

**Table 1.** Weather Research and Forecasting model setting

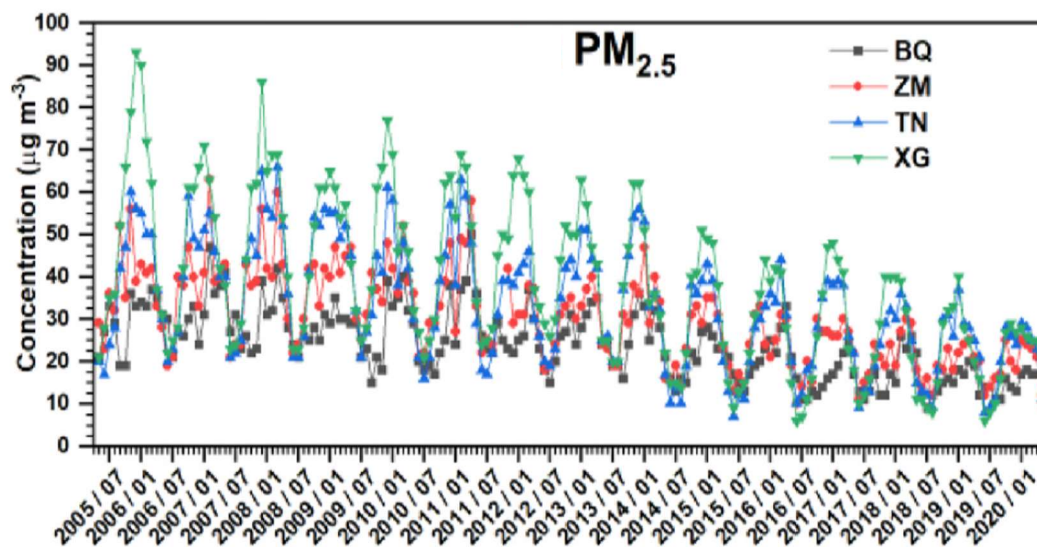
Parameters	WRF Setting
Version	WRF v3.7.1
Microphysics	WSM 5-class scheme
Cumulus Parameterization	Kain-Fritsch
Planetary Boundary Layer	YSU scheme
Surface Layer	MM5 Monin-Obukhov scheme
Land Surface	Unified Noah land-surface model
Urban Surface	No
Longwave Radiation	cam scheme
Shortwave Radiation	cam scheme
SST_update	Yes

**Table 2.** Community Multiscale Air Quality model setting

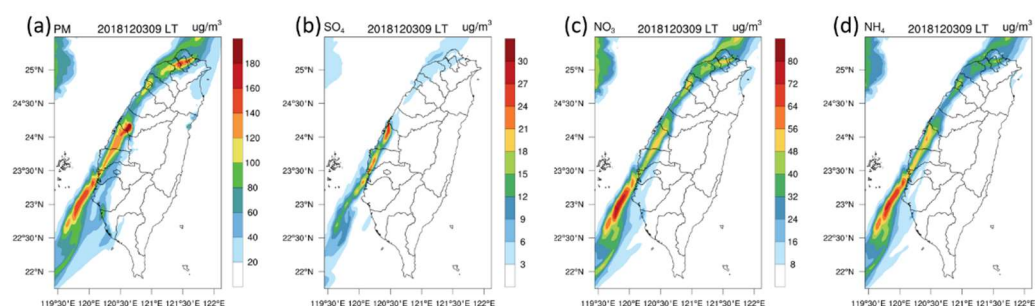
Parameters	CMAQ Setting
Version	CMAQ v5.2.1
Chemical mechanism	Cb06
Horizontal advection	Yamo
Vertical advection	WRF input
Horizontal mixing/diffusion	Multiscale
Aerosol	Aero 6
Cloud option	Acm ae6
Emission	TEDS 9.0

**Table 3.** Model domain and resolution setting

		Domain 1	Domain 2	Domain 3	Domain 4
WRF	Vertical Layer	45	45	45	45
	Grid Size	91×91	166×169	223×223	223×223
	FDDA	Yes	Yes	Yes	No
CMAQ	Resolution	81 km	27 km	9 km	3 km
	Vertical Layer	6	15	15	15
	Grid Size	70×80	70×80	70×80	90×135

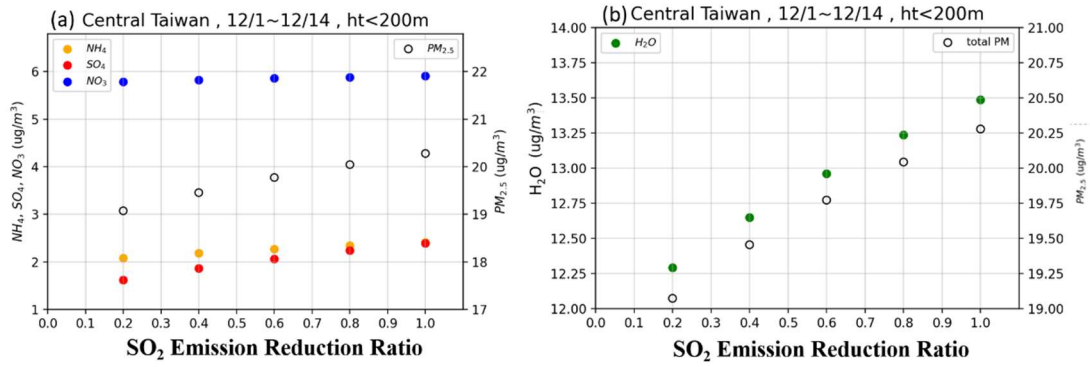


**Fig. 1.** Long-term trends of PM<sub>2.5</sub> and gaseous pollutants monitored at the Banqiao, Zhongming, Tainan, and Xiaogang stations. (Chuang et al., 2021)



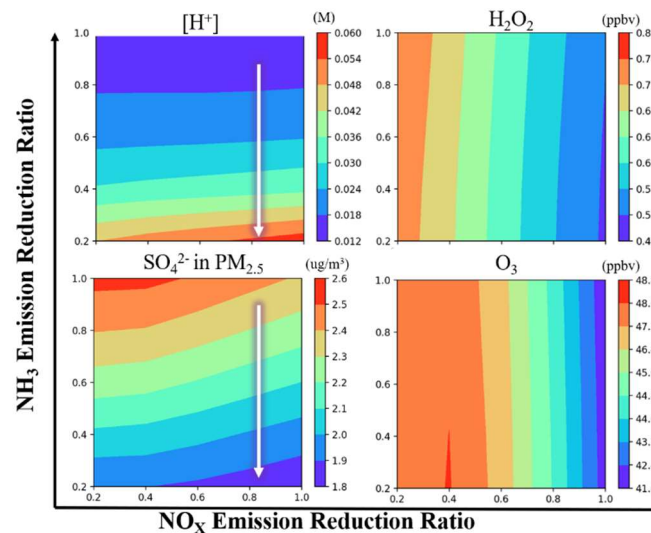
**Fig. 2.** Surface layer concentration in Taiwan of CTL run. Select a representative distribution. (a) the total PM concentration. (b) sulfate (c) nitrate (d) ammonium formation in PM.

PM and its secondary inorganic formations show similar spatial patterns, with the highest concentrations occurring in the western region of Taiwan. This area is characterized by flat terrain, high population density, and intense industrial activities, making it a major residential and industrial agglomeration zone.

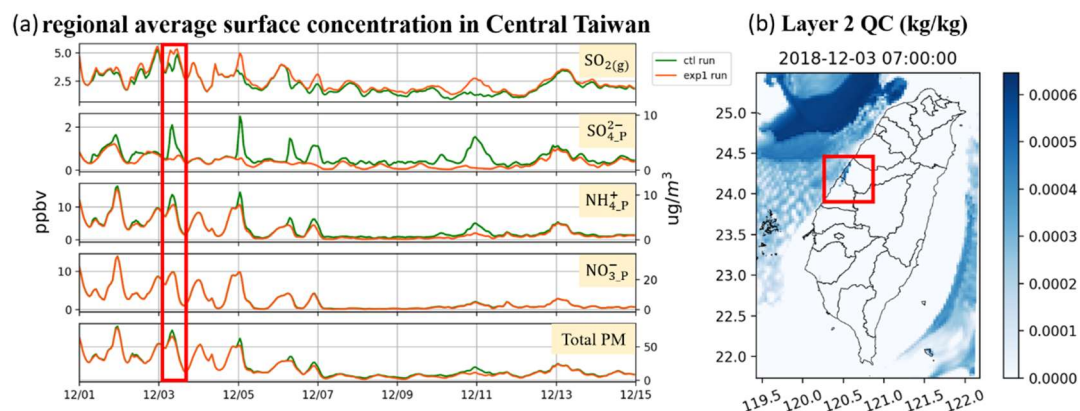


**Fig. 3.** CTL and EXP2 run ( $\text{SO}_2$  emission adjustment experiments). (a) sulfate, nitrate, ammonium formation, and total  $\text{PM}_{2.5}$  surface layer average concentration in Central Taiwan. (b) liquid water formation and total  $\text{PM}_{2.5}$  surface layer average concentration in Central Taiwan.

As  $\text{SO}_2$  emission reduces, sulfate has the most significant decrease. To maintain the pH balance, ammonium also decreases. The slight change in nitrate is attributed to the decline in sulfate and ammonium, which leads to a reduction in water in PM. As a result, the amount of dissolved nitric acid also decreases.



**Fig. 4.** CTL run and EXP3 run ( $\text{NO}_x$  and  $\text{NH}_3$  emission reduction separately or simultaneously experiments). Surface layer average concentration of proton in  $\text{PM}_{2.5}$ , sulfate in  $\text{PM}_{2.5}$ , gas phase hydrogen peroxide and gas phase ozone in Central Taiwan. The proton and sulfate in  $\text{PM}_{2.5}$  decrease as the  $\text{NH}_3$  emission reduces. But the hydrogen peroxide and ozone do not have the same pattern.



**Fig. 5.** We select a grid point with both strong aqueous phase chemistry and high cloud water. Specifically, we chose a grid point located on the second layer (approximately 64.54 meters above sea level) in the model, situated on the Taichung coast (120.5053E, 24.203N) at 7:00 am local time on December 3, 2018.