

# Fate and transport of titanium dioxide nanoparticles in porous media

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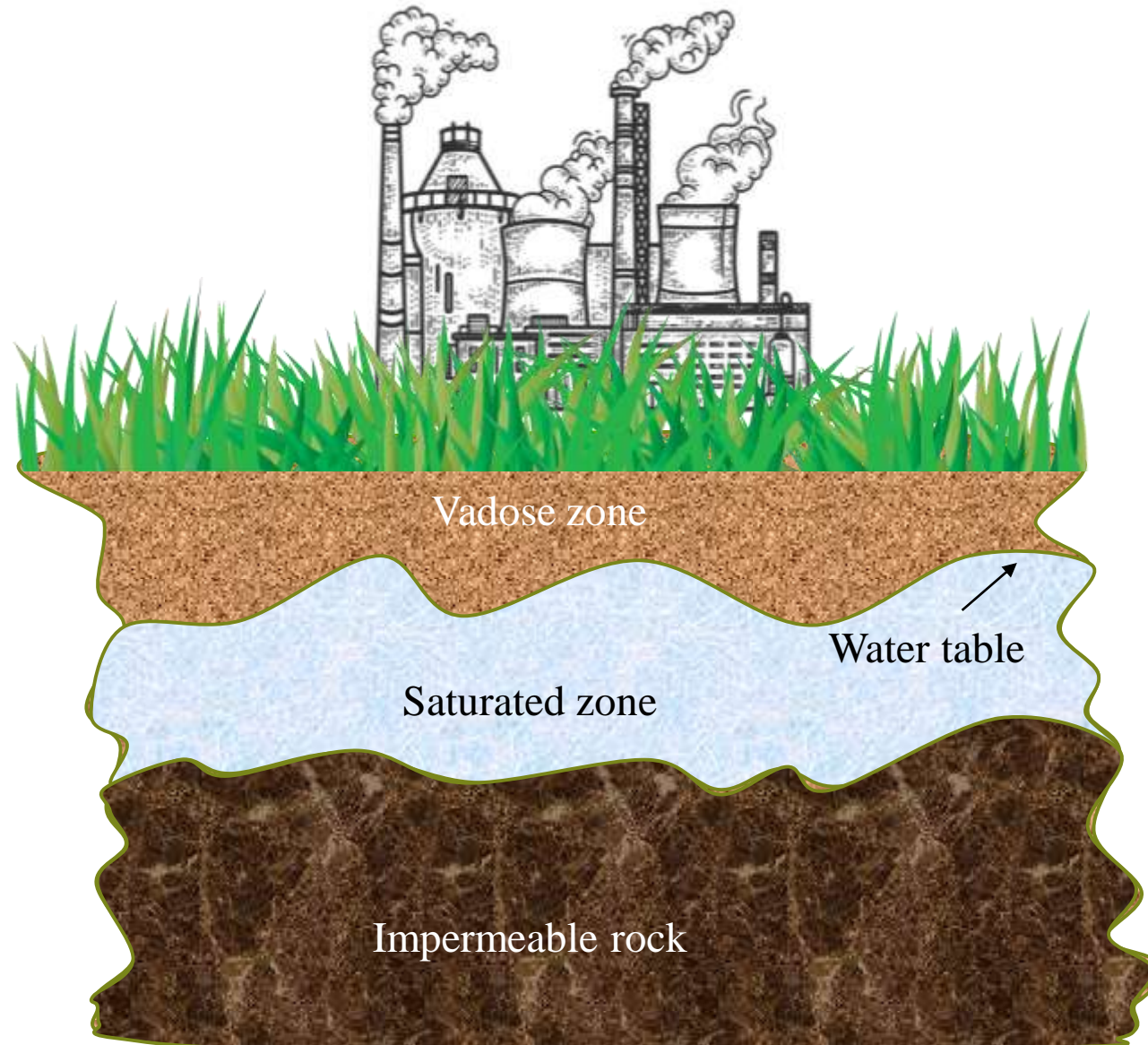
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# Introduction:

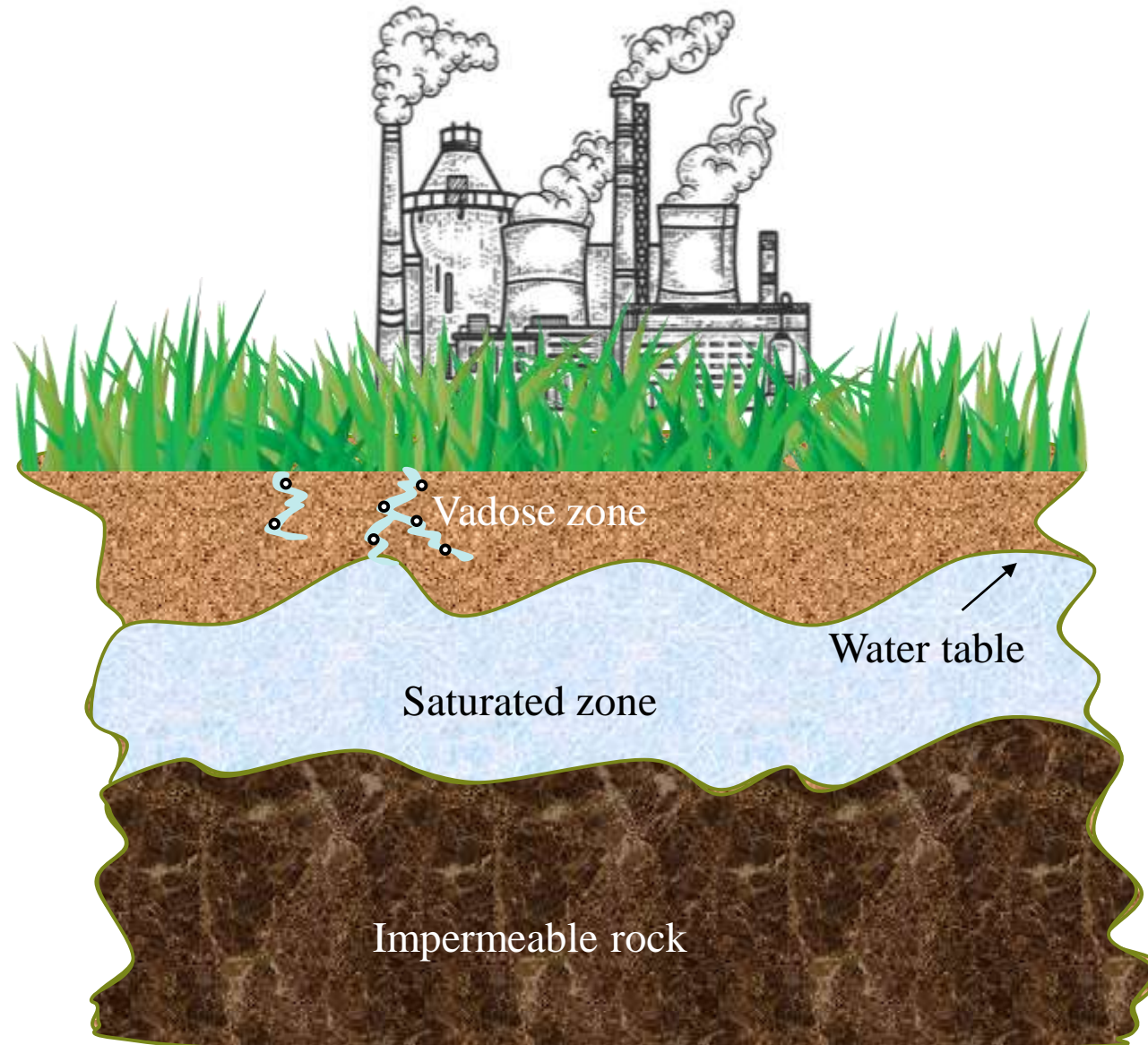


- **What are ENPs?** size 1–100nm.
- $\text{nTiO}_2$  (metal oxide based) global production is the highest (5000 tons/year) (*Park et al., 2017*)
- Highest estimated soil concentration (0.4 ug/kg) (*Park et al., 2017*)
- **$\text{nTiO}_2$  applications:**  
sunscreen lotions, burn treatment, dental, antibacterial, photocatalytic, paint pigment
- **Sources of  $\text{nTiO}_2$  discharge into the subsurface:**  
untreated industrial & agricultural waste,  
landfill leachate,  
WWTP discharge,  
nano-remediation

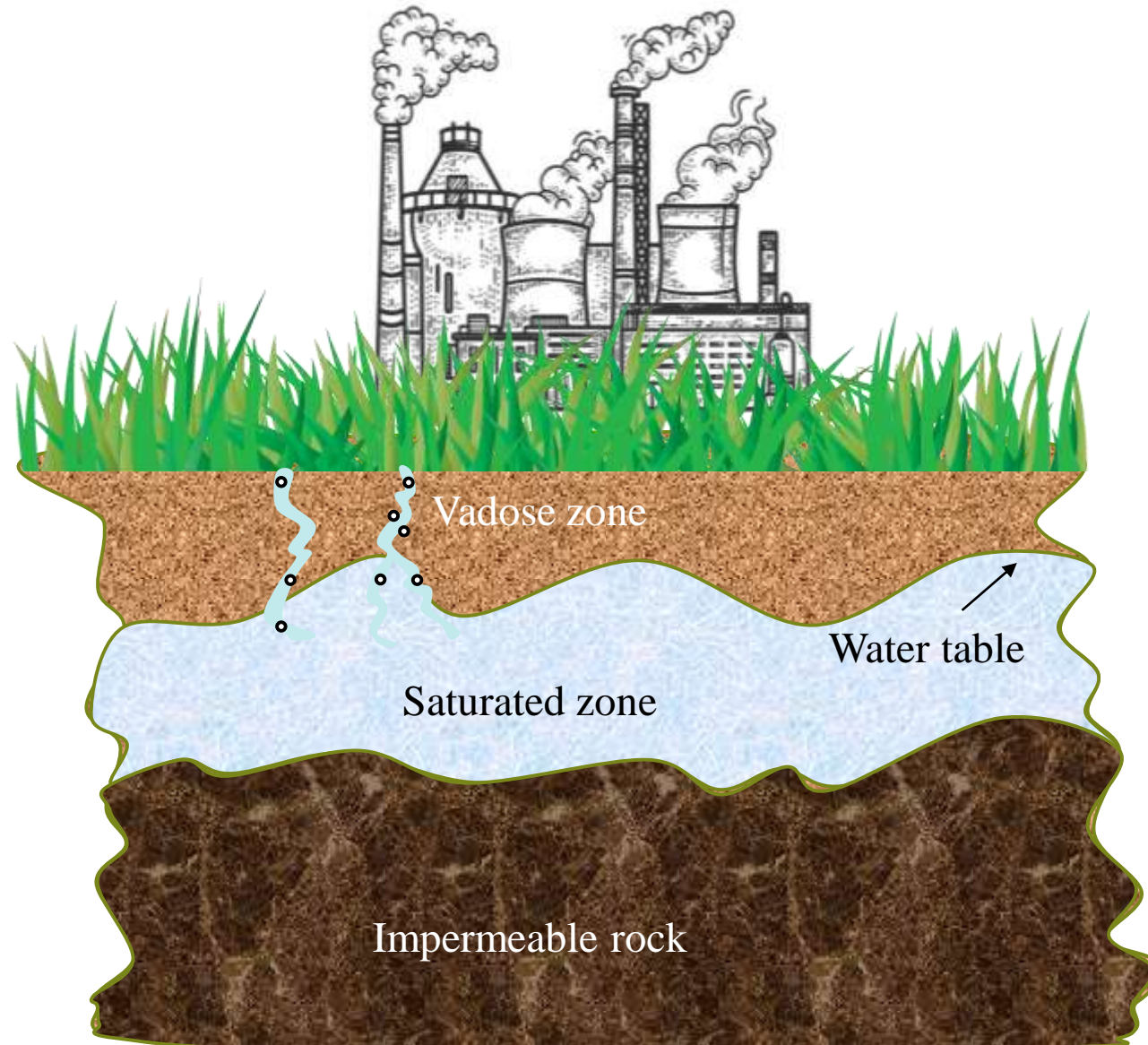
# Flow pathway of ENPs into the subsurface:



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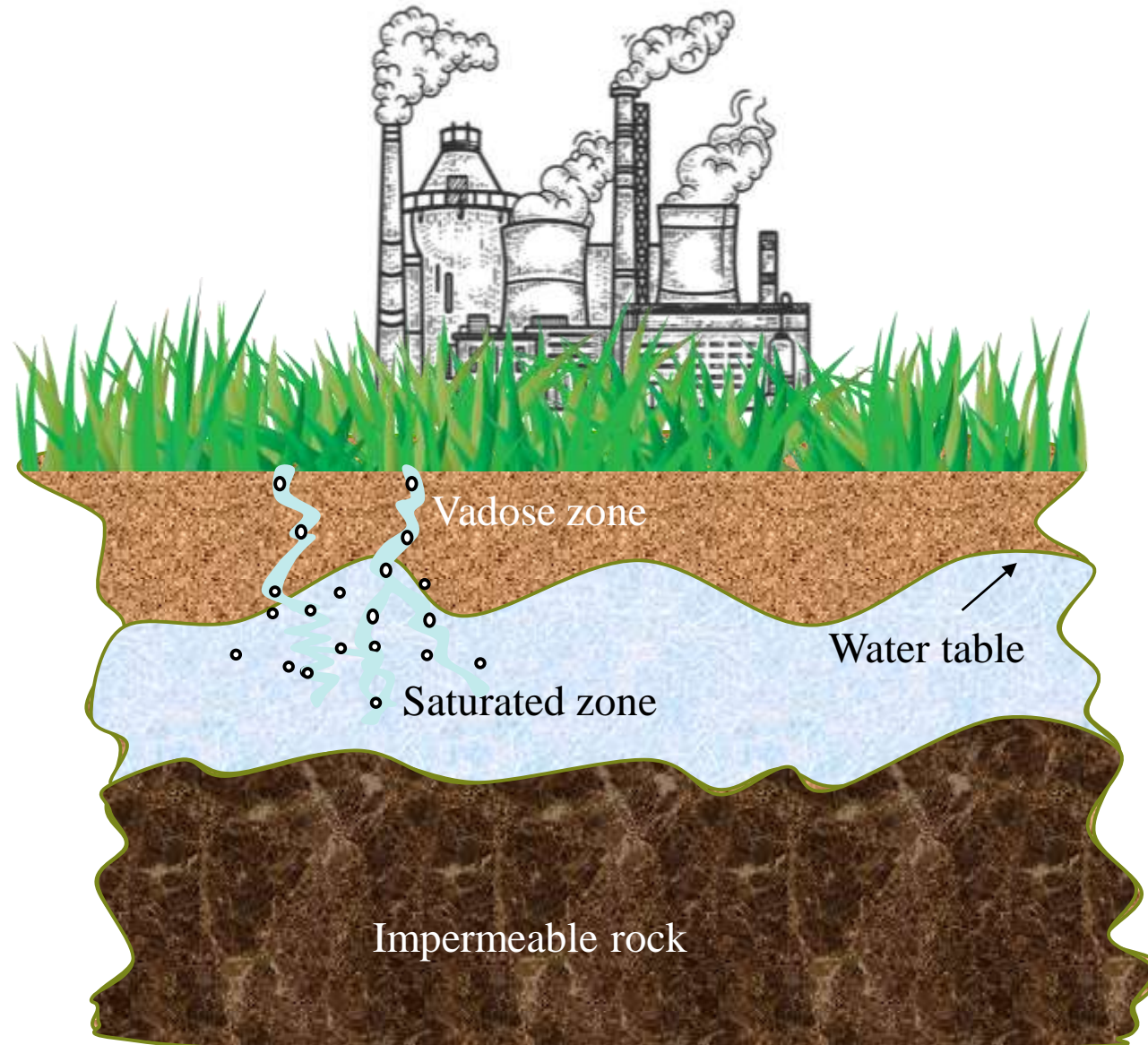


# Flow pathway of ENPs into the subsurface:

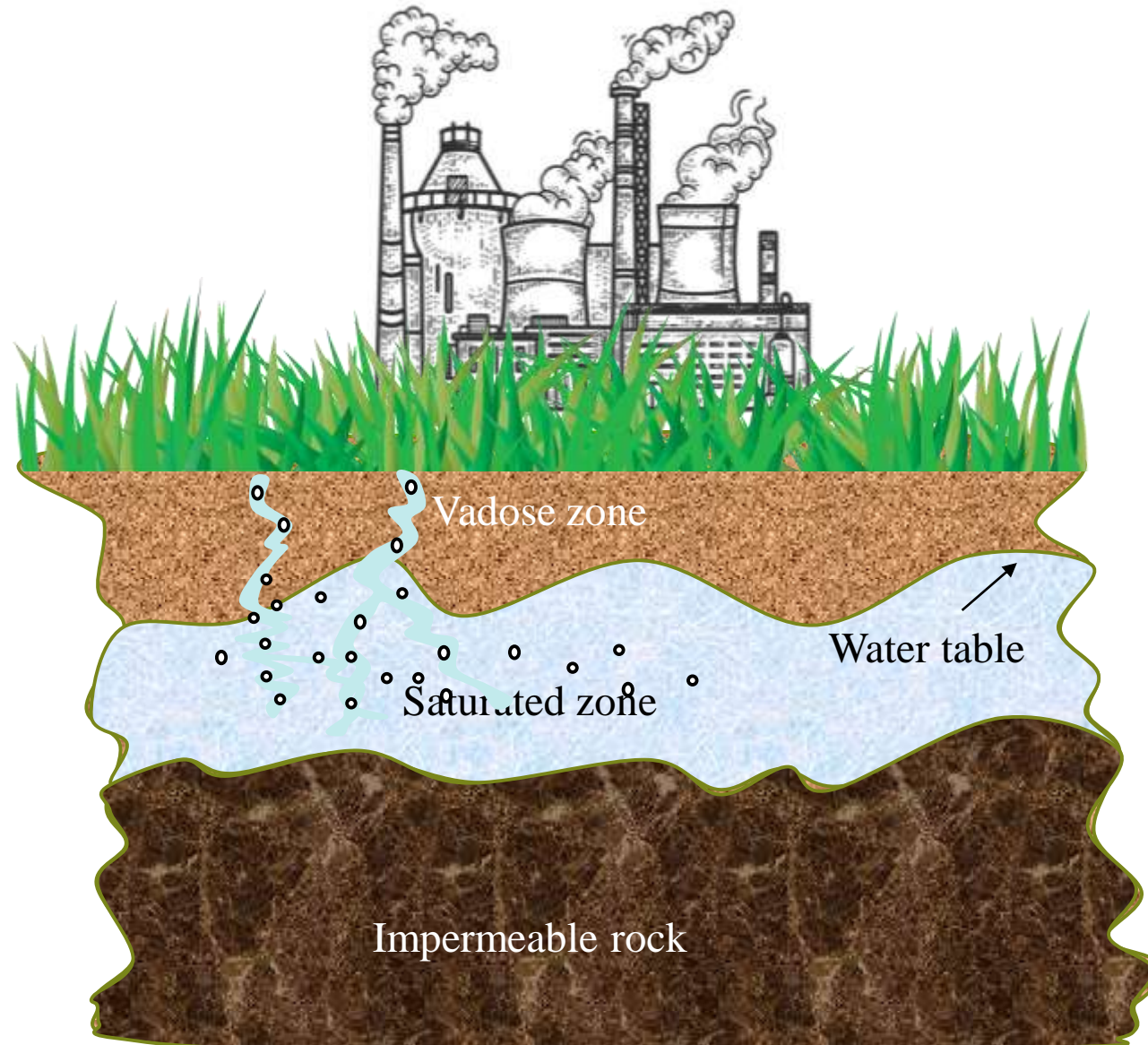




# Flow pathway of ENPs into the subsurface:



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## Objective:



- To understand nTiO<sub>2</sub> fate and transport in porous media through column experiments and modeling



# Methodology:

## *nTiO<sub>2</sub>: Column transport experiments:*

### *Experimental conditions:*

- Clean acid washed sand
- *Tracer experiment*

0.1 M NaCl solution: 3 PV, DIW: 3 PV

flow velocity: 1 ml/min

- *nTiO<sub>2</sub> experiment*

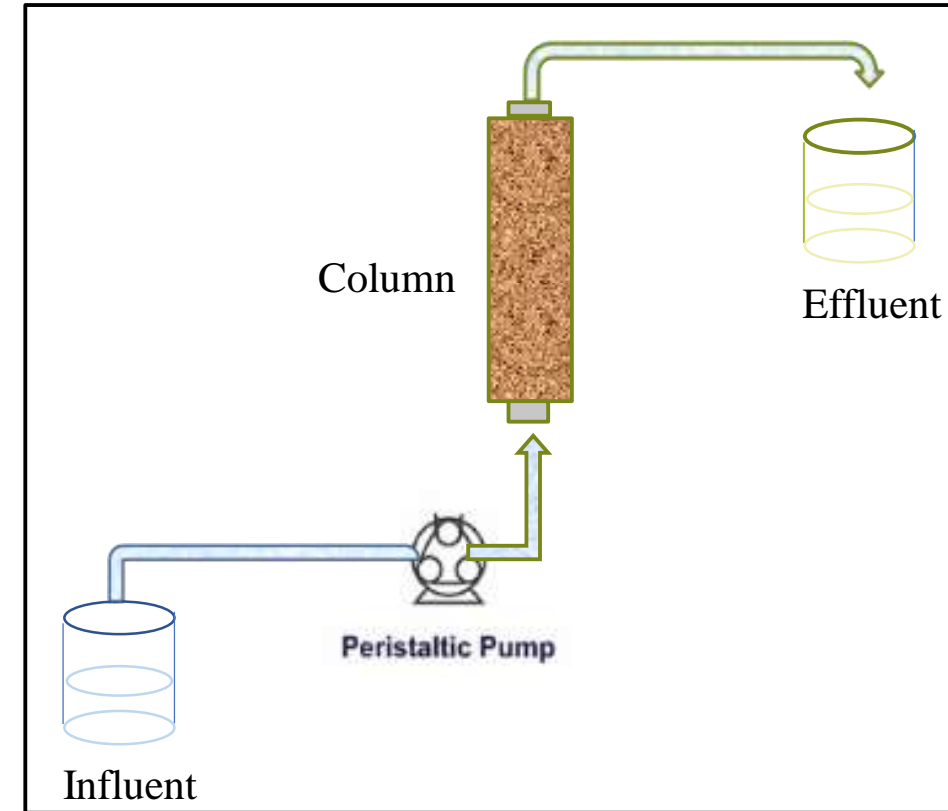
pH = 7

DIW: 10 PV, BS (0.01 M NaCl): 10 PV

nTiO<sub>2</sub> influent concentration: 15 mg/L

nTiO<sub>2</sub>: 3 PV, BS: 3 PV

- Measurements by Turbidimeter



**Column experiment setup**

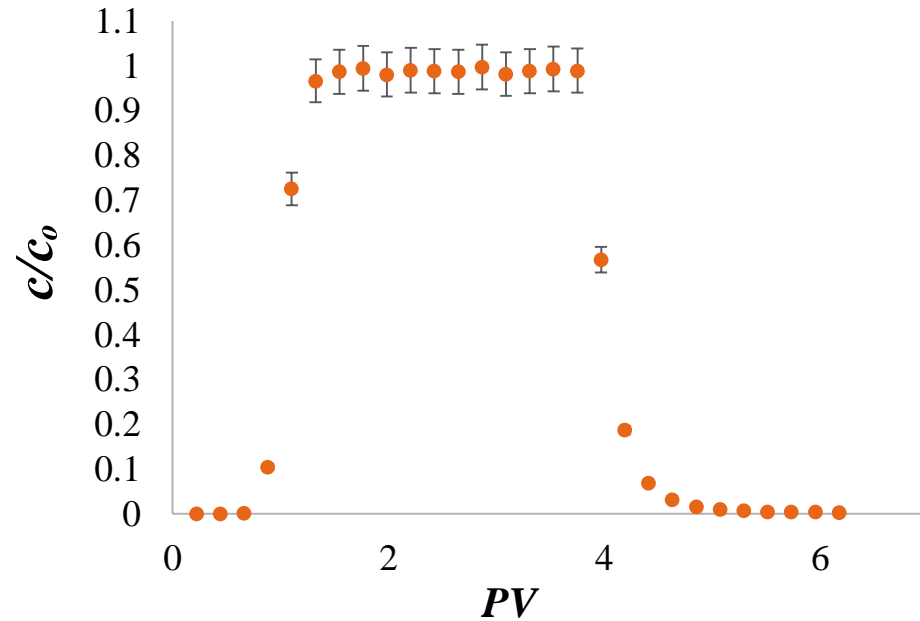


# Experimental Results:

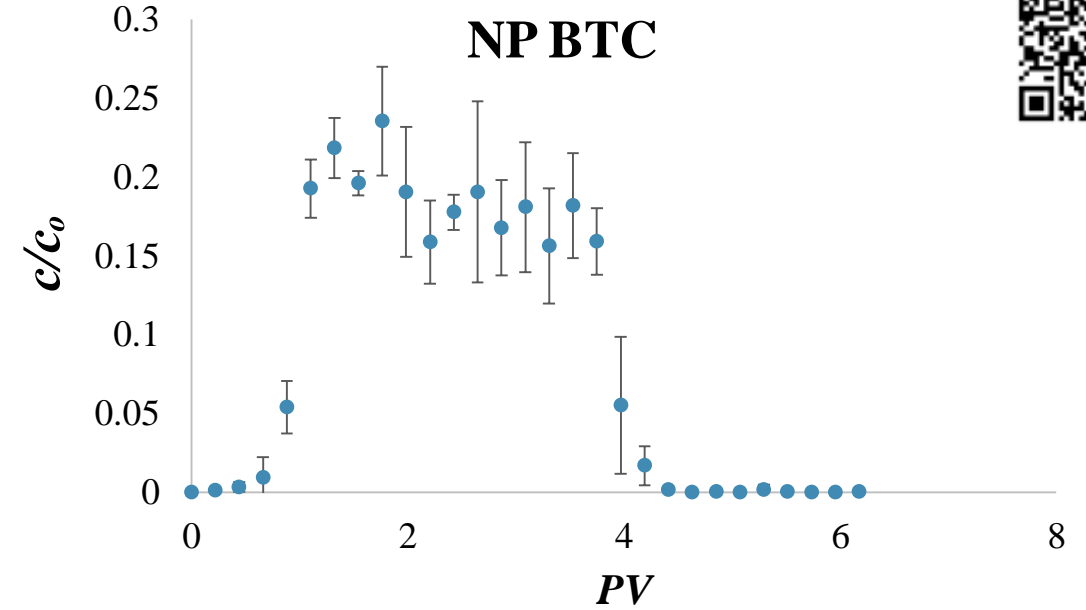
## $n\text{TiO}_2$ transport experiment:



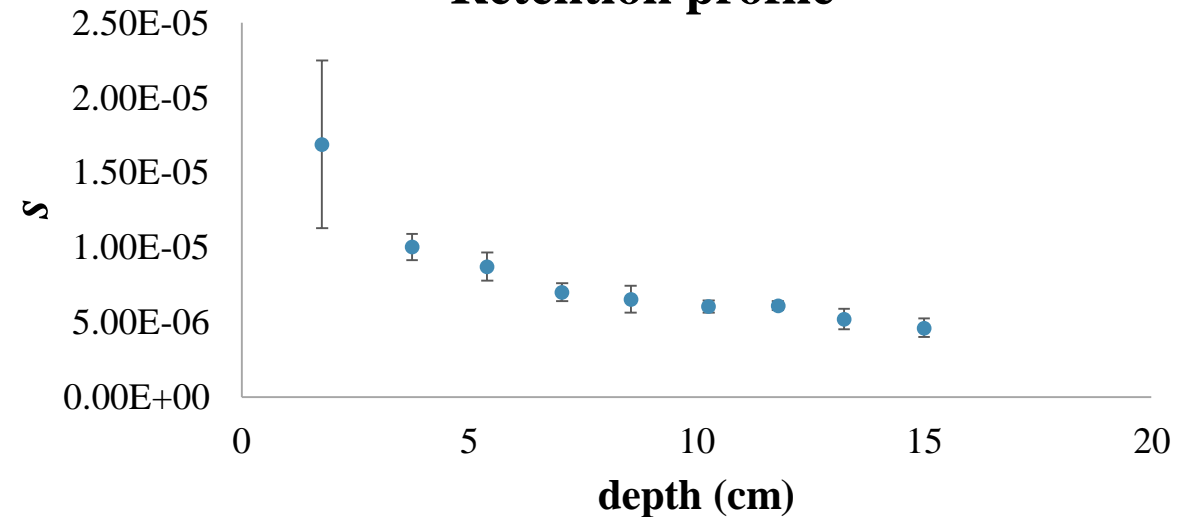
**Tracer BTC**



**NP BTC**



**Retention profile**



# Mathematical Model:

## Ripening mechanism observed:

$$\frac{\partial c}{\partial t} = D \frac{\partial^2 c}{\partial x^2} - v \frac{\partial c}{\partial x} - k_{att} \psi c + \frac{\rho_b}{\theta} k_{det} s$$

$$\frac{\rho_b}{\theta} \frac{\partial s}{\partial t} = k_{att} \psi c - \frac{\rho_b}{\theta} k_{det} s$$

$$\psi = 1 + A s^\beta$$

Initial condition:  $c(x, 0) = 0, s(x, 0) = 0$

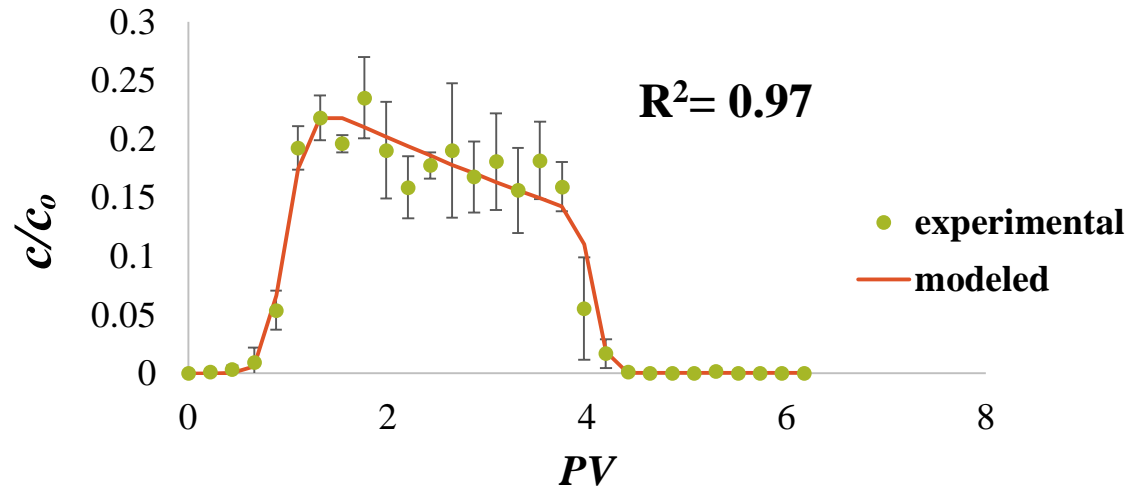
Boundary condition:  $c(0, t) = c_o \ (t < t_0)$   
 $c(0, t) = 0 \ (t \geq t_0)$   
 $-D \frac{\partial c}{\partial x}(L, t) = 0$

Solution obtained through COMSOL



- $\theta$  is the porosity of porous media before the injection of the particles [-],
- $k_{att}$  is the first-order coefficient of particle attachment in the secondary minimum ( $T^{-1}$ )
- $k_{det}$  is the first-order coefficient of particle detachment from the secondary minimum ( $T^{-1}$ )
- $\psi$  is dimensionless function accounting for particle ripening
- $\beta=1$ , is an empirical parameter that controls the shape of the colloid spatial distribution
- $A$  multiplier coefficient

## Comparison of simulated and experimental BTC and retention profile:



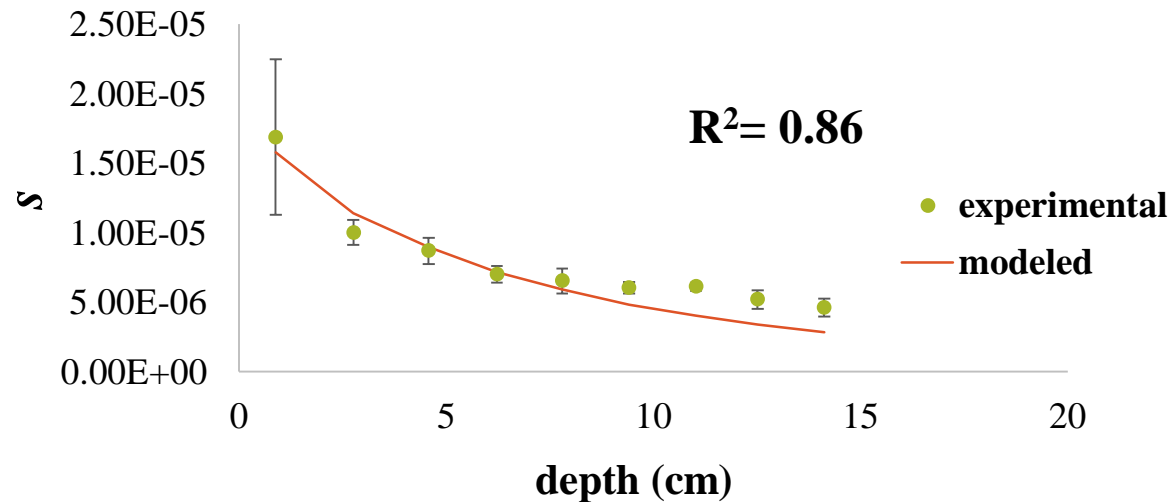
Optimized parameter values using COMSOL:

$$k_{att} = 0.052926/\text{min}$$

$$k_{det} = 1 \times 10^{-5}/\text{min}$$

$$A = 50997$$

### Breakthrough curve for nTiO<sub>2</sub> through porous media



### Retention profile for nTiO<sub>2</sub> through porous media

## Conclusion:



- $\text{nTiO}_2$  transport through porous media has been observed to follow ripening mechanism.
- Experimental results were fitted with a 1D model accounting for ripening.

## Future Work:

- Column experiments for studying the transport of  $\text{nTiO}_2$  in the presence of biofilm is going on, followed by developing a mathematical model.



# References:



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**Thank you**