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Introduction

Water facilitates deformation processes throughout the Earth's crust (Rubie, 1986). Knowledge of how fluid-rock interactions proceed at the grain-scale informs our understanding of larger scale processes.

The calcium sulfate system is used as an accessible analogue to study mineral transformations that occur at elevated temperature and pressure (e.g. mineral transformations in serpentinite, (Heard and Rubey, 1966)). Here we use the calcium sulfate system to perform pressure and temperature controlled cyclical dehydration-rehydration (D-R) experiments using whole rock gypsum.

Our experiments were conducted on the PSICHÉ beamline of the SOLEIL synchrotron using 4D synchrotron computed microtomography (4D µCT). The technique enables direct observation of mineral reactions and textural evolution while the sample is at reaction pressure and temperature.



Figure 2: Mjölnir rig, miniature x-ray translucent triaxial rig used for these experiments. uses 3.2 mm * 10 mm cylindrical samples and controls the confining pressure, axial load and pore fluid pressure independently using hydraulic fluids. This rig features a heating system via band heaters and can measure the sample temperature via thermocouples.



Butler, I., Fusseis, F., Cartwright-Taylor, A., and Flynn, M., 2020, Mjölnir: a miniature triaxial rock deformation apparatus for 4D synchrotron X-ray microtomography: Journal of synchrotron radiation, v. 27, no. 6, p. 1681-1687. Freitas, D., Butler, I. B., Elphick, S. C., Gilgannon, J., Rizzo, R. E., Plumper, O., Wheeler, J., Schleputz, C. M., Marone, F., and Fusseis, F., 2024, Heitt Mjolnir: a heated miniature triaxial apparatus for 4D synchrotron microtomography: Journal of Synchrotron Radiation, v. 31, no. 1, p. 150-161 Rubie, D. C., 1986, The catalysis of mineral reactions by water and restrictions on the presence of aqueous fluid during metamorphism: Mineralogical Magazine, v. 50, no. 357, p. 399-415. || Heard, H. C., and Rubey, W. W., 1966, Tectonic implications of gypsum dehydration: Geological Society of America Bulletin, v. 77, no. 7, p. 741-760.

Imaging the textural evolution of gypsum through a metamorphic cycle



Figure 1: The experimental temperature profile (black line) for dehydration (pink) and rehydration cycles (blue), and the sample height change (blue line) measured by an LVDT. There is sample shortening clearly evident during dehydration and a subtle height decrease during rehydration. Dehydration was carried out at 125°C, while rehydration was allowed to proceed at 50°C.

Dehydration at 125°C

♦ Hydrostatic load was maintained throughout (20 MPa) with an applied pore fluid pressure of 5 MPa.

800 µm

800 µm

Figure 4: Scanning Electron Microscope images of textures and textural relationships from a sample that has undergone a dehydration and rehydration cycle. (A) shows cross-cutting acicular bassanite with porosity between grains. (B) shows bassanite florets which are sections across the acicular crystals. Grains are skeletal with porous cores and are surrounded by moats of porosity. (C) elongate and floret shaped sections of acicular bassanite with gypsum from rehydration filling both grain cores and moat-like porosity. (D) Gypsum infilling leaving only microporosity.

Research Questions

- shrinkage or swelling in the sample?

- during a mineral D-R cycle.
- for preferred grain orientation.
- within the experimental sample.
- fills skeletal pore space and inter-grain pore space.
- early growth not discernible in our data.







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 \diamond How do mineral textures evolve during a D-R metamorphic cycle?

 \diamond Are neoformed mineral textures inherited from the pre-existing phase? \diamond What does gypsum look like as a product phase of re-hydration?

 \diamond The D-R reactions involve significant volumetric changes. By quantifying the total and solid volume change can we understand the consequences of connected porosity and see any evidence of

Rehydration at 50°C

 \diamond These data are the first time resolved tomographic investigation of textural evolution

 \diamond The nucleation, growth and spatial location of both the bassanite phase during dehydration and gypsum phase during rehydration appears random with no evidence

 \diamond Mineral transformation and porosity infilling during rehydration is patchy and localised

 \diamond Gypsum formed during rehydration is distributed across bassanite grain boundaries,

 \diamond Gypsum grains produced by rehydration appear small, <1.3 micron with nucleation and