



# Mechanisms of in-situ remediation of soil with lead and sulfate contaminants using multiple binder strategies: experimental and numerical studies

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**1** Motivation   **2** Methodologies   **3** Preliminary Results   **4** Main Conclusions

SCAN ME

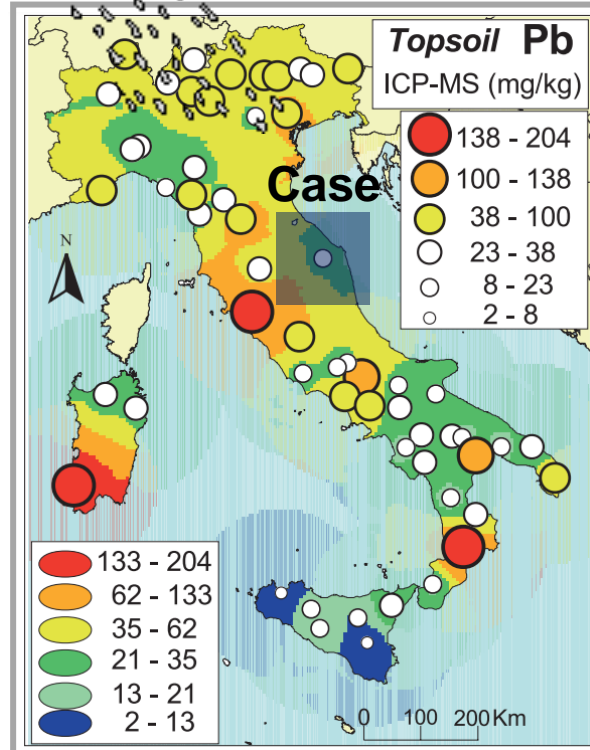


Abstract Information



Lab Webpage

## Contaminated soil : From business to solid waste



### Objective 1 Fate of lead in contaminated soils:

☞ Effects of weathering process on lead distribution;

### Objective 2 Remediation strategies determination:

☞ In-situ, High performance, Low-carbon footprint, Profitable;

### Objective 3 Monitoring and predicting the leaching behavior:

☞ Using the geochemical modeling to prove the binder-soil system is stable and long-lasting in a less cost and time-consuming way.

Figure. Interpolated concentration maps of Pb in Italian topsoil. (Limit in soil: **85 mg/kg**; Limit in water: **0.01mg/L**) (Ref. De Vivo et al. (2008)).

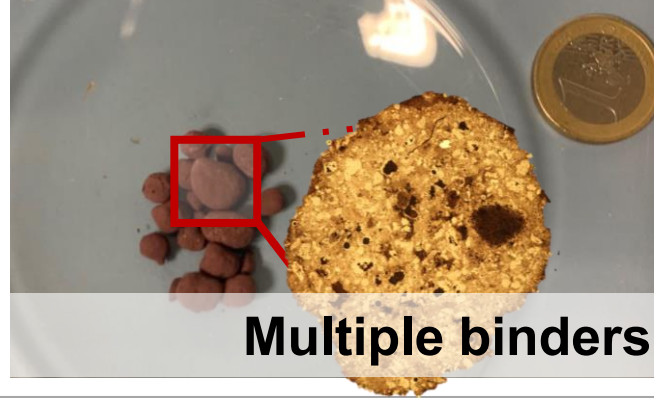
## In-situ remediation: from lab to land

## Laboratory measurements

### 1 SAMPLING PROCESS



### 2 IN-SITU REMEDIATION



### 3 LEACHING BEHAVIOUR



## Geochemical modeling approach

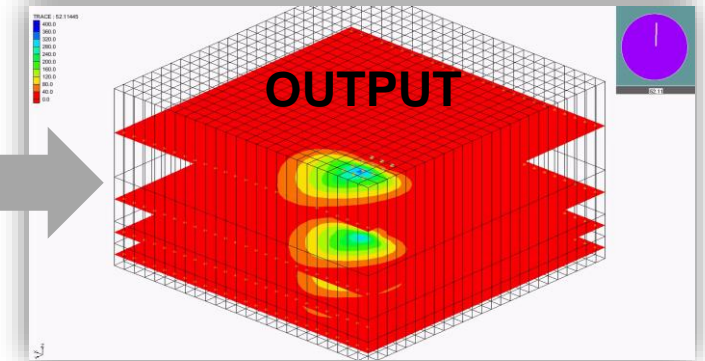
### INPUT

*Characterization results  
Thermodynamic database*



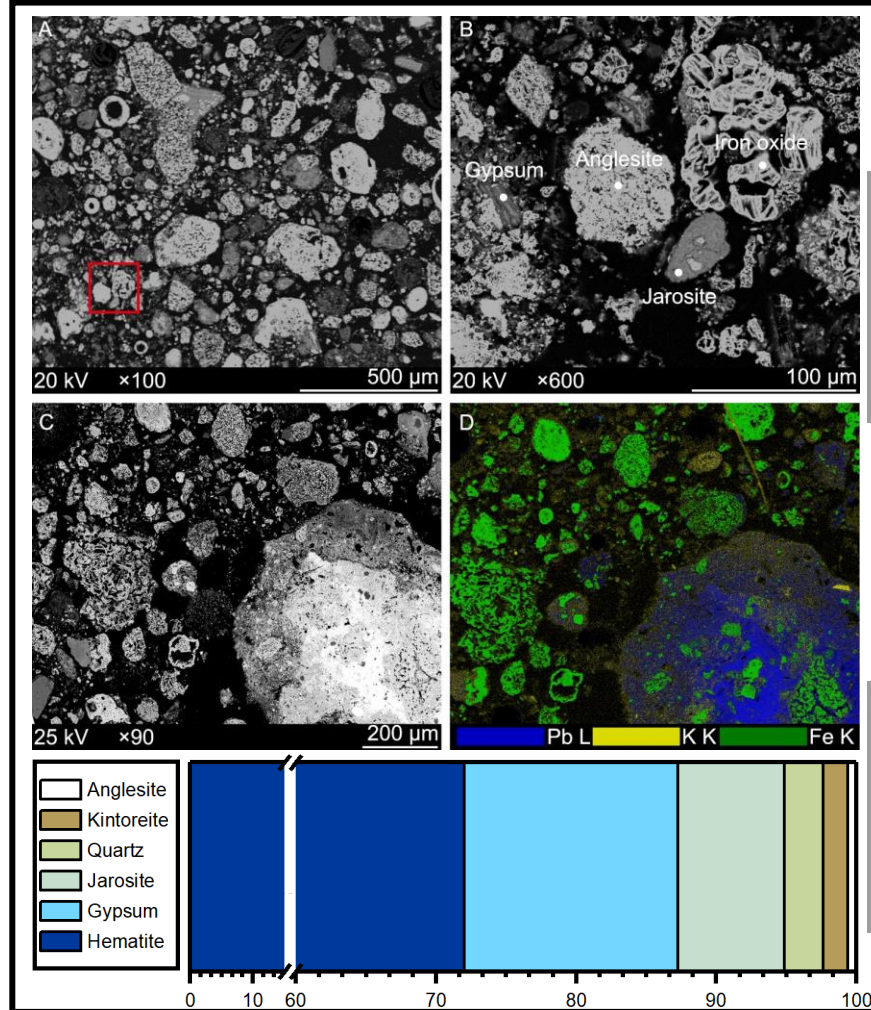
**MODELING APPROACH**  
SURFACE ADSORPTION  
TRANSPORT  
EQUILIBRIUM  
REACTION  
ION EXCHANGE  
SOLUTION SPECIES

### OUTPUT



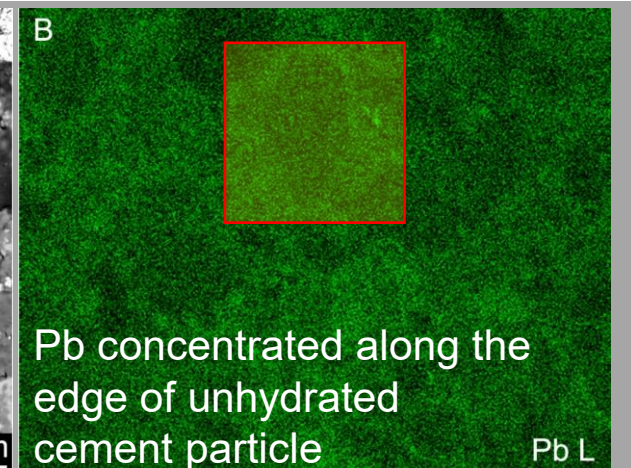
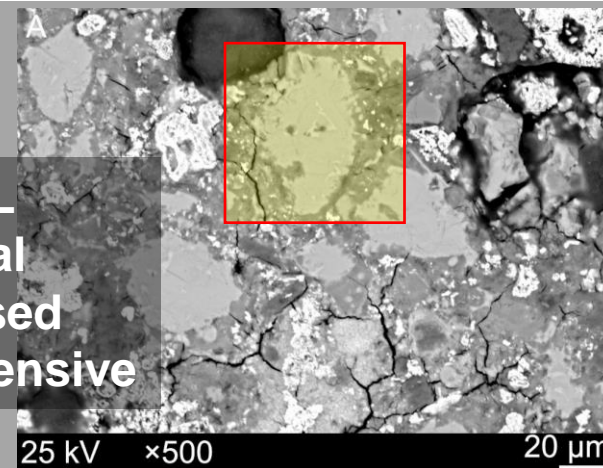
# 3 Preliminary Results

## Lead in the contaminated soil

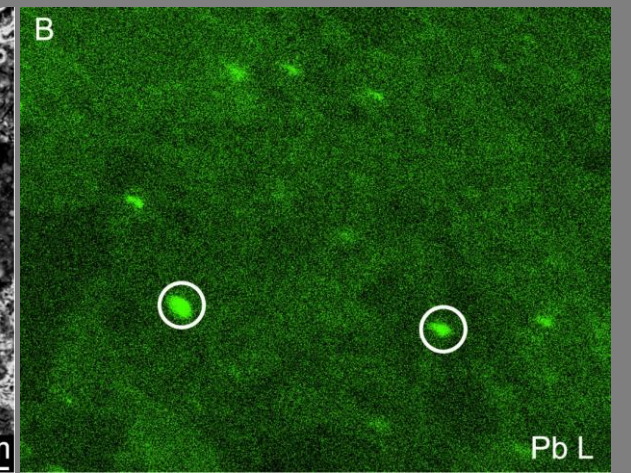
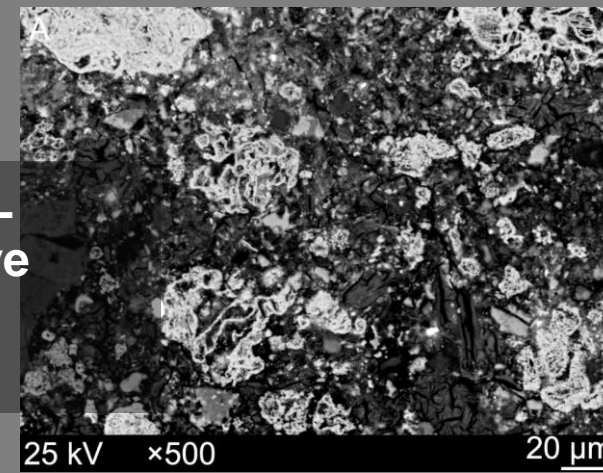


## Lead in the pellets (Traditional and alternative binder)

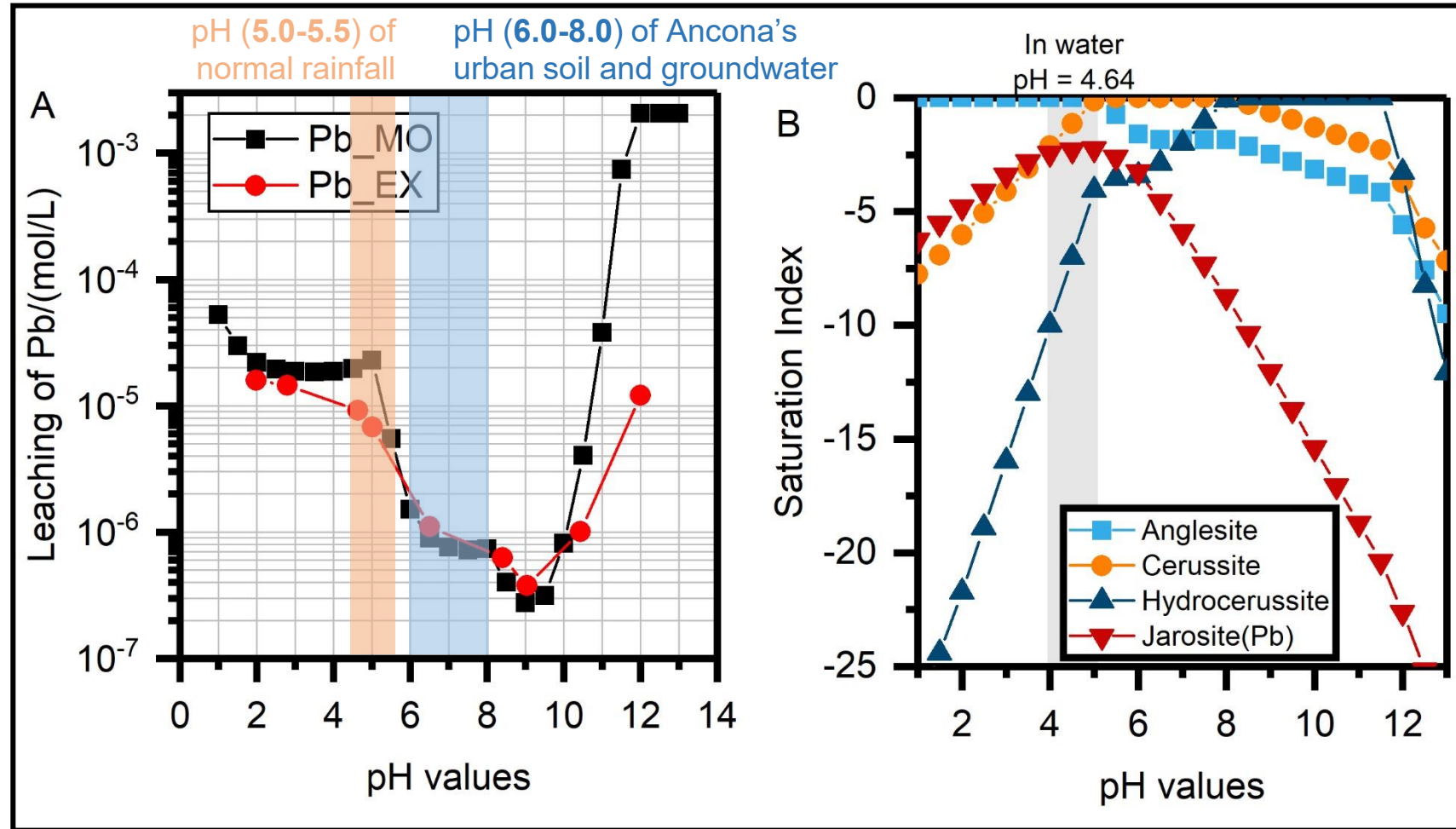
OPC-SOIL  
Traditional  
Widely used  
Less expensive



CAC-SOIL  
Alternative  
OPC free  
Costly



## Lead in the solution via leaching experiments (EX) and geochemical modeling (MO)



**Highly amphoteric  
Pb leaching behavior**



**Sensitive to aquatic  
ecosystems**

**Precipitation/dissolution  
of carbonates**



**CO<sub>2</sub> from the  
atmosphere**

# 4 Conclusions

## The fate of Pb in contaminated soil:

- Anglesite and kintoreite are the main Pb-bearing crystalline phases, which may be generated from the oxidation of pyrite particles.
- Sulfates (jarosite) were confirmed as the Pb-bearing phases through the coexistence of jarosite-type compound and adsorption.

## The fate of Pb in stabilized pellets:

- Pb was well dispersed within the cementitious matrix.
- Different immobilization mechanisms and performances were shown with the applied binders.



SCAN ME

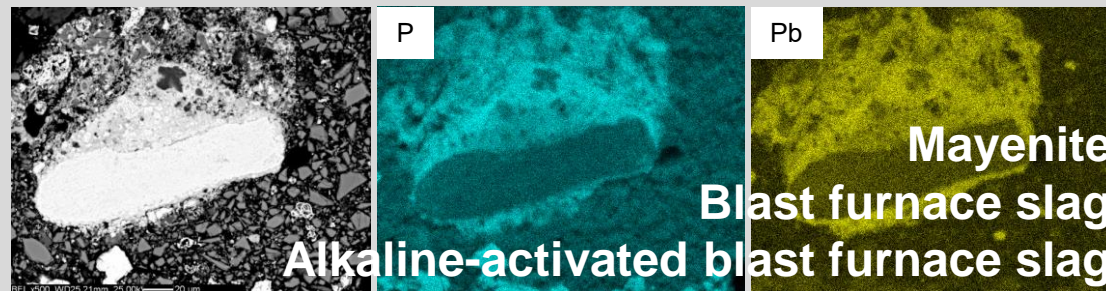


Abstract Information

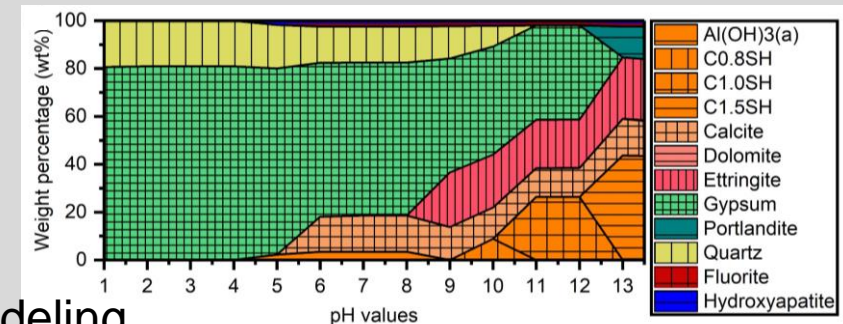


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More results. If you have any questions or curiosity, please just drop me an Email ([yikai.liu@phd.unipd.it](mailto:yikai.liu@phd.unipd.it)).



Or Modeling



**Thanks for your attention!**