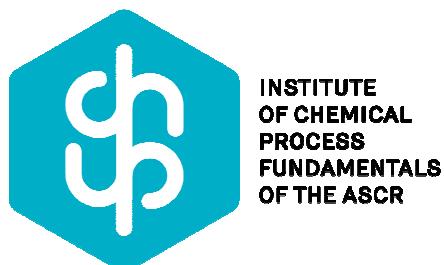


# Source apportionment of fine aerosol at a rural background site in Central Europe based on seasonal distributions of dicarboxylic acids, sugars and related compounds

Petr Vodička<sup>1,2</sup>, Kimitaka Kawamura<sup>2</sup>, Dhananjay K. Deshmukh<sup>2</sup>, Petra Pokorná<sup>1</sup>,  
Jaroslav Schwarz<sup>1</sup>, Vladimír Ždímal<sup>1</sup>

<sup>1</sup> Institute of Chemical Process Fundamentals, Czech Academy of Sciences, Prague, Czech Republic

<sup>2</sup> Chubu Institute for Advanced Studies, Chubu University, Kasugai 487–8501, Japan



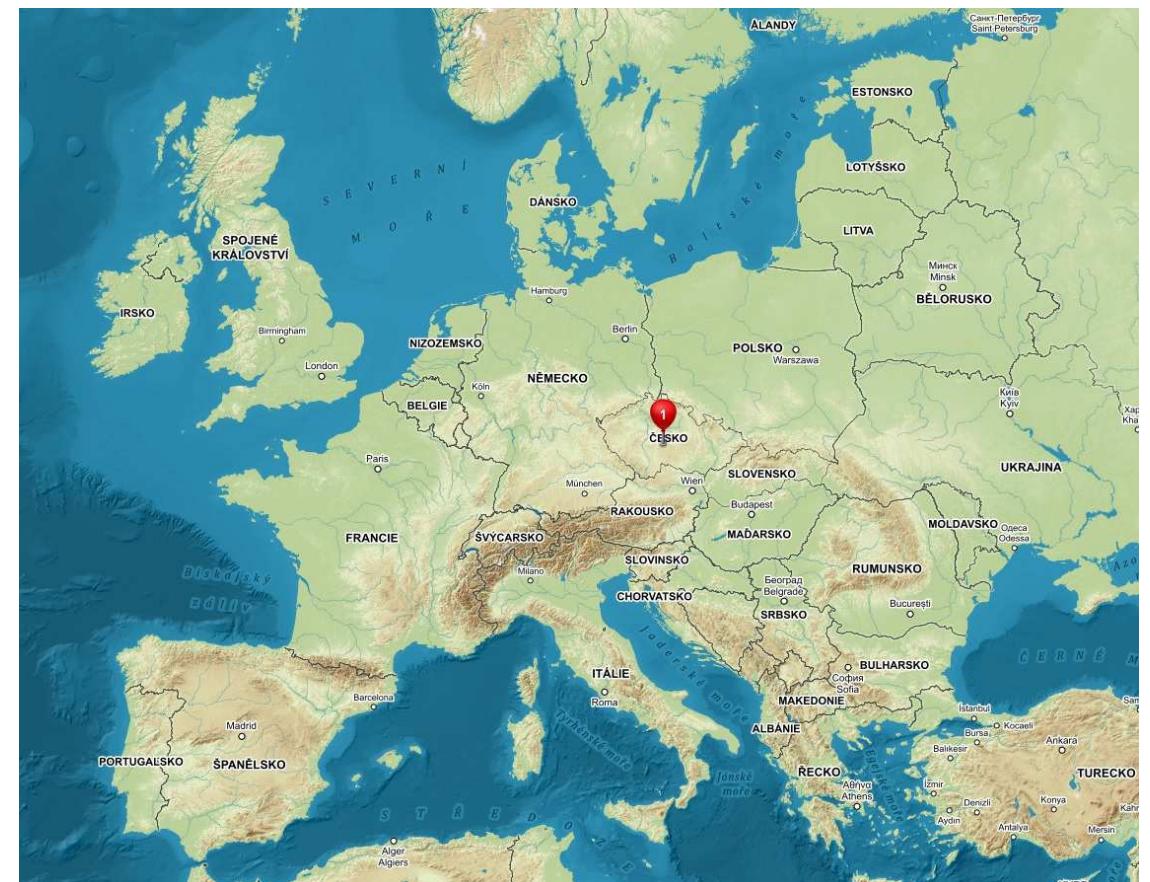


# Site – Košetice Observatory

Central European rural background site

Characteristic: agricultural landscape and forests, out of range of major sources of pollution

location: N: 49° 34' 24.13" E: 14° 4' 49.67" Altitude: 534 m ASL



# Sampling

- PM1 aerosol fraction, quartz fiber filters
- Period: 27 Sep 2013 – 9 Aug 2014 (every 2<sup>nd</sup> day)
- 146 samples with 24-h time resolution + blanks



# Analyses



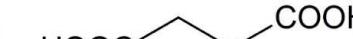
## Normal-chain diacids



Oxalic, C<sub>2</sub>



Malonic, C<sub>3</sub>



Succinic, C<sub>4</sub>



Glutaric, C<sub>5</sub>



Adipic, C<sub>6</sub>



Pimelic, C<sub>7</sub>



Suberic, C<sub>8</sub>

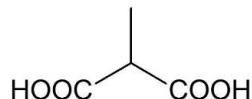


Azelaic, C<sub>9</sub>

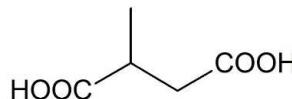


Undecanedioic, C<sub>11</sub>

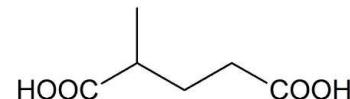
## Branched-chain diacids



Methylmalonic, iC<sub>4</sub>

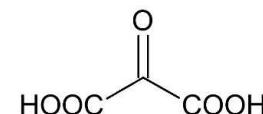


Methylsuccinic, iC<sub>5</sub>

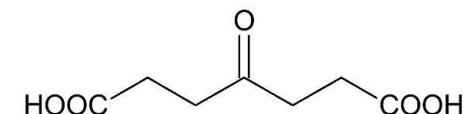


2-Methylglutaric, iC<sub>6</sub>

## Ketodiacids



Ketomalonic, kC<sub>3</sub>

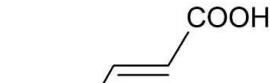


4-Ketopimelic, kC<sub>7</sub>

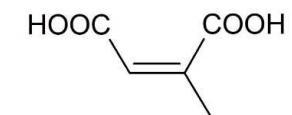
## Unsaturated diacids



Maleic (cis), M

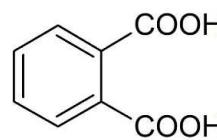


Fumaric (trans), F

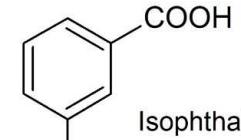


Methylmaleic (Citraconic), mM

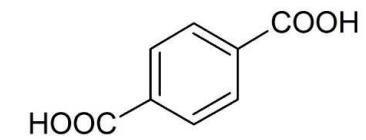
## Aromatic diacids



Phthalic, Ph



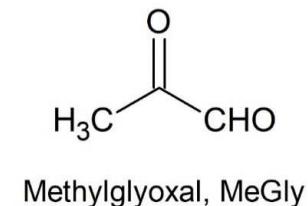
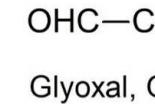
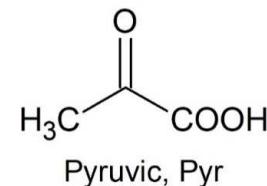
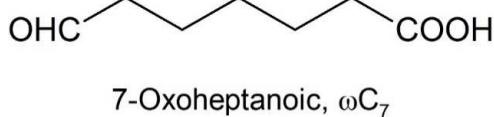
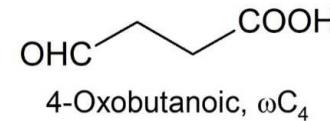
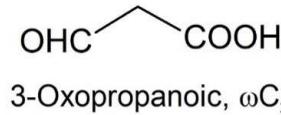
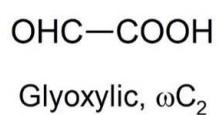
Isophthalic, iPh



Terephthalic, tPh

# Analyses

Diacid's  
related  
compounds:



Saccharides:

- anhydrosugars (levoglucosan, mannosan, galactosan)
- sugar alcohols (arabitol, mannitol, erythritol)
- primary sugars (glucose, fructose, mannose, galactose, sucrose)

Total PM<sub>1</sub> mass and Carbon → EC + OC

Water-soluble ions: SO<sub>4</sub><sup>2-</sup>, NO<sub>3</sub><sup>-</sup>, Cl<sup>-</sup>, oxalate NH<sub>4</sub><sup>+</sup>, K<sup>+</sup> → LWC calculation

Meteorology: Wind speed + direction, RH, Temperature, Global radiation

# Overview

Year-round averages [%]:

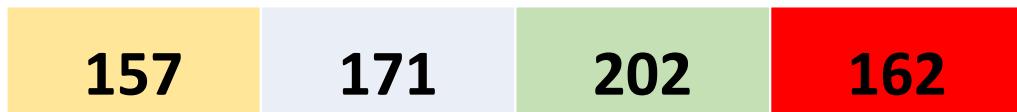
$$\text{OC/PM}_1 = 25.1$$

$$\Sigma \text{tDA-C/OC} = 2.14$$

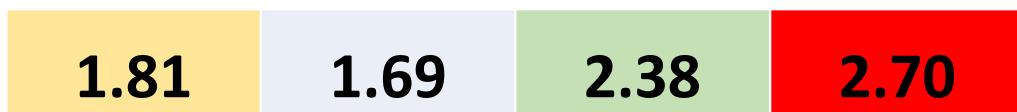
Avg. seasonal distribution



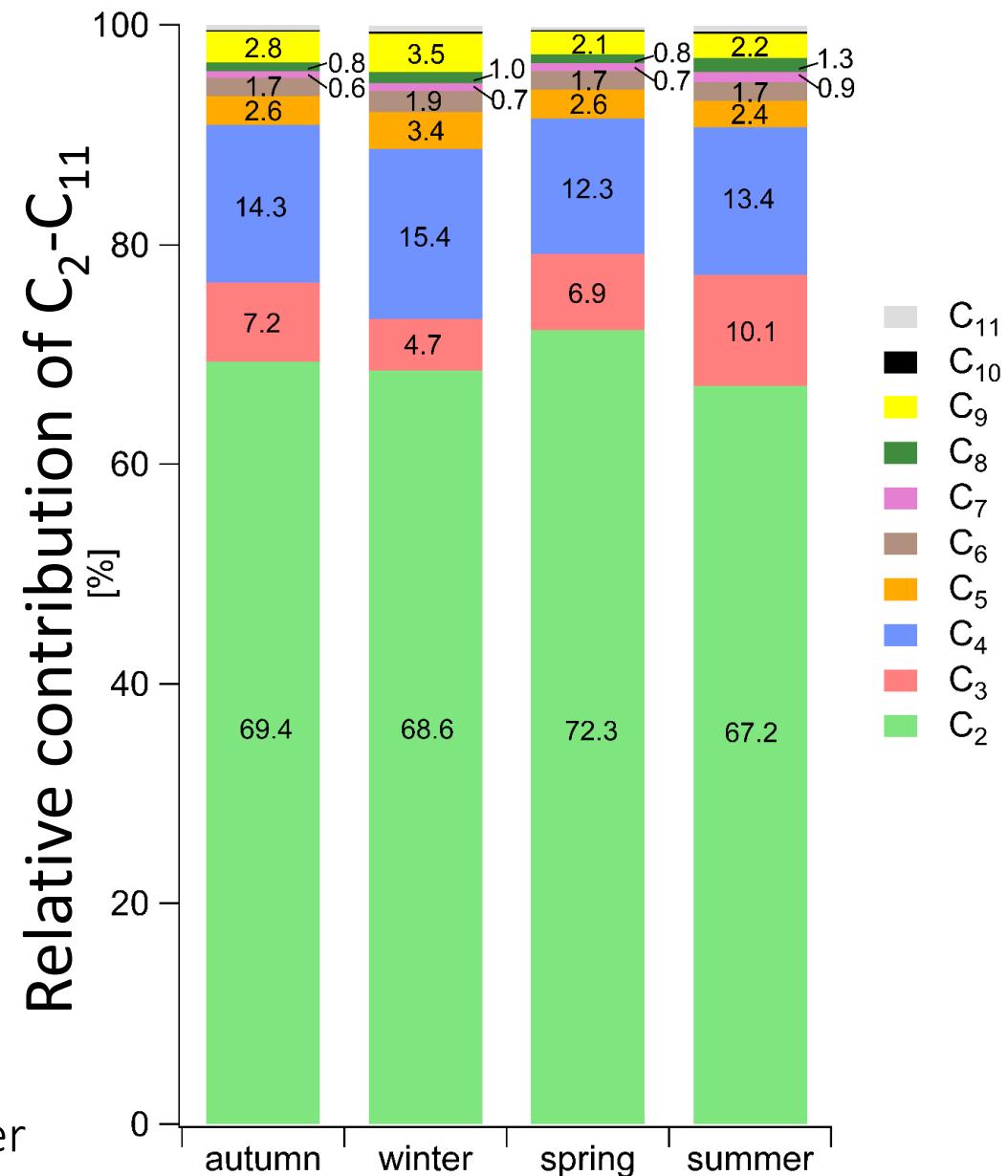
$\Sigma \text{tDA}$  [ng/m<sup>3</sup>]



$\Sigma \text{tDA-C/OC}$  [%]



More intense SOA formation and aging in summer



Normal-chain diacids vs. LWC, T and O <sub>3</sub> correlations (r)		Autumn			Winter			Spring			Summer		
		LWC	T	O <sub>3</sub>	LWC	T	O <sub>3</sub>	LWC	T	O <sub>3</sub>	LWC	T	O <sub>3</sub>
Oxalic	C <sub>2</sub>	<b>0.64</b>	0.22	0.03	<b>0.81</b>	-0.26	0.08	<b>0.65</b>	0.20	<b>0.41</b>	<b>0.53</b>	<b>0.59</b>	<b>0.53</b>
Malonic	C <sub>3</sub>	<b>0.43</b>	<b>0.42</b>	0.20	<b>0.79</b>	-0.23	0.13	<b>0.51</b>	0.26	<b>0.42</b>	<b>0.51</b>	<b>0.56</b>	<b>0.54</b>
Succinic	C <sub>4</sub>	<b>0.59</b>	0.21	-0.01	<b>0.76</b>	-0.28	0.09	<b>0.60</b>	0.10	0.25	<b>0.49</b>	<b>0.54</b>	<b>0.59</b>
Glutaric	C <sub>5</sub>	<b>0.64</b>	0.06	-0.08	<b>0.65</b>	-0.19	0.14	<b>0.62</b>	-0.02	0.22	<b>0.53</b>	<b>0.55</b>	<b>0.65</b>
Adipic	C <sub>6</sub>	<b>0.58</b>	0.10	0.03	<b>0.64</b>	-0.24	0.17	<b>0.54</b>	0.03	0.29	<b>0.44</b>	<b>0.65</b>	<b>0.59</b>
Pimelic	C <sub>7</sub>	<b>0.63</b>	0.05	0.00	<b>0.63</b>	-0.18	0.14	<b>0.41</b>	0.06	0.20	<b>0.24</b>	<b>0.58</b>	<b>0.63</b>
Suberic	C <sub>8</sub>	<b>0.47</b>	0.05	0.07	<b>0.42</b>	-0.14	0.04	0.29	0.26	<b>0.33</b>	0.11	<b>0.82</b>	<b>0.53</b>
Azelaic	C <sub>9</sub>	<b>0.51</b>	-0.20	-0.08	<b>0.38</b>	-0.19	-0.08	<b>0.37</b>	0.04	0.15	0.09	<b>0.63</b>	<b>0.38</b>
Sebacic	C <sub>10</sub>	-0.18	-0.21	-0.01	<b>0.48</b>	-0.32	-0.01	<b>0.54</b>	-0.37	0.04	0.11	<b>0.58</b>	<b>0.71</b>
Undecanedioic	C <sub>11</sub>	0.28	0.30	0.13	<b>0.46</b>	-0.16	-0.09	0.20	<b>0.32</b>	<b>0.33</b>	0.20	<b>0.70</b>	<b>0.44</b>

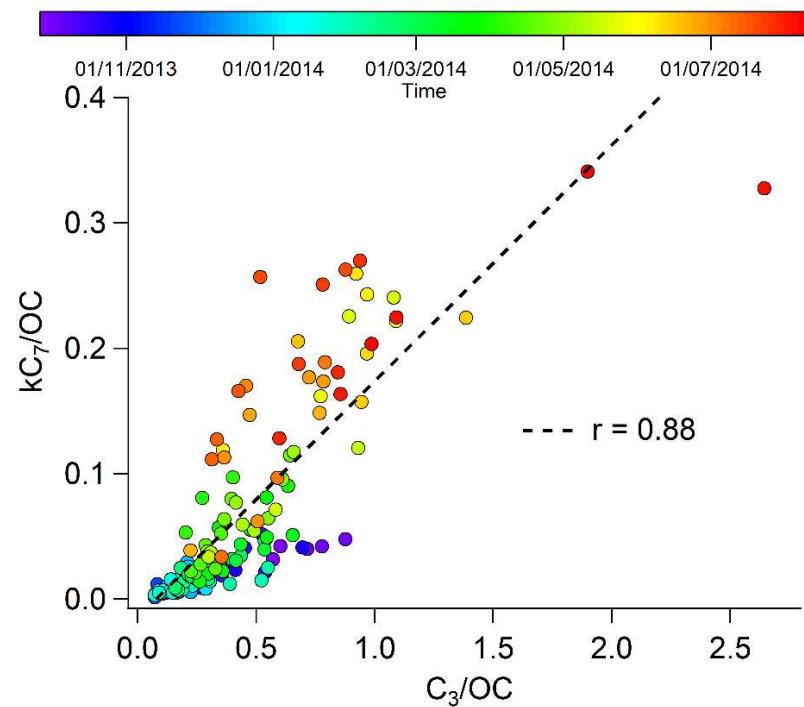
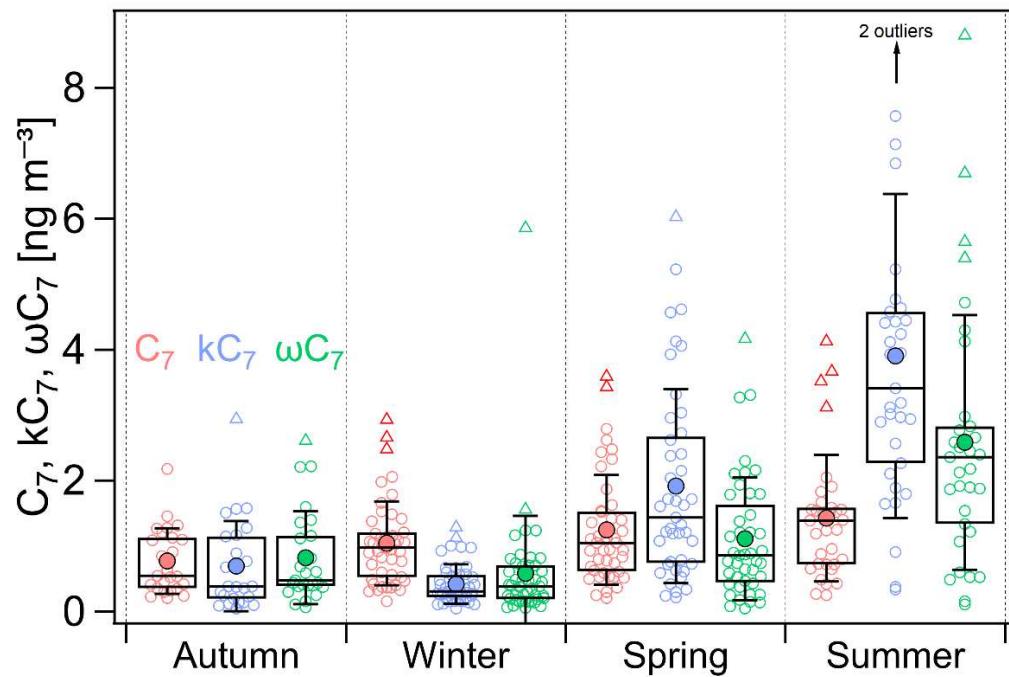
# Correlation analyses (r)

- Temp., GR ... rough proxy of seasonal changes
- O<sub>3</sub> ... indicator of the strength of photooxidation
- RH, LWC ... indication of aqueous phase reactions

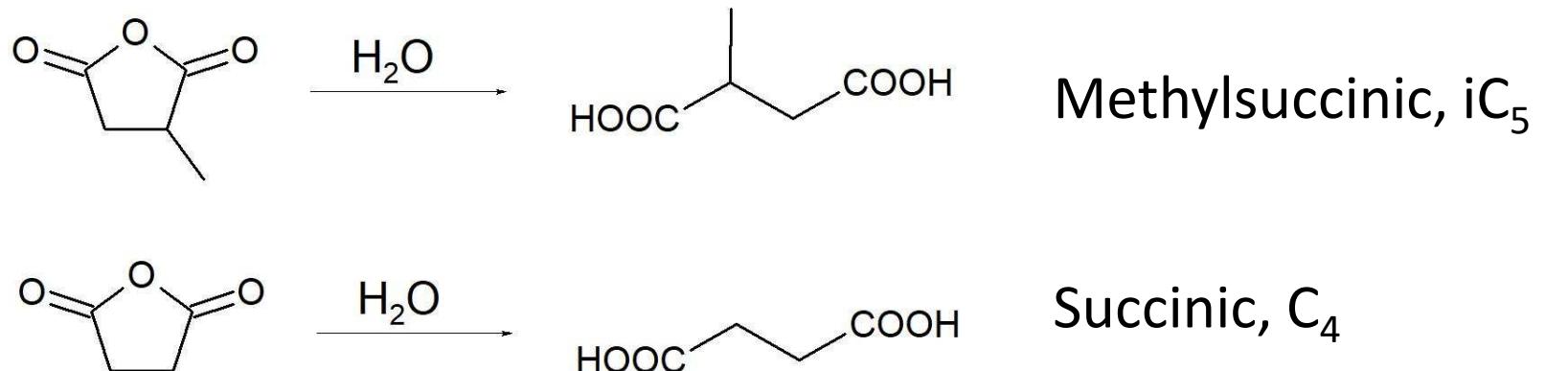
	Year correlations	Temperature	Global radiation	RH	O <sub>3</sub>	LWC
Malonic	C <sub>3</sub>	0.47	0.55	-0.41	0.49	0.34
Methylmalonic	iC <sub>4</sub>	0.56	0.52	-0.39	0.43	0.17
4-Ketopimelic	kC <sub>7</sub>	0.77	0.79	-0.61	0.69	0.06
7-Oxoheptanoic	ωC <sub>7</sub>	0.59	0.57	-0.46	0.51	0.12
Methylsuccinic	iC <sub>5</sub>	-0.64	-0.48	0.42	-0.43	0.65
Maleic	M	-0.72	-0.59	0.45	-0.50	0.55
Methylmaleic	mM	-0.72	-0.59	0.41	-0.48	0.52
Levoglucosan	LVG	-0.70	-0.54	0.42	-0.48	0.60

# 7-C acids

- formed probably by oxidation of (n-7) unsaturated fatty acids
- (n-7) fatty acids abundant in *Brassica napus* (Mukherjee & Kiewitt, 1980, *Planta*)
- markers of biogenic sources?



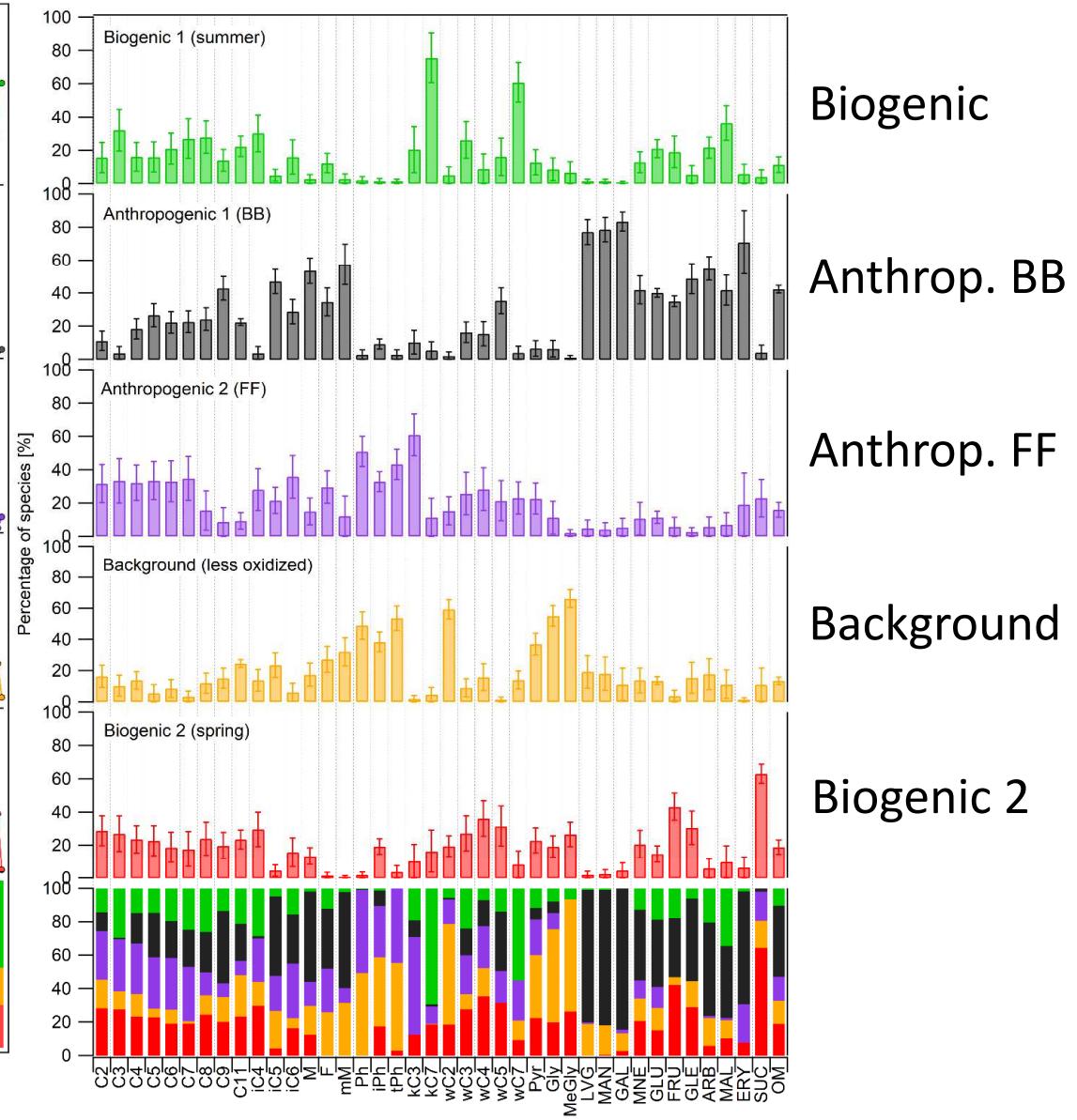
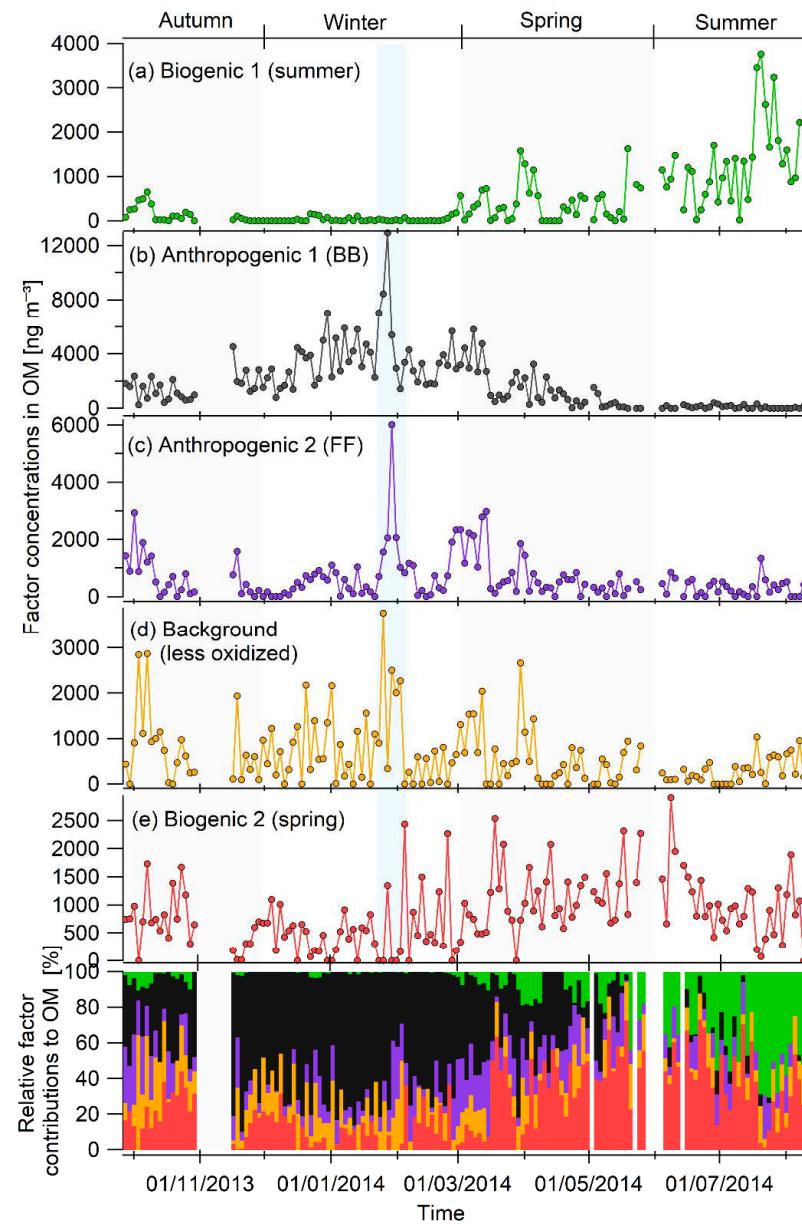
# Hydrolysis of furandiones



Year correlations	Temperature	Global radiation	RH	$O_3$	LWC
$iC_5$	-0.64	-0.48	0.42	-0.43	0.65
M	-0.72	-0.59	0.45	-0.50	0.55
mM	-0.72	-0.59	0.41	-0.48	0.52
LVG	-0.70	-0.54	0.42	-0.48	0.60

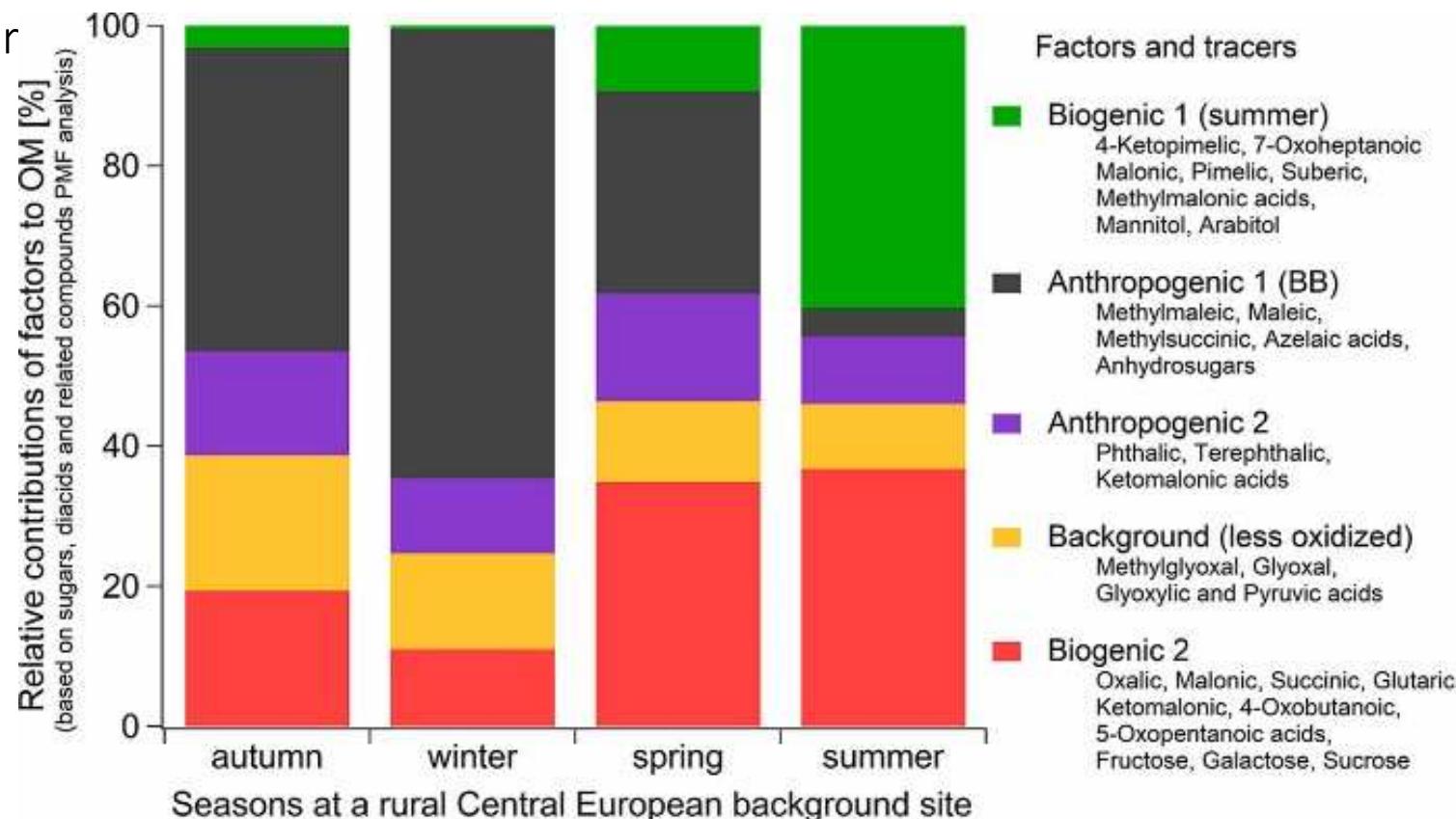
Furandiones: products of photooxidation of anthropogenic VOC (Al-Naiema et al. 2017, Atmos. Environ.)

PMF



# Conclusions

- Clear difference between winter and summer diacids composition in PM<sub>1</sub>
- Biogenic, anthropogenic and background factors resolved by PMF analysis
- Methylsuccinic, aromatic and unsaturated aliphatic acids as typical anthropogenic tracers
- Organic acids with seven and three carbons determined as biogenic tracers
- Two mechanisms of SOA formation - preferred aqueous-phase in winter, while gaseous-phase in summer.



## Reference:



### Anthropogenic and biogenic tracers in fine aerosol based on seasonal distributions of dicarboxylic acids, sugars and related compounds at a rural background site in Central Europe

Petr Vodička <sup>a,b,\*</sup>, Kimitaka Kawamura <sup>b,\*\*</sup>, Dhananjay K. Deshmukh <sup>b,1</sup>, Petra Pokorná <sup>a</sup>, Jaroslav Schwarz <sup>a</sup>, Vladimír Ždímal <sup>a</sup>

<sup>a</sup> Institute of Chemical Process Fundamentals, Czech Academy of Sciences, Rozvojová 1/135, 165 00, Prague 6, Czech Republic

<sup>b</sup> Chubu Institute for Advanced Studies, Chubu University, 1200 Matsumoto-cho, Kasugai, 487-8501, Japan

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Thank you for your attention!