



On the potential use of highly oxygenated organic molecules (HOMs) as indicators for ozone formation sensitivity

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Introduction

- Ozone (O₃) protects lives from harm of solar ultraviolet (UV) radiation in the stratosphere but is toxic in the troposphere [1].
- Tropospheric O₃ is a key oxidant, and source of other oxidants, for various volatile organic compounds (VOCs) [2].
- Recently, highly oxygenated organic molecules (HOMs) were identified as a new compound group formed from oxidation of many VOCs, making up a significant source of secondary organic aerosol (SOA) [3]. [4].
- The pathways forming HOMs involve autoxidation of peroxy radicals (RO₂), formed in many VOC oxidation reactions.
- RO₂ + NO, followed by NO₂ photolysis, is a net source for O₃ [5]. It also affects HOM distributions by competing with other bimolecular reactions.
- Thus, we aim to assess whether HOM ratios can function as a real-time indicator for O₃ formation sensitivity.**

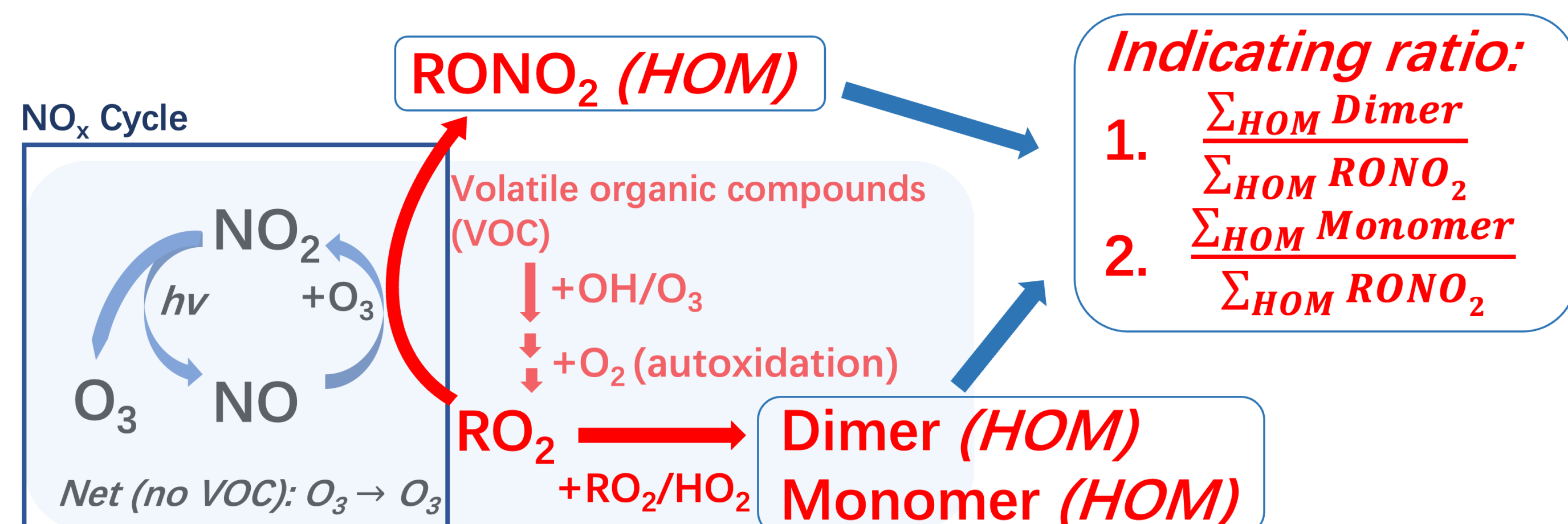


Fig.1 A sketch of the intrinsic connection between HOM and O₃ formation.

Methods

Experiments:

- Monoterpene oxidation experiments, conducted in the COALA chamber [6] (2 m³).
- “Steady-state mode” [7], [8] with continuous flow: each stage of experiments lasted around 3 times the residence time.
- Input: O₃, α-pinene, NO₂, and UV light (400 nm).

Instrumentation:

- NO₃-CIMS [9] (nitrate-adduct Chemical Ionization Mass Spectrometer): online measurements of HOMs.
- PTR (Proton Transfer Reaction Time-of-Flight mass spectrometer): online measurements of α-pinene.
- NO-NO₂ and O₃ analyzers: online NO_x and O₃ measurements.

Model:

- A simple 0-D box model was constructed to exclusively simulate the concentrations of O₃ and its precursors, NO_x and α-pinene, and to generate O₃ isopleth diagrams. It was also used to determine NO₂ photolysis rates.

Conclusions

- Due to the intrinsic connection between the formation pathways of O₃ and HOMs, the ratio of HOM dimers or non-nitrate monomers to HOM organic nitrates could be used to determine O₃ formation regimes (either VOC- or NO_x- limited).
- Given the fast formation and short lifetimes of HOMs, the HOM-based indicating ratios can describe the O₃ formation in real time.

- Despite the success of our approach in this simple laboratory system, the applicability to the much more complex atmosphere remains to be determined.



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Main results

- Both indicating ratios ($\frac{\sum_{HOM} DI}{\sum_{HOM} ON, O>8}$, $\frac{\sum_{HOM} Mono, O>8}{\sum_{HOM} ON, O>8}$) correspond with changes in O₃ concentrations expectedly, suggesting a possible O₃ formation sensitivity (Fig.2). Moreover, the model worked well.
- Model and experimental results (Fig. 3): HOM ratios can qualitatively predict O₃ formation regimes (either VOC- or NO_x- limited).

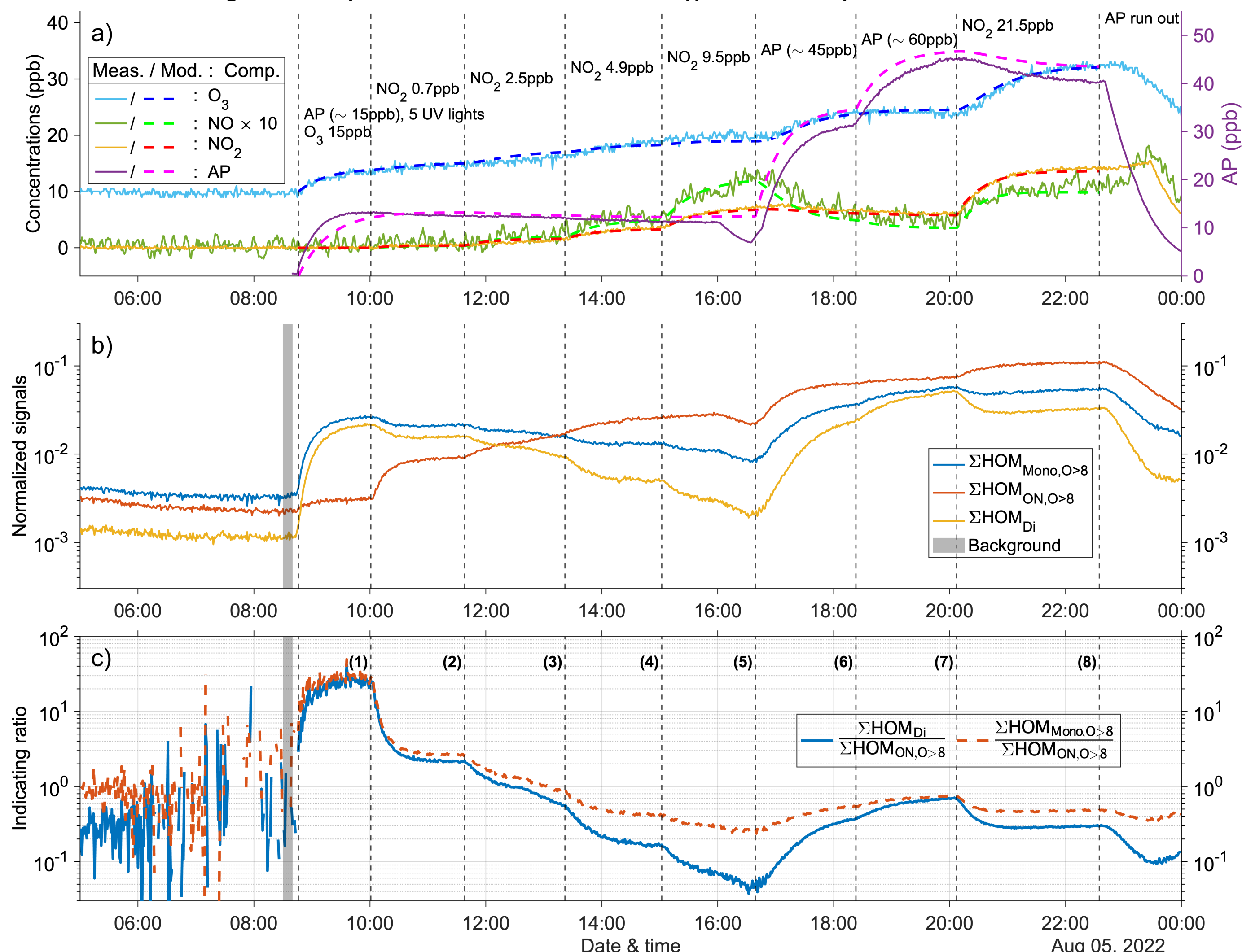


Fig.2 Time series of experiment 2. with 15ppb α-pinene and 15ppb O₃ as initial inputs: a) measured and modeled gas concentrations, b) normalized signals of HOMs, c) indicating ratios.

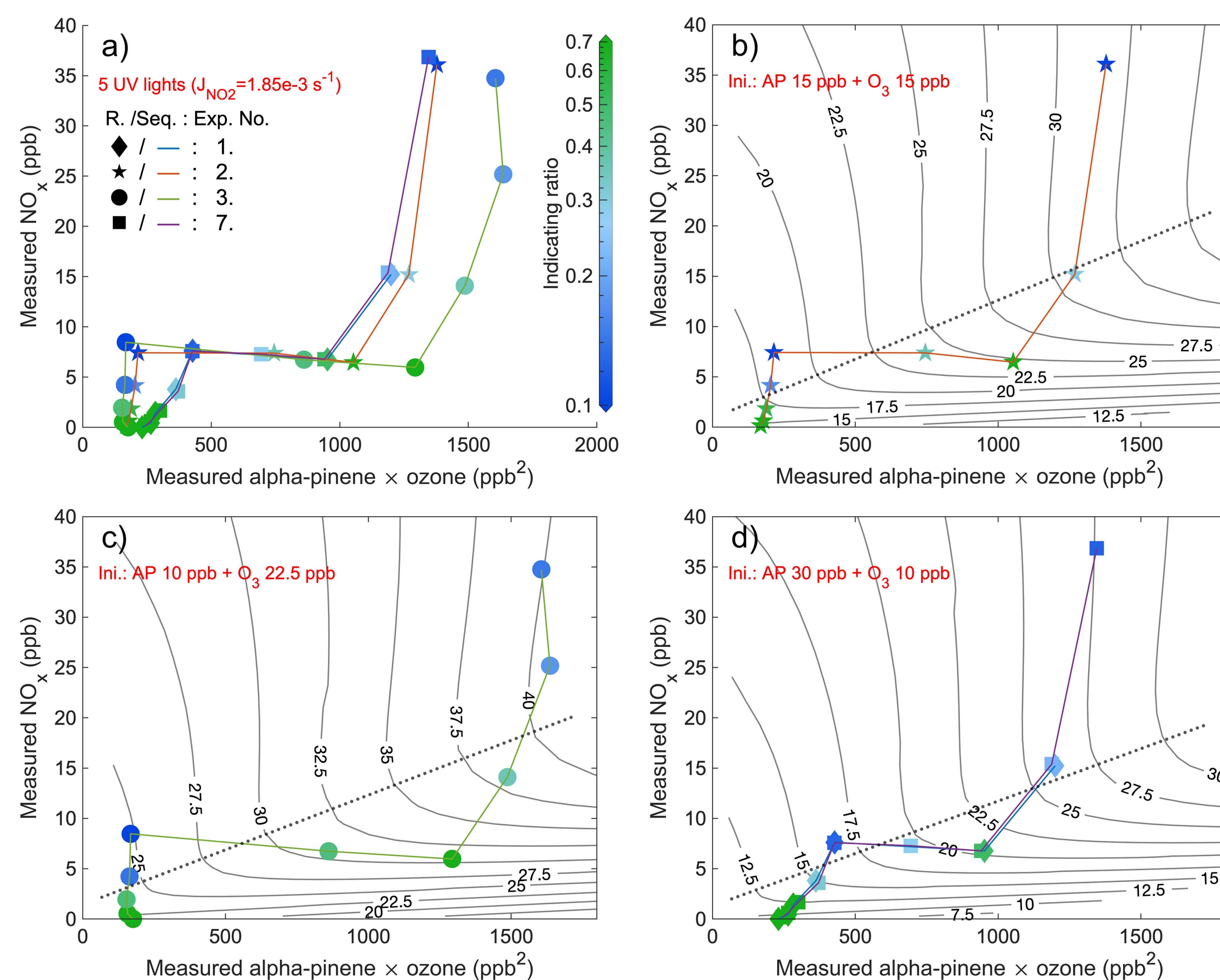


Fig.3 Steady-state indicating ratio $\frac{\sum_{HOM} DI}{\sum_{HOM} ON, O>8}$ of experiments from 4 days with five UV lights. EKMA curves (O₃ isopleths) were simulated by the box model.

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