



# Geochemical monitoring of the geothermal reservoirs using a high-temperature downhole sampler

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### Why downhole sampling?



#### Examples of corrosion and scaling issues

Clogged holes in liner from well in N-Iceland





Corroded liner due to mixing of fluids from two feed zones

ÍSOR's spinner tool damaged after logging a new well to 2350 m depth





Logging tool came black out of the well



Hydrogen attack leads to diminished material properties

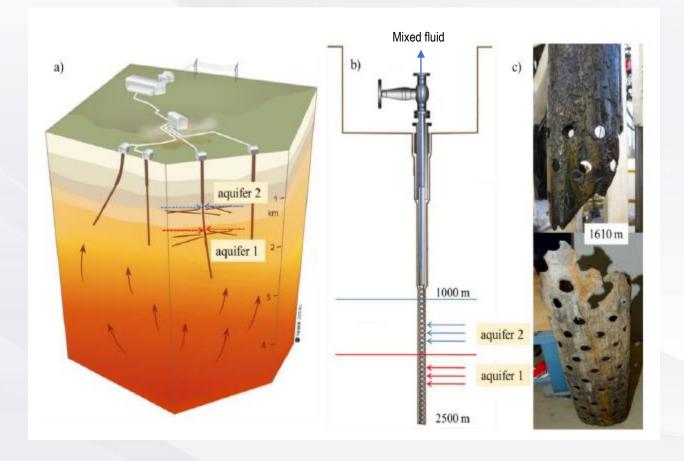




### Why downhole sampling?



- Lack of knowledge about fluid-properties of distinct aquifers feeding a singular well in hot and super-hot geothermal fields leads to long term and high-cost geothermal utilization problems with surface and downhole installations (liner corrosion), and increased greenhouse gas emissions.
- Optimised casing design, requires knowledge on fluid composition at depth and therefore an extension of the P-, T- range at which existing samplers are enabled to collect fluid samples.
- High-temperature wells: Temperatures are typically around 200-300°C
- Ultra-high-temperature wells: Temperatures can be much higher, 400-550°C







### Why design a new sampler?



#### Types of samplers:

- Positive displacement samplers
- Vacuum samplers
- Flow-through samplers
- Pumps
- U-tube

Flow through samplers from Thermochem and Kuster can sample fluids up to 350 °C

→ Great interest in sampling fluids at depth at that are greater than 350 °C



Kuster



Thermochem



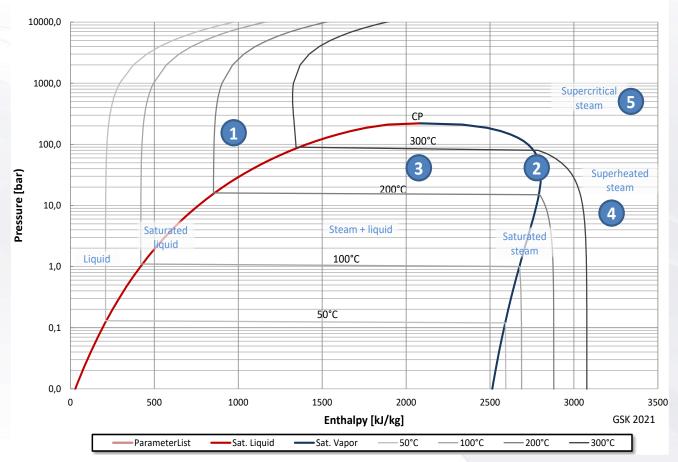




### Sampling Scenarios in HT- to super-HT wells



#### **Pressure-Enthalpy Plot for Water**



- 1. Liquid (low-temperature or saturated liquid)
- 2. Steam (high-temperature saturated steam)
- 3. Two-phase (high-temperature above flashing zone)
- 4. Superheated (one phase steam at all depths within the well)
- 5. One-phase supercritical (one phase steam at all depths within the well)





### **Pressure of Contained Sample**



#### **Design basis and assumptions:**

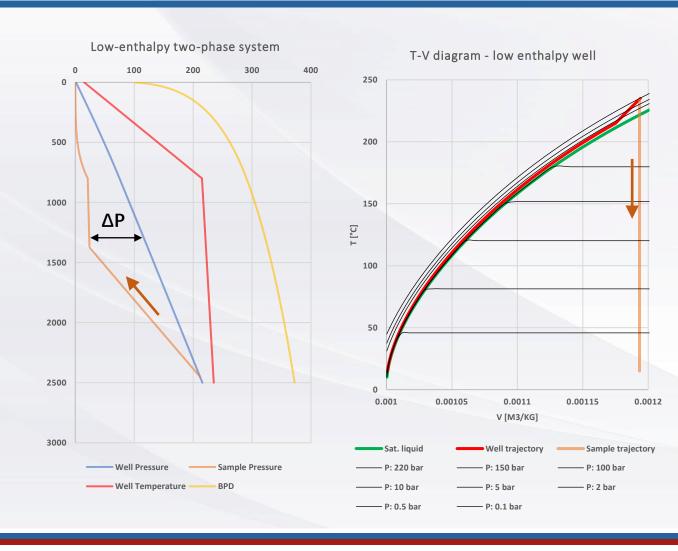
- Sample taken at a certain depth and sample pulled out of the well
- Pressure inside the sample bottle calculated on its way up the well
- Differential pressure is the difference in P\_well and P sample

**Assumption 1**: No NCGs in the fluid – only water

 How does the presence of NCGs effect the pressure conditions?

**Assumption 2**: Thermal equilibrium between the sample and the well fluid at all depths

**Assumption 3**: The thermodynamic process inside the sample bottle when it is pulled out of the well is isochronic, i.e. happens at constant specific volume [m3/kg]





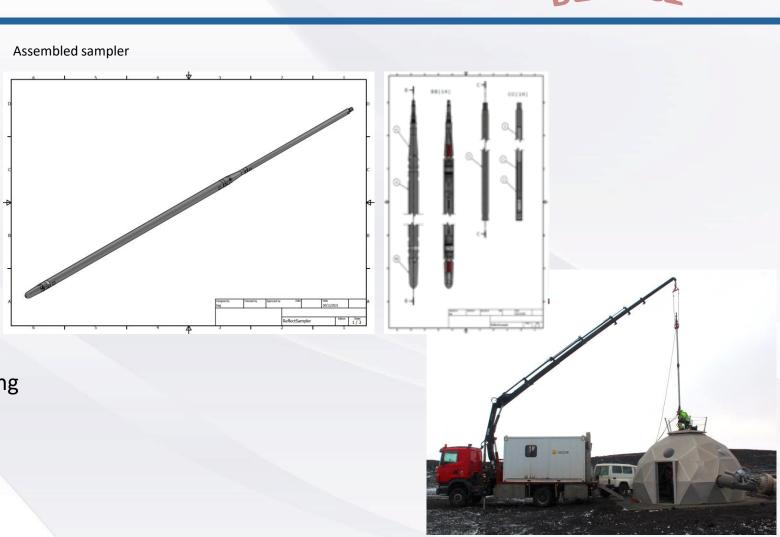
### **Current status – Design and Prototype Build**



- ✓ Sampler design complete
- Sampling volume: >1 Liter
- Ranging temperature limits
  - 230°C Viton o-ring seals
  - 300°C Kalrez o-ring seals
  - Up to 500°C metal-to-metal seals

#### **Next Steps:**

- Prototypes and materials
- Prototype built and ready for in situ testing









## Thank you for your attention!

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