

Factors controlling the dissolved ^{137}Cs seasonal fluctuations in the Abukuma River under the influence of the Fukushima Nuclear Power Plant accident

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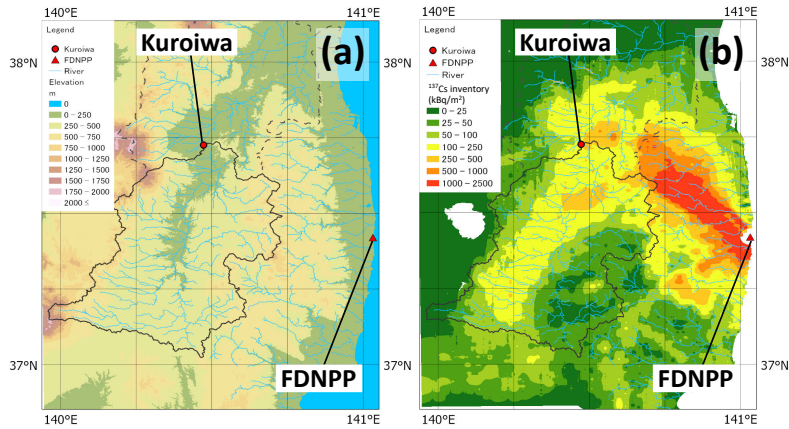


Figure 1. (a) Main and sub streams in the Abukuma River Basin, (b) the ^{137}Cs inventory from MEXT (2011).

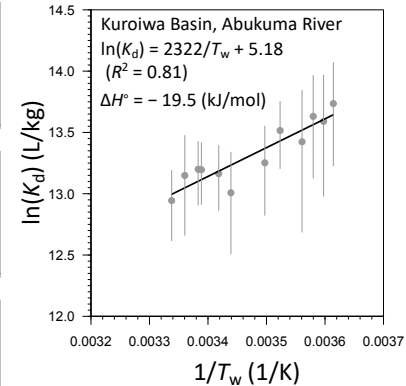
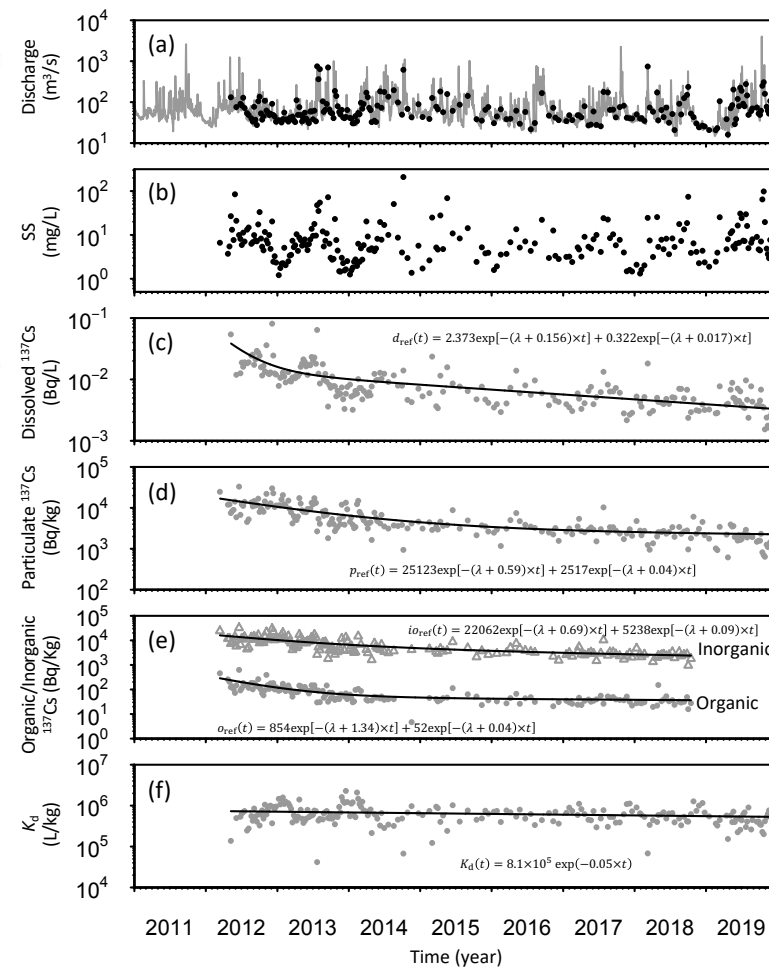


Figure 3. Relationship of the log-scale solid-liquid distribution coefficient (K_d) to inverse water temperature ($1/T_w$). Gray dots indicate the observed values. Black dots and bars represent the averages and the standard deviation for monthly K_d . The solid line ($\ln(K_d) = 2,322/T_w + 5.18$; $R^2 = 0.81$) expresses the linear regression line of best fit.

Figure 2. Long-term trends of (a) river discharge*, (b) dissolved ^{137}Cs activity concentration, (c) particulate ^{137}Cs activity concentration, (d) inorganic ^{137}Cs (triangles) and organic ^{137}Cs activity concentration (solid circles), (e) solid-liquid distribution coefficient (K_d). The gray line and solid circles indicate the observed river discharge rate at the Kuroiwa Point and the river discharge rate on the sampling day, respectively.

Details are available.
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- Dissolved ^{137}Cs in the Abukuma River displays significant seasonal fluctuations.
- The main factors determining the seasonality of dissolved ^{137}Cs activity concentrations are water temperature and K^+ .
- The standard enthalpy of ^{137}Cs desorption from sediments was determined to be approximately -19.3 kJ/mol .

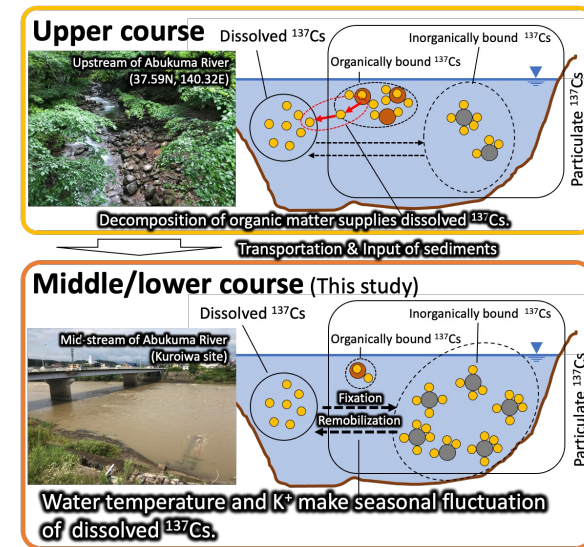


Figure 4. Schematic diagram of ^{137}Cs dynamics and factors controlling seasonal fluctuations in the middle course of the Abukuma River.