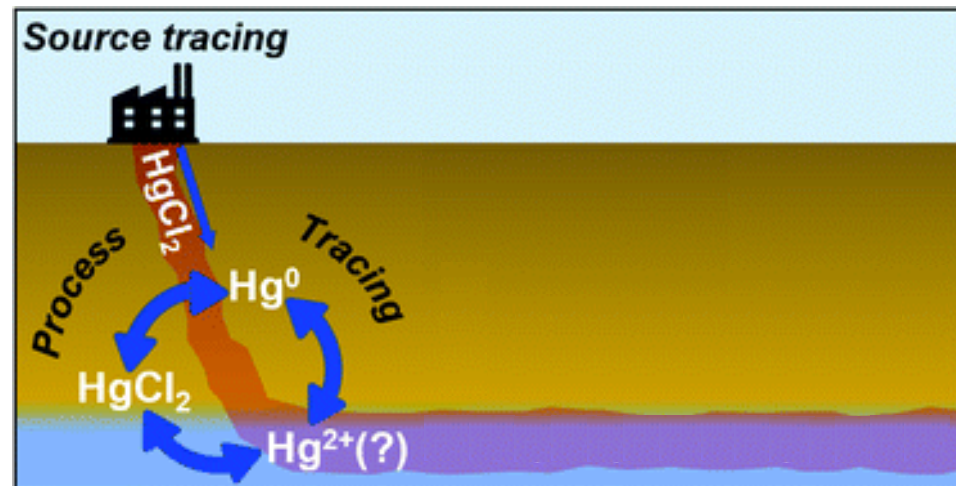




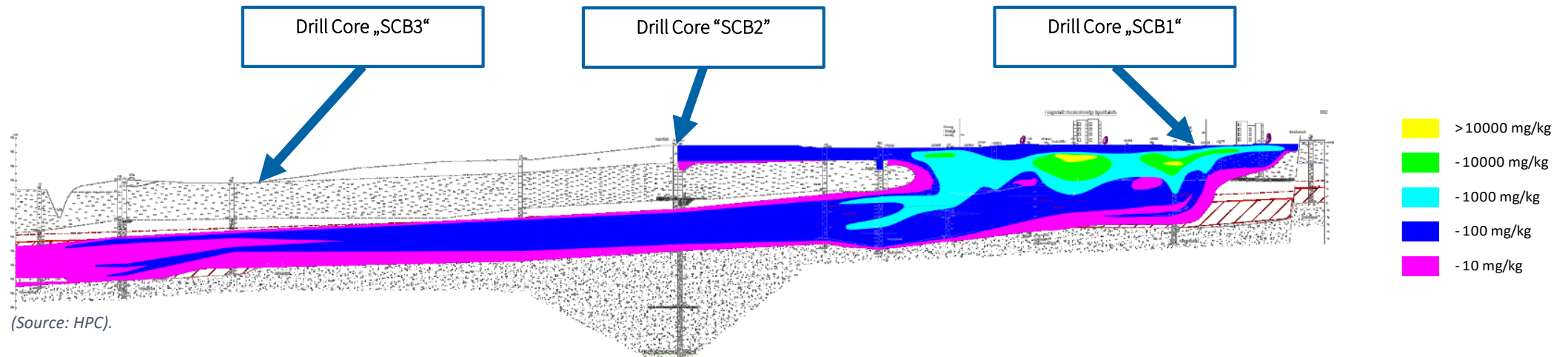
How Hg isotope source signatures can be overprinted by biogeochemical processes in the subsurface of contaminated legacy sites

Lorenz Schwab, David S. McLagan, Lu Chen, Jan Pietrucha, Stephan M. Kraemer, Harald Biester, and Jan G. Wiederhold



Introduction

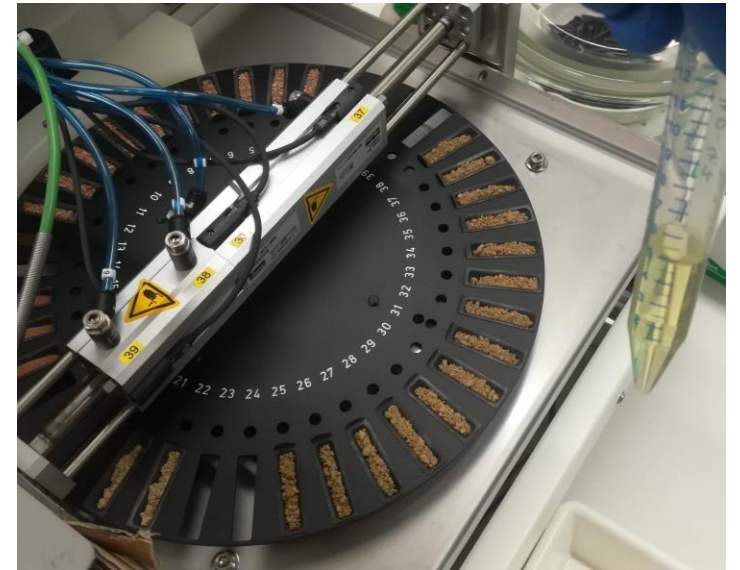
- Contaminated legacy sites can act as long-term sources from which mercury (Hg) is released to waterbodies, soils and air
- Example from a former „kyanization“ facility – treatment of timber with HgCl_2



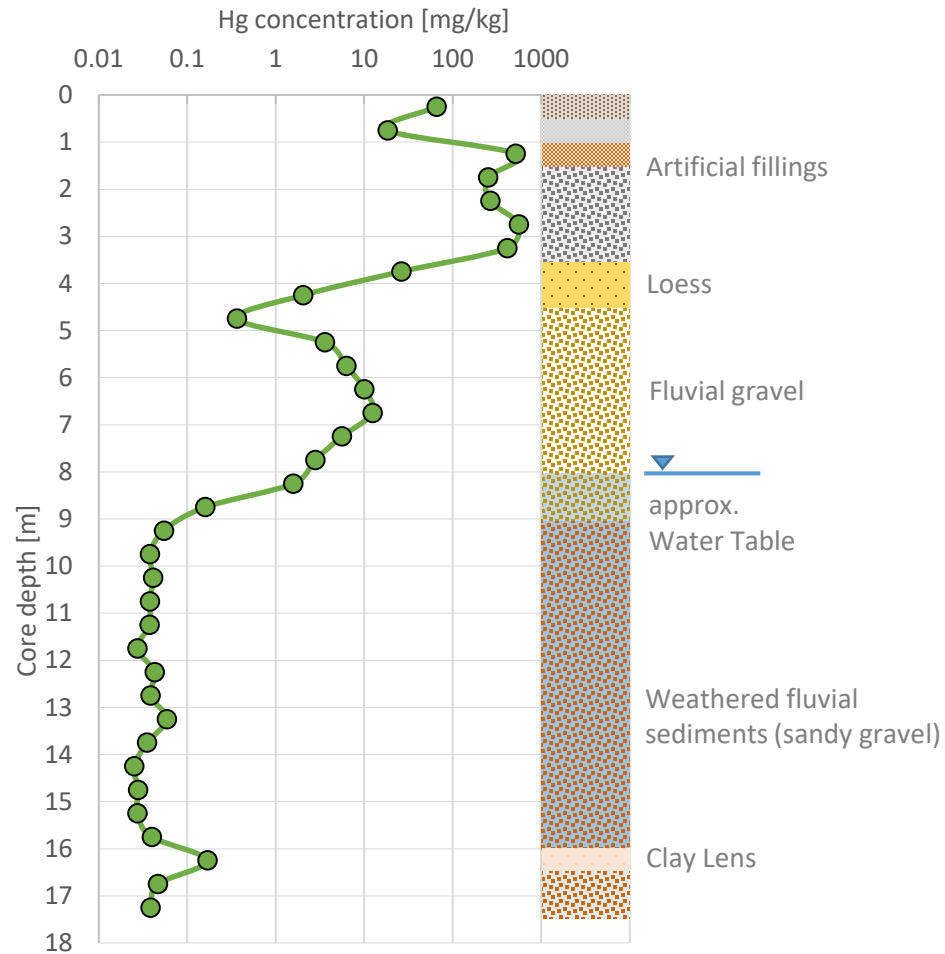
- Goal: Use Hg stable isotopes to better understand in situ transformation processes

Sampling / Methods

- General characterization and Hg concentrations (THg)
 - Pyrolytic thermodesorption (PTD)
 - Sequential extractions (SEP)
- Hg binding forms
 - Bulk samples
 - Extracts from SEP

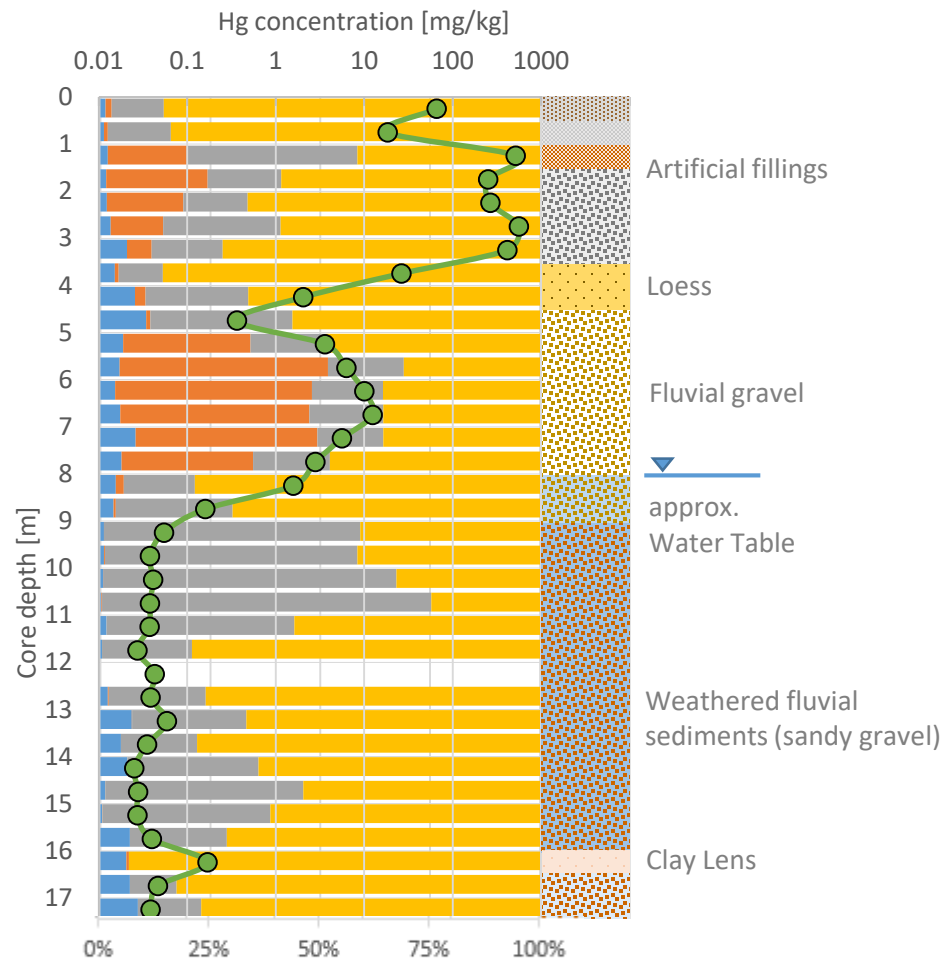


Results demonstrated on one drill core - THg



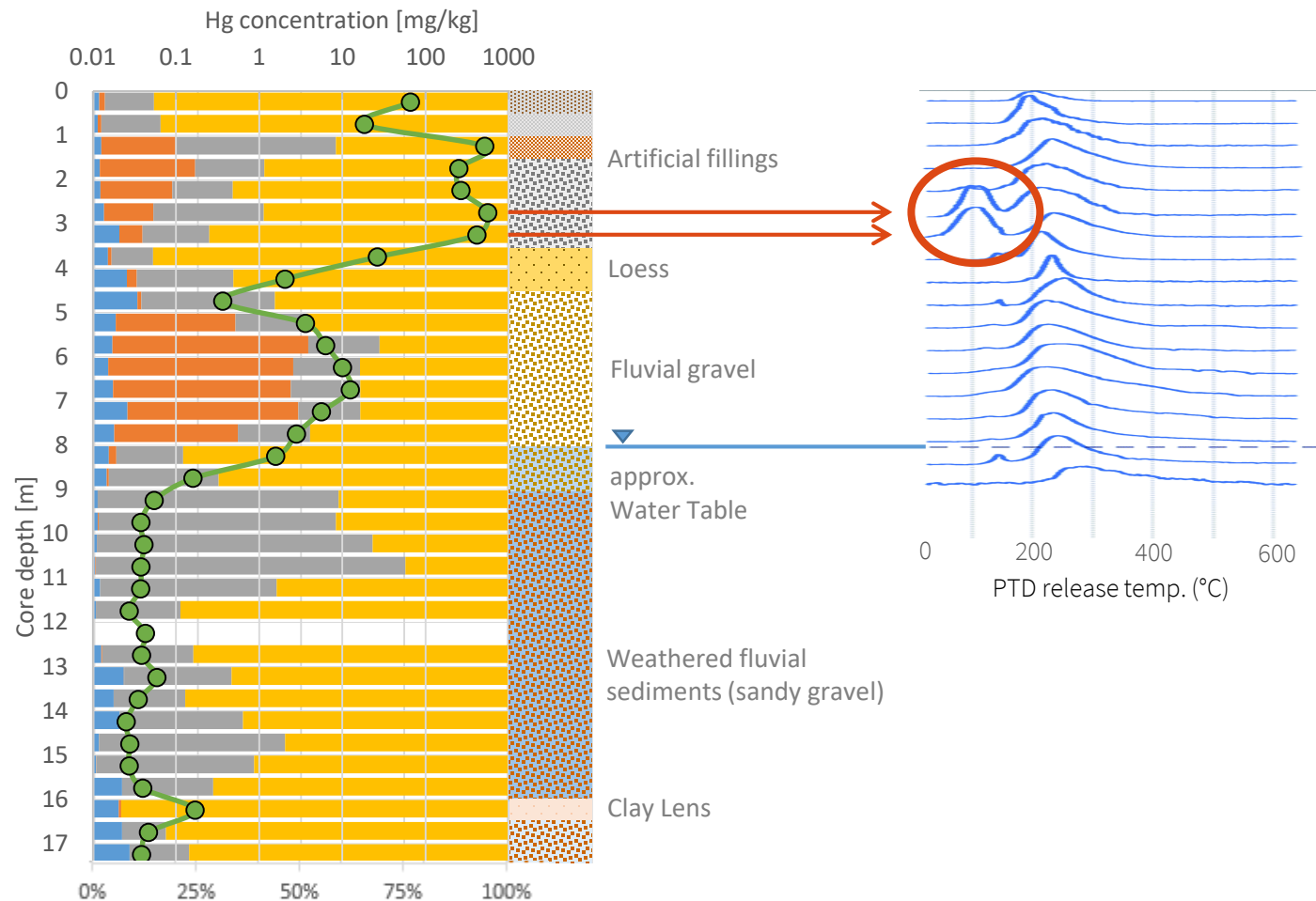
- Highly contaminated upper most meters (artificial fillings)
- Below the loess layer there is much less Hg

Results demonstrated on one drill core - SEP



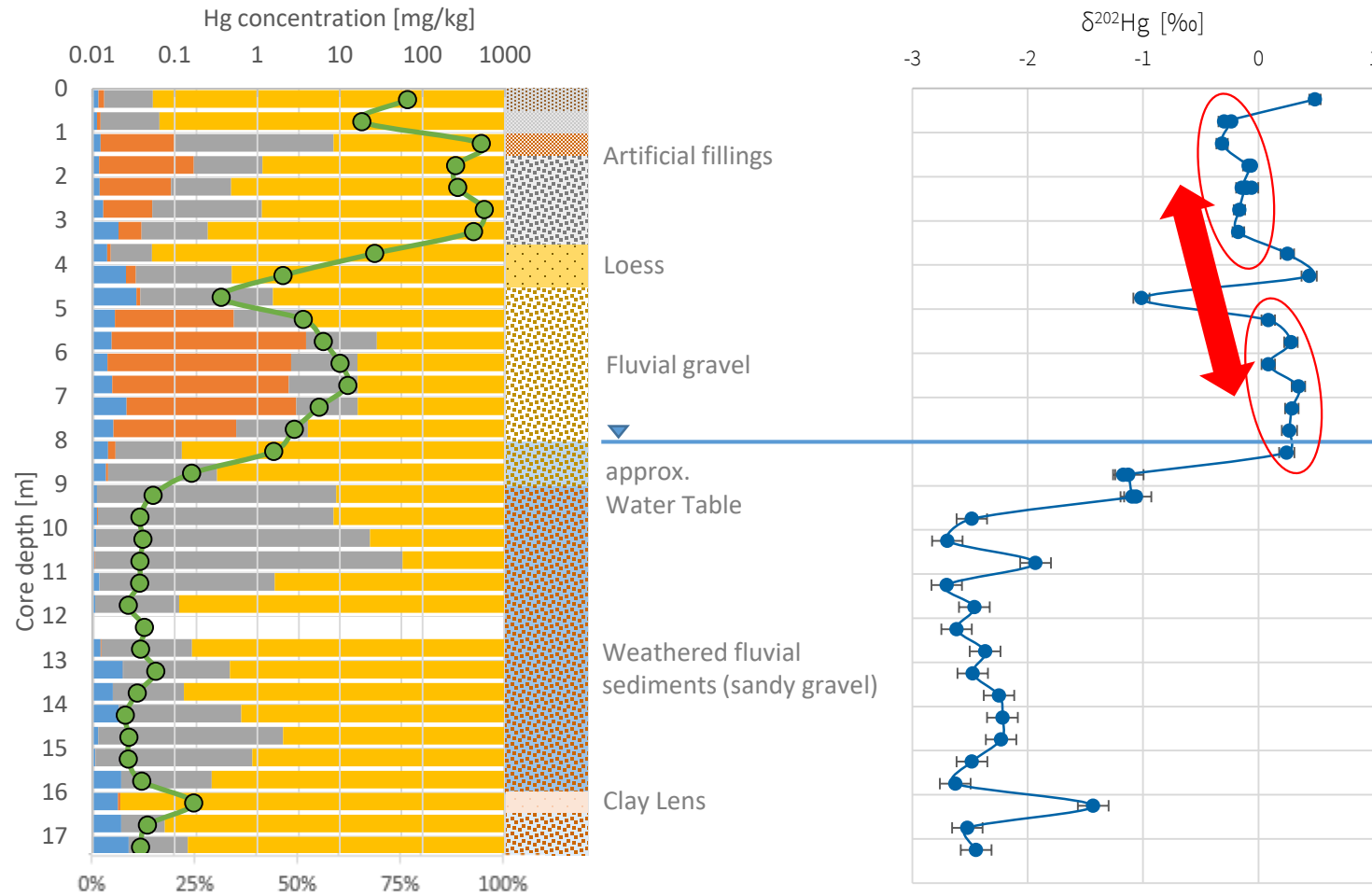
- In SEP HgCl_2 is usually attributed to F1
 - There is Hg in F1, but it is a relatively small fraction
- Clearly more Hg in F2 extract in the highly contaminated layers
 - Not all HgCl_2 extracted in F1?

Results demonstrated on one drill core - PTD



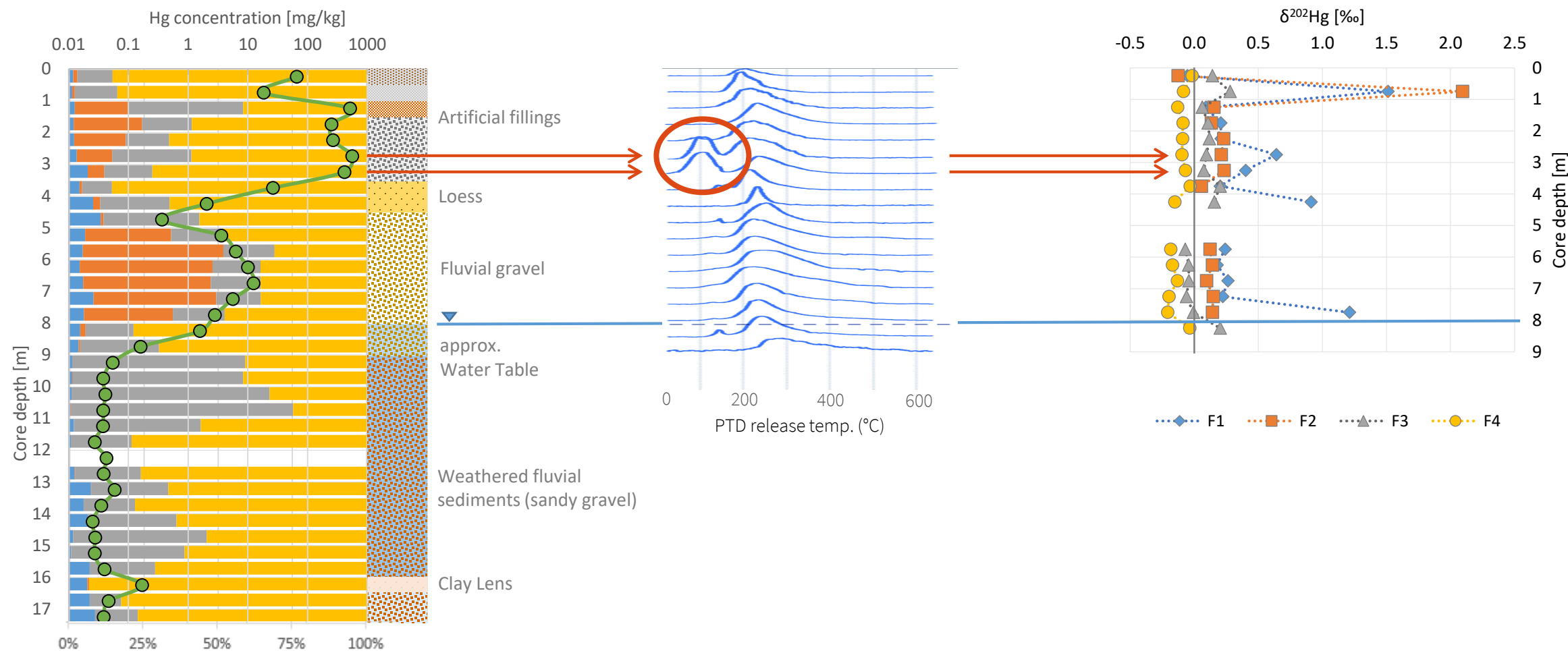
- Hg(0) above loess layer
- Reflected by Isotopes?
 - Would cause an isotopically light extract

Results demonstrated on one drill core – Bulk isotopes



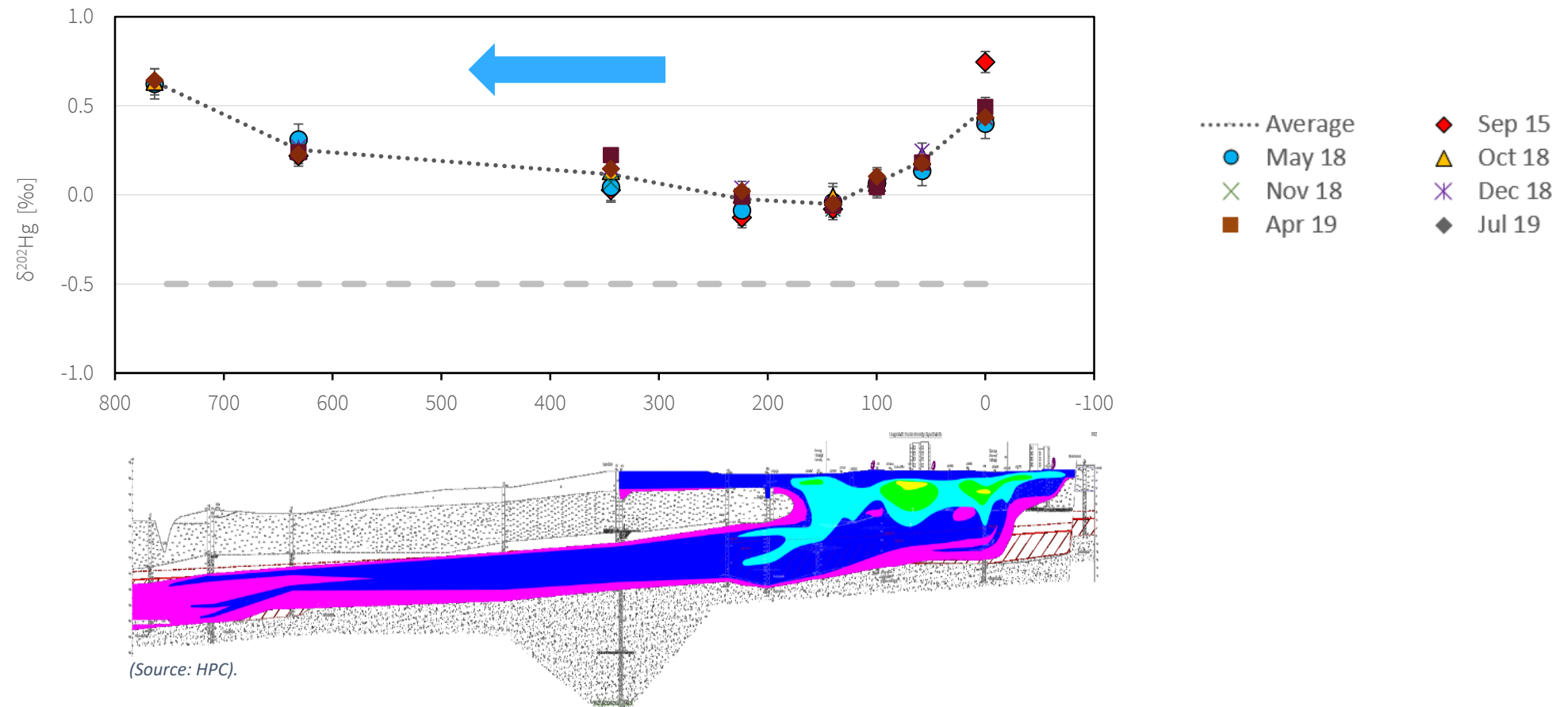
- Major shifts in the isotopic values along the depth profile of drill cores
 - Sorption processes
→ lighter value in remaining pool
 - Downward transport
→ continuously increasing $\delta^{202}\text{Hg}$ value
- Bulk isotope values would not reflect Hg(0) formation

Results demonstrated on one drill core – SEP isotopes



Hg isotopes in groundwater

- Progressively more positive liquid phase $\delta^{202}\text{Hg}$ values compared to the solid phase





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Demystifying mercury geochemistry in contaminated soil–groundwater systems with complementary mercury stable isotope, concentration, and speciation analyses†

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