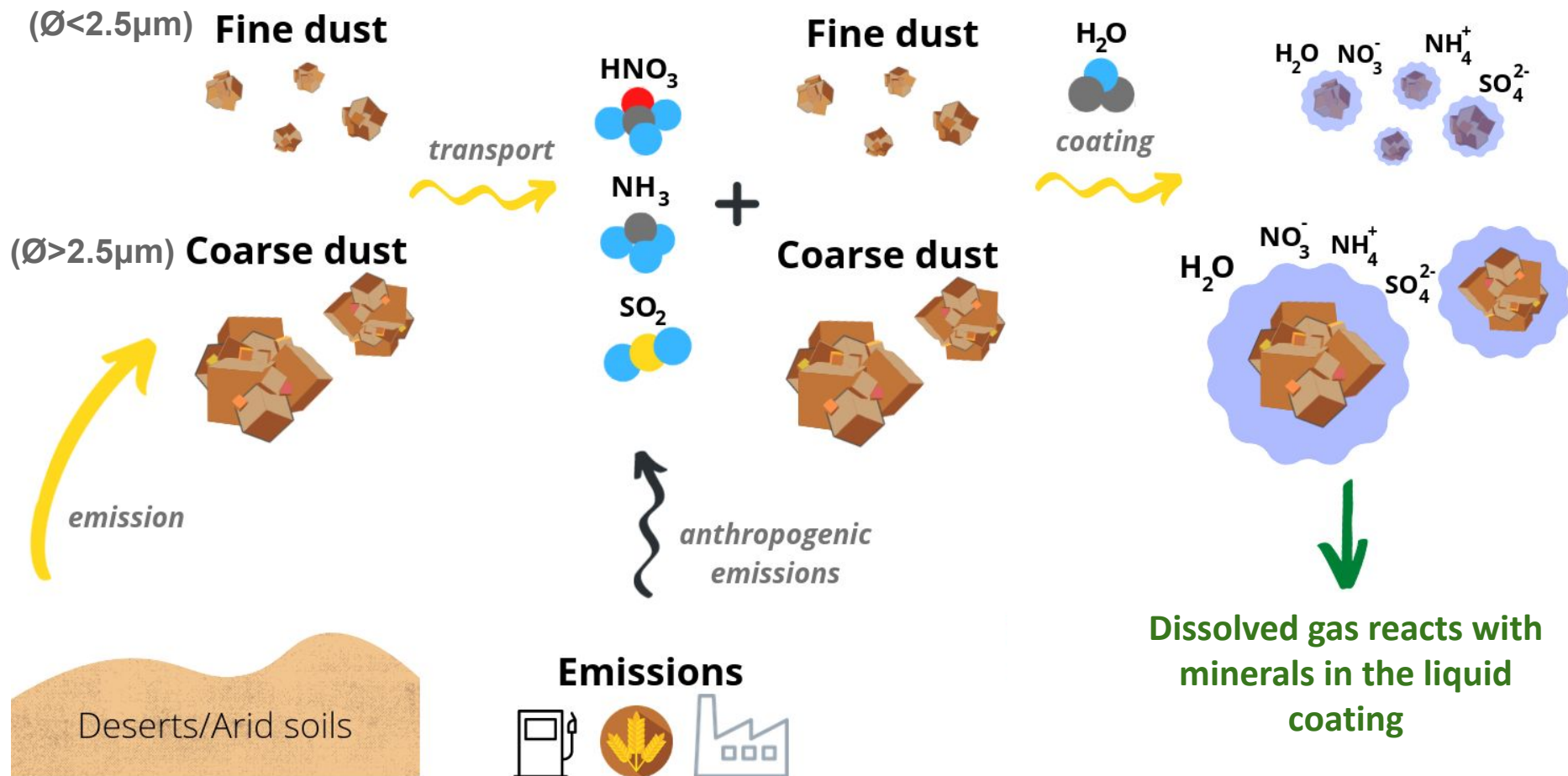


$\text{HNO}_3$   
38% RH



Source: Krueger et al., 2004

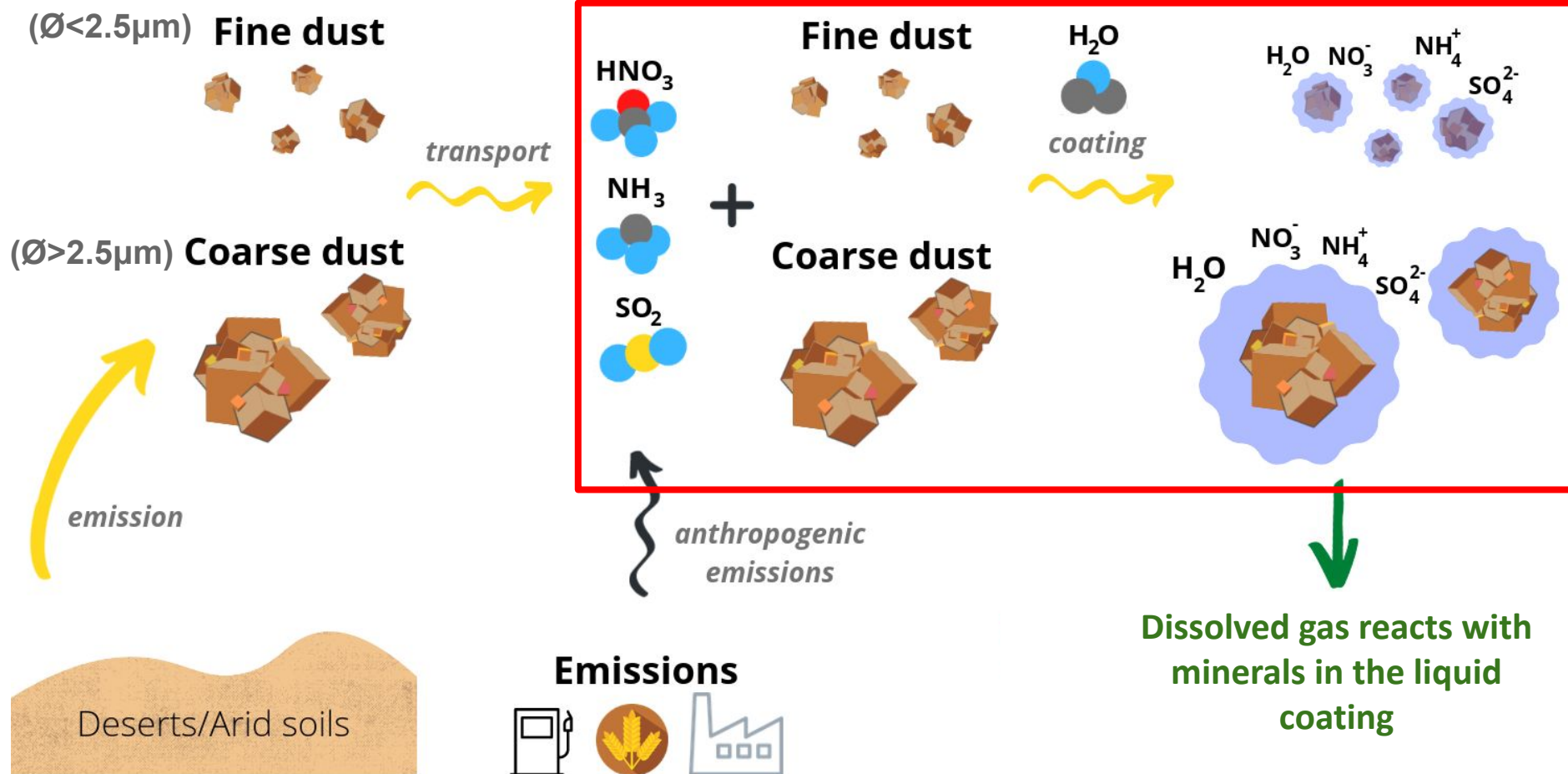
# (1/4) Introduction: Dust heterogeneous chemistry



# (1/4) Introduction: Dust heterogeneous chemistry



How to simulate this?



# (2/4) Implementation in MONARCH model



**MONARCH**

Atmospheric chemistry model

Multiscale Online Non-hydrostatic

AtmospheRe Chemistry

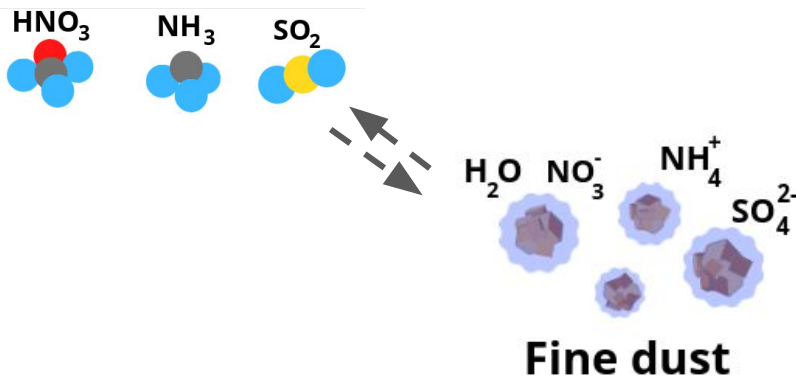
1

**FINE DUST**  
( $\varnothing < 2.5 \mu\text{m}$ )

## 1. Thermodynamic equilibrium with:

1.1. **EQSAM v.03d** (Metzger et al., 2002)

2.1. **ISORROPIA II** (Fountoukis and Nenes, 2007)



2

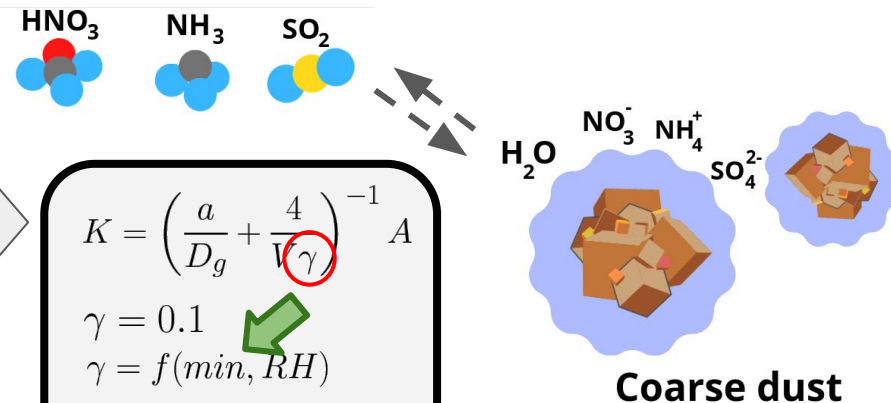
**COARSE DUST**  
( $\varnothing > 2.5 \mu\text{m}$ )

## 2.1. Irreversible uptake reaction with uptake coefficients = **0.1** or **f(min,RH)**

(Hauglustaine et al., 2014)

## 2.2. Thermodynamic equilibrium (kinetically limited) with ISORROPIA II

(Pringle et al., 2010)



Gas left

$$K = \left( \frac{a}{D_g} + \frac{4}{V\gamma} \right)^{-1} A$$

$\gamma = 0.1$   
 $\gamma = f(\text{min}, RH)$

Vlasenko et al. (2006)

# (3/4) Evaluation: Dust event - Wang et al. (2017)



Significant impacts of heterogeneous reactions on the chemical composition and mixing state of dust particles: A case study during dust events over northern China

Zhe Wang<sup>a,b</sup>, Xiaole Pan<sup>b,\*</sup>, Itsushi Uno<sup>a</sup>, Jie Li<sup>b</sup>, Zifa Wang<sup>b</sup>, Xueshun Chen<sup>b</sup>, Pingqing Fu<sup>b</sup>, Ting Yang<sup>b</sup>, Hiroshi Kobayashi<sup>c</sup>, Atsushi Shimizu<sup>d</sup>, Nobuo Sugimoto<sup>d</sup>, Shigekazu Yamamoto<sup>e</sup>

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<sup>c</sup> University of Yamanashi, Yamanashi, Japan

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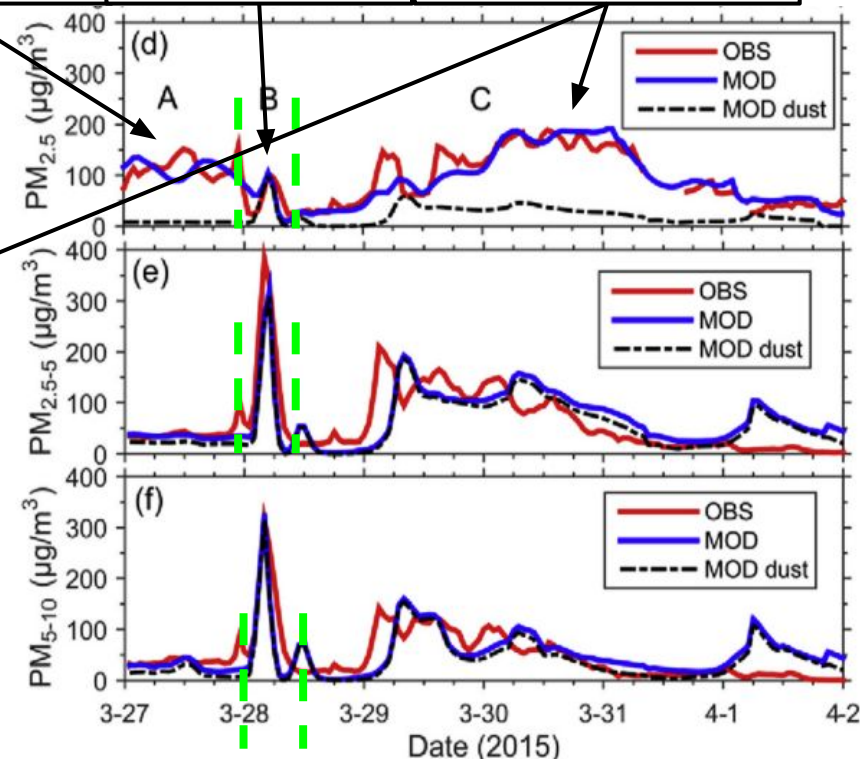
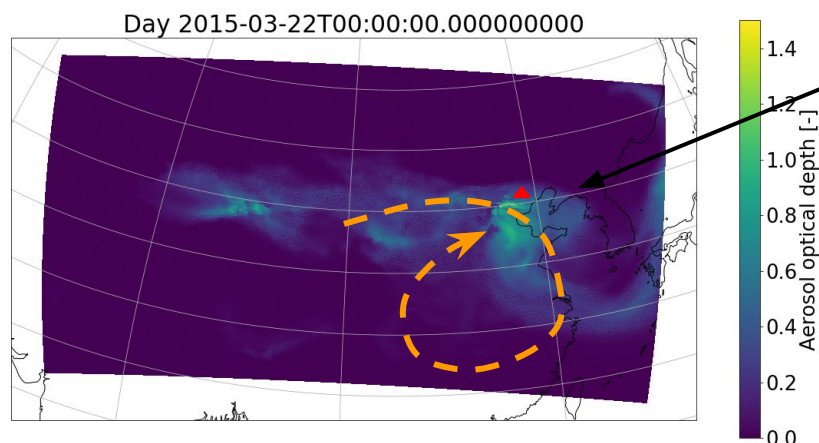
<sup>e</sup> Fukuoka Institute of Health and Environmental Sciences, Fukuoka, Japan



- Period: **March-April 2015**
- Obs.: **Beijing Institute of Atmospheric Physics**
- Model: **NAQPMS**
- Emissions: **MIX inventory**

| Anthropogenic | Pure dust | Dust+Anthropogenic |
|---------------|-----------|--------------------|
|---------------|-----------|--------------------|

Dust AOD (550nm) from MONARCH

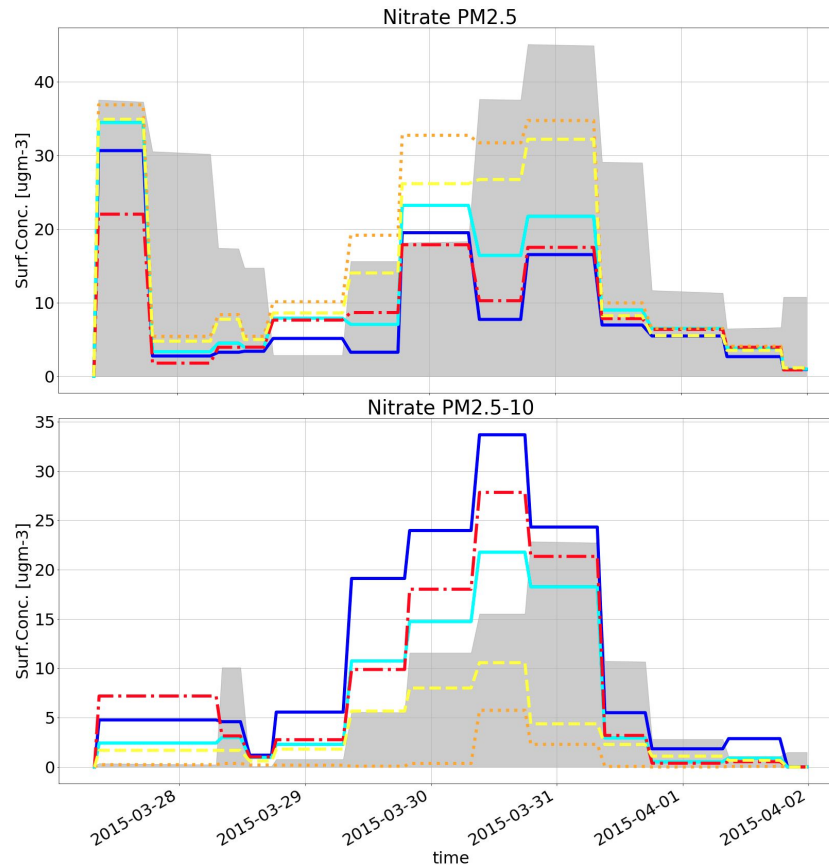


# (4/4) Results and conclusions



## Evaluation with Wang et al. (2017):

- **Dust + anthropogenic** species event over Beijing in March-April 2015.
- Average dust mineralogy: **7.7%  $\text{Ca}^{2+}$** , **2.1%  $\text{K}^+$**  and **1.7%  $\text{Mg}^{2+}$**

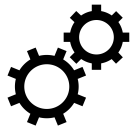


- Results show high variability depending on:
  - 1) Models used (EQSAM/ISORROPIA),
  - 2) Dust mineralogy in fine and coarse modes,
  - 3) Uptake coefficient used in the coarse mode.
- Double use of ISORROPIA refines results.
- Inclusion of dust mineralogy show the extreme complexity of the system.
- The different approximations used cover the range of observations.

# Next steps



Perform **global-scale simulations**...

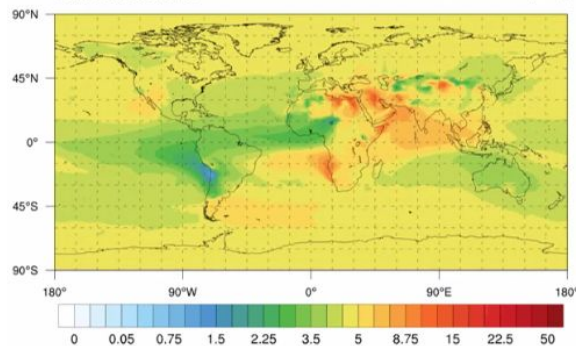


**Configurations** and **parameterizations**

**Calcite surface concentration (%)**

MONARCH 1°x1.4°  
C1999 2006-2008

annual mean calcite  
mineral fraction (%w)



**Dust mineralogical inventories**

... to understand better

✓ **the effect of dust mineralogy on  
dust heterogeneous chemistry**

# Thank you for your attention

This project is part of the ERC Grant:

## FRAGMENT

FRontiers in dust minerAloGical  
coMposition and its Effects upoN climaTe  
(Working package 3)



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**EGU 2022**



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