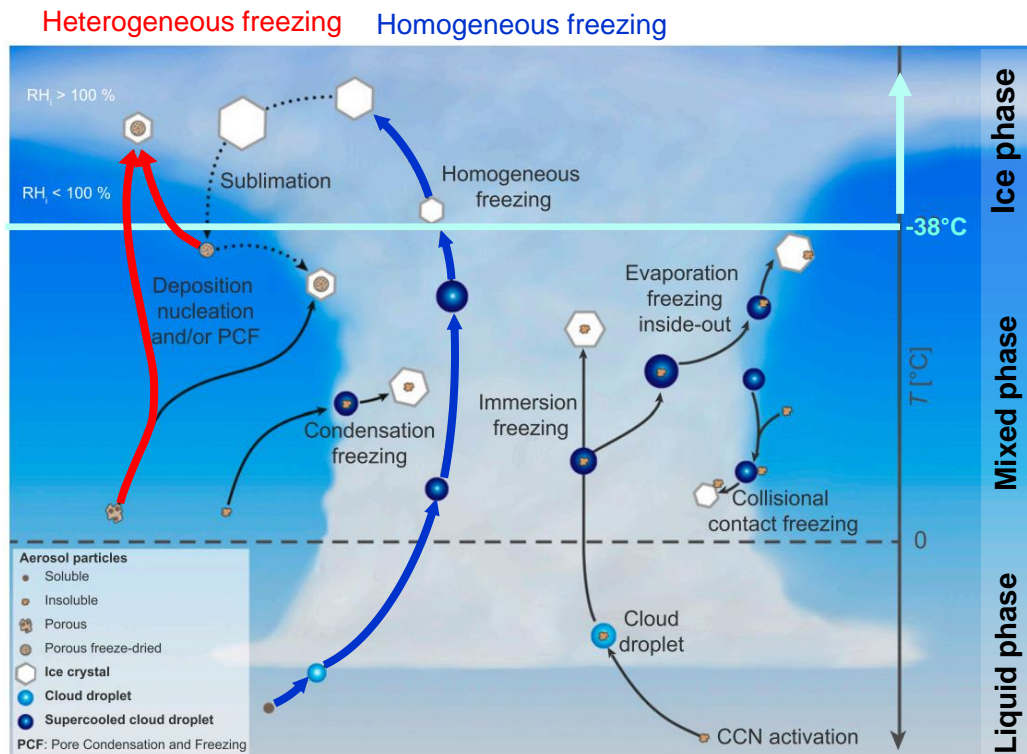


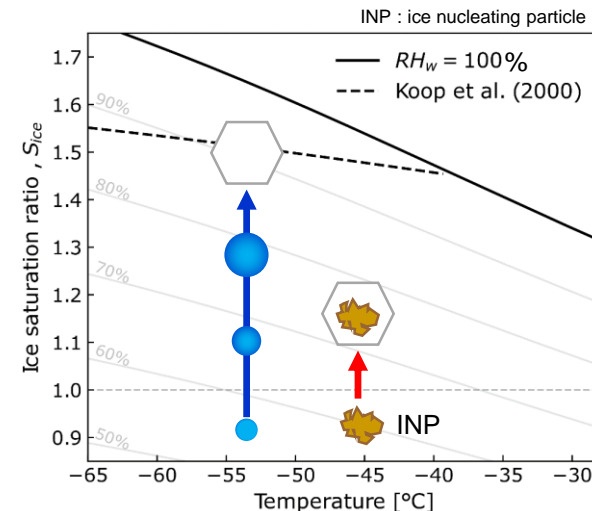
Crystallization, deliquescence and ice nucleation ability of ammoniated sulphate particles in the cirrus cloud temperature range

B. Bertozzi, R. Wagner, K. Höhler, H. Saathoff, T. Leisner, and O. Möhler
Institute of Meteorology and Climate Research (IMK)

Ice nucleation in the atmosphere



Kanji et al., 2017



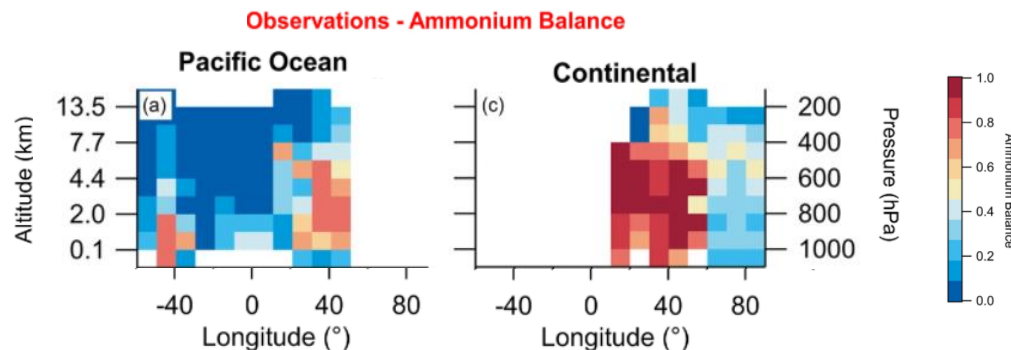
Competition homogeneous vs heterogeneous

Microphysical and optical properties of the cloud

Different radiative effect

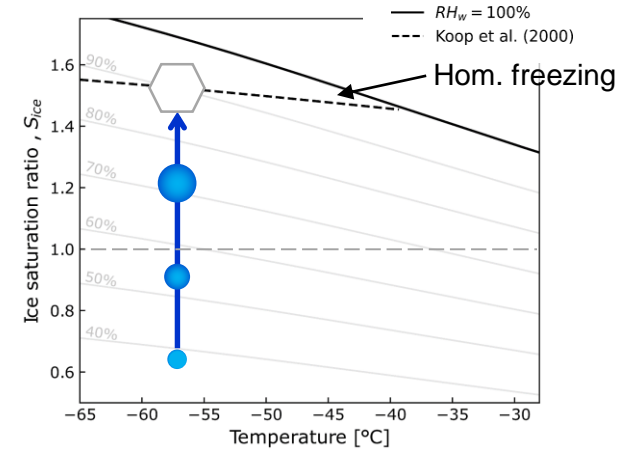
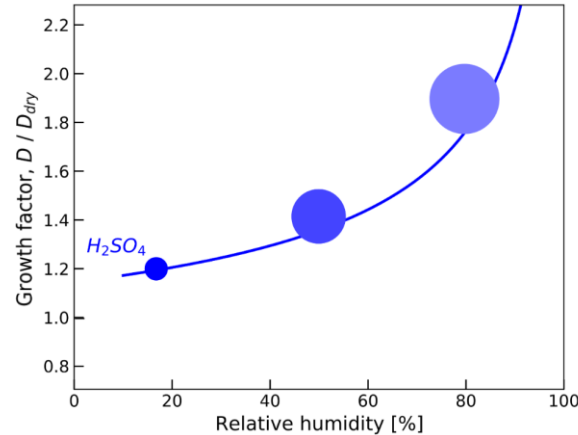
Aerosol particles in the upper troposphere (UT)

- Sulphate particles
 - Geographical and vertical variability in neutralization state (acidity)
- Organics
 - Often internally mixed with sulphate
- Crustal elements (e.g. mineral dust)
 - Coated by a S-rich layers
- Marine origin (e.g. sea salt)
- Soot
- Ammonium nitrate (Höpfner et al., 2019)
- ...

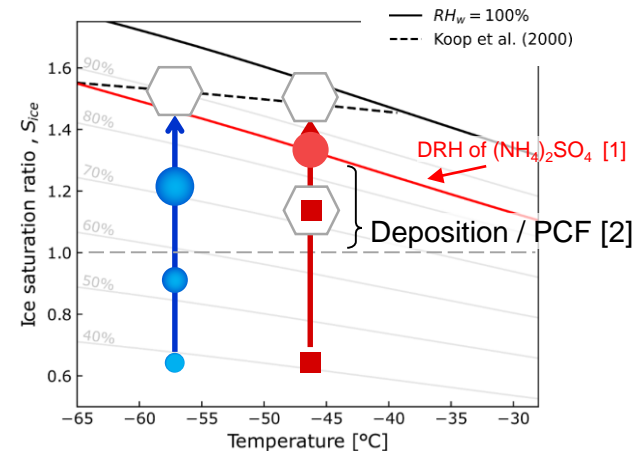
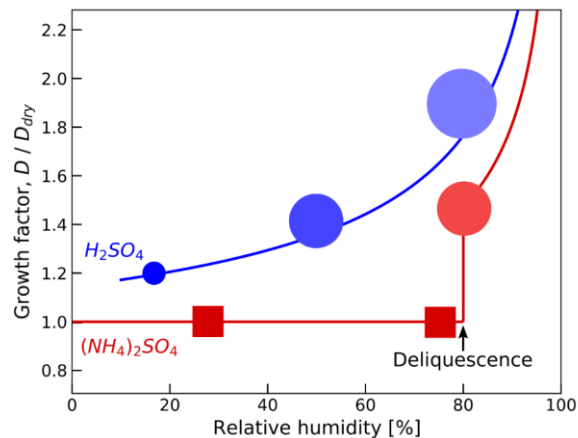
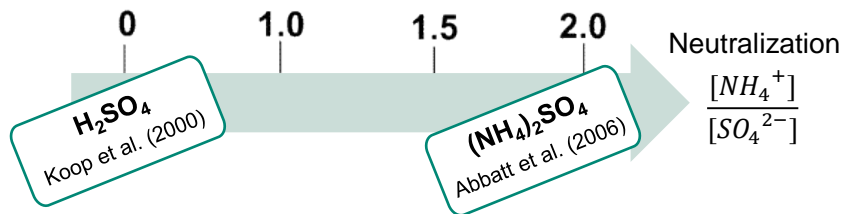


Nault et al., 2021

Ice nucleation ability of sulphate particles



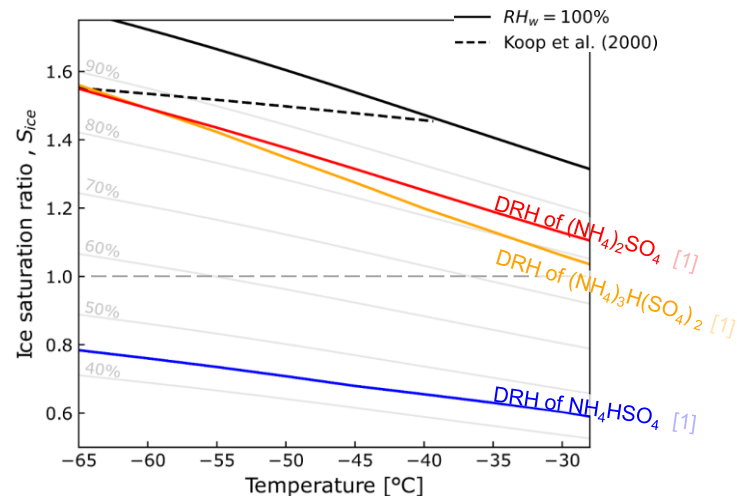
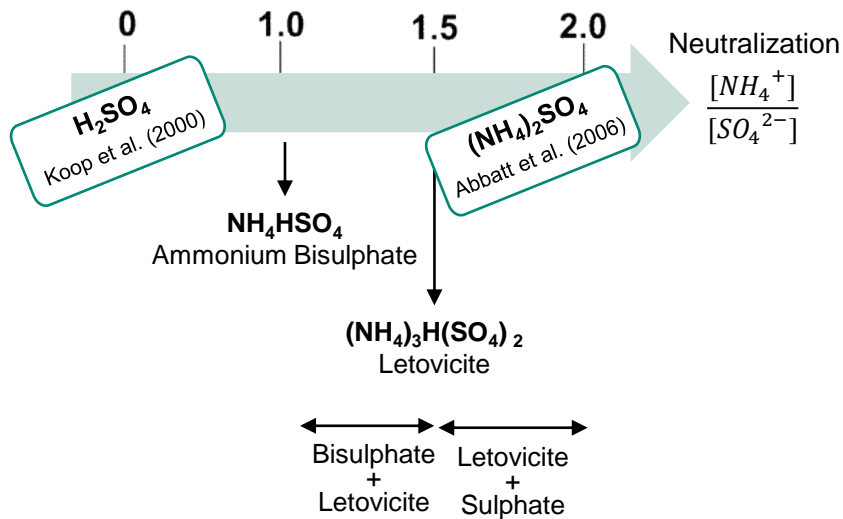
Ice nucleation ability of sulphate particles



[1] Clegg et al., 1998

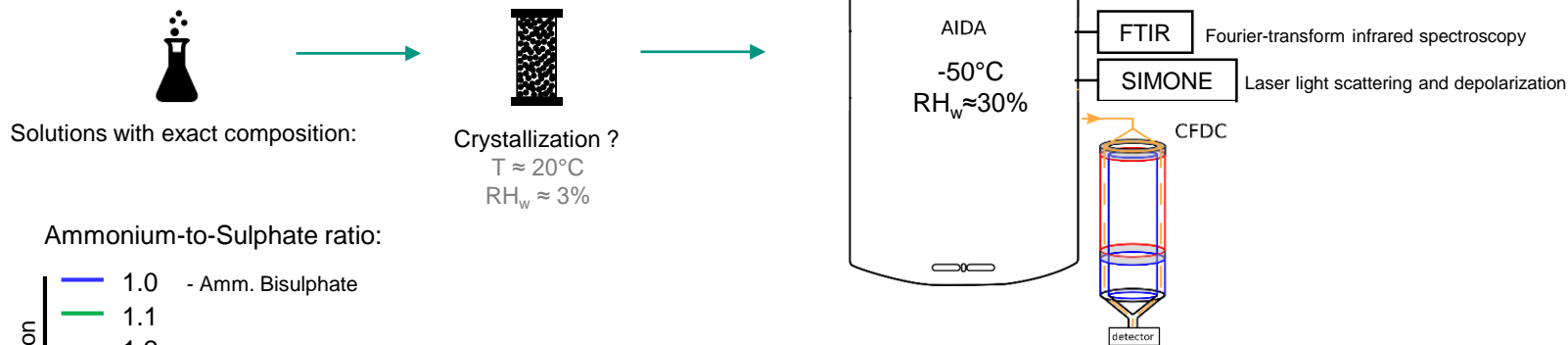
[2] Abbatt et al., 2006

Ice nucleation ability of sulphate particles

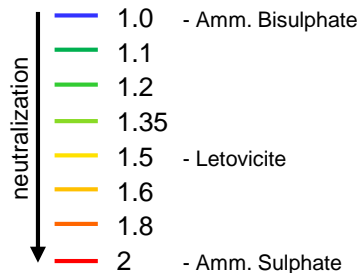


[1] Clegg et al., 1998

Experiments with bulk solutions



Ammonium-to-Sulphate ratio:

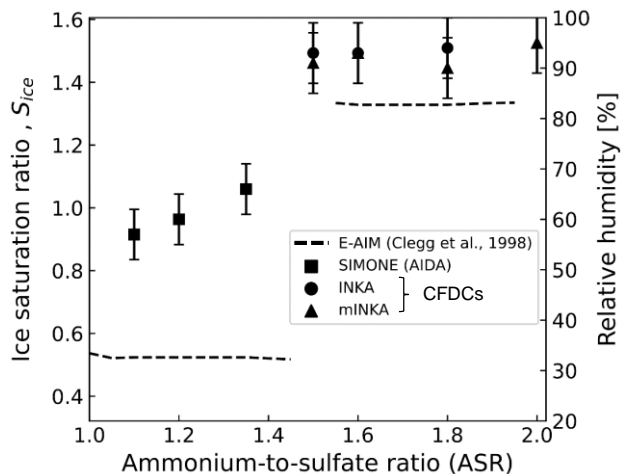


Deliquescence and IN ability:

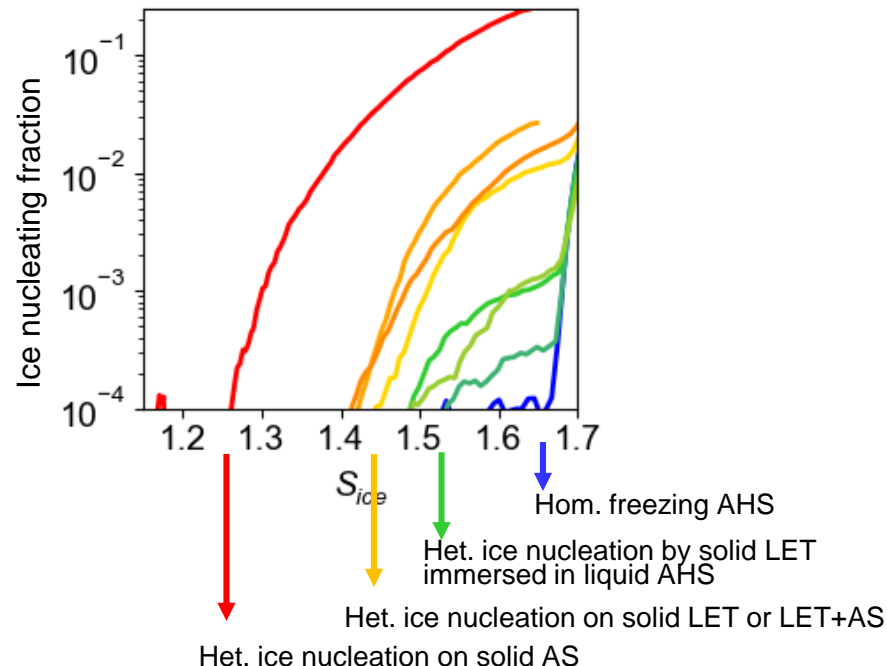
- Expansion cooling experiments
 $-50^\circ\text{C} < T < -60^\circ\text{C}$
 $0.7 < S_{ice} < \text{hom. freezing}$
- Continuous flow diffusion chambers
 $-40^\circ\text{C} < T < -60^\circ\text{C}$
 $1.0 < S_{ice} < \text{water saturation}$

Experiments with bulk solutions - Results

Partial deliquescence
(from -56 to -52°C)

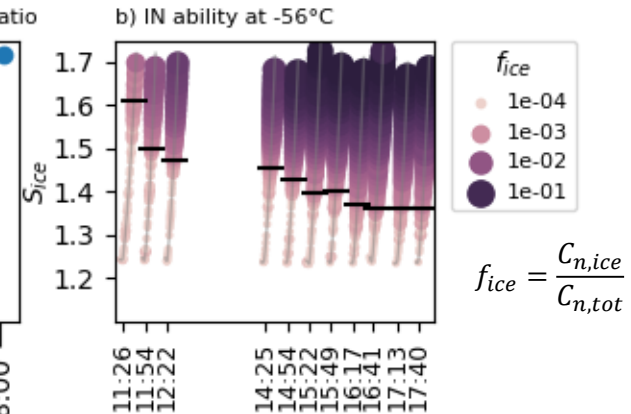
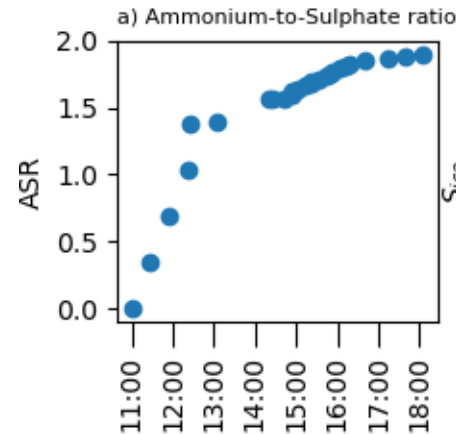
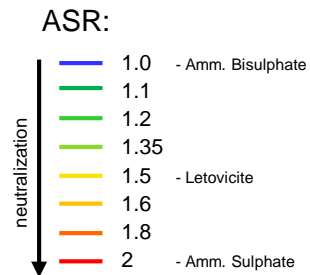
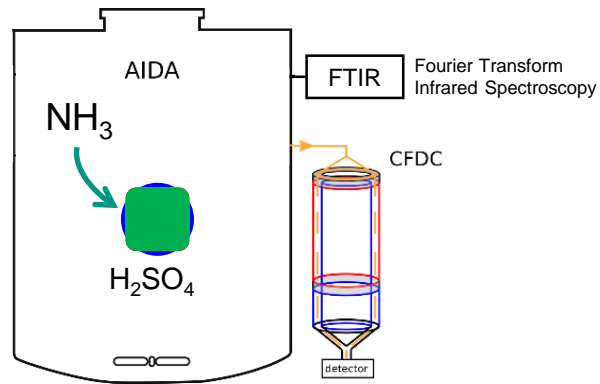


Ice nucleation (-56°C)



AHS : Ammonium bisulphate
LET : Letovicite
AS : Ammonium sulphate

In situ neutralization experiments



In situ neutralization experiments

