Testing the use of Sodium Polytungstate for an efficient extraction of microplastic particles from river sediments



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Introduction

Rivers:

- Primary pathways of MPs to the sea^{1, 2, 3},
- Reservoirs of MPs stored in sediments^{4,5}.

Kowing microplastics content in river sediments is critical.

The optimization of a methodology for the separation and classification of microplastics from sediments is crucial.



Several studies highlight that density separation using a high-density concentrated saline solution is one of the most reliable and efficient separation methods^{6, 7, 8, 9}.

Aim of the Study

Testing the use of Sodium Polytungstate as a density separation agent.

What is the recovery rate of this method?

References

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Methodology

1. Creation of artificial sediments

20-25 grams of sediment plus 50 MPs per sample for a total of 52 samples

Plastic-free sediments (Pleistocene alluvial deposits)



Selected & added microplastic particles



PET films $(\rho = 1.38 \text{ g} \cdot \text{cm}^{-3})$



PET fibres $(\rho = 1.38 \text{ g} \cdot \text{cm}^{-3})$

2. Organic matter removal

Wet Peroxide Oxidation^{10,11}



3. Density separation

- **Sodium Polytungstate** (ρ from 3.1 g \cdot cm⁻³ to 1.6 g \cdot cm⁻³)
- Indermediate density between sediments and plastic^{12,13},
- Non-toxic^{12,13},
- Easily recoverable and reusable^{12,13},
- Successfully used for the separation of minerals and rocks¹³.

The procedure:

- Addition of the liquid to the samples, centrifugation and settling, - Collection of the suspended particulate matter,
- Transfer of the suspension to cellulose filters,
- Microplastics counting under a stereomicroscope.







Preliminary Results

Used particle size ranges

- Mixture A: 63 - 250 µm

- Mixture B: lower than 63 µm
- Mixture C: 50% A + 50% B

PVC fragments $(\rho = 1.4 \text{ g} \cdot \text{cm}^{-3})$





Procedure 1





Procedure 1

- Low recovery rate due to:
- MPs trapped in sediments,
- MPs dispersion during different steps.
- Too strong organic matter removal,

Procedure 2

- High recovery rate due to:
- Use of the same container for all the steps,
- Larger containers: 500 ml,
- Lower amount of sediments (20 g).

Conclusions

- Test tube capability (50 ml) vs the amount of sediments (25 g). - Destruction of the several layers of wich glitters are made of.