



# Preparation and calibration of $^{17}\text{O}$ -enriched nitrite isotope standards

Toward reliable  $\Delta^{17}\text{O}$  measurements of atmospheric  $\text{NO}_2$  and HONO

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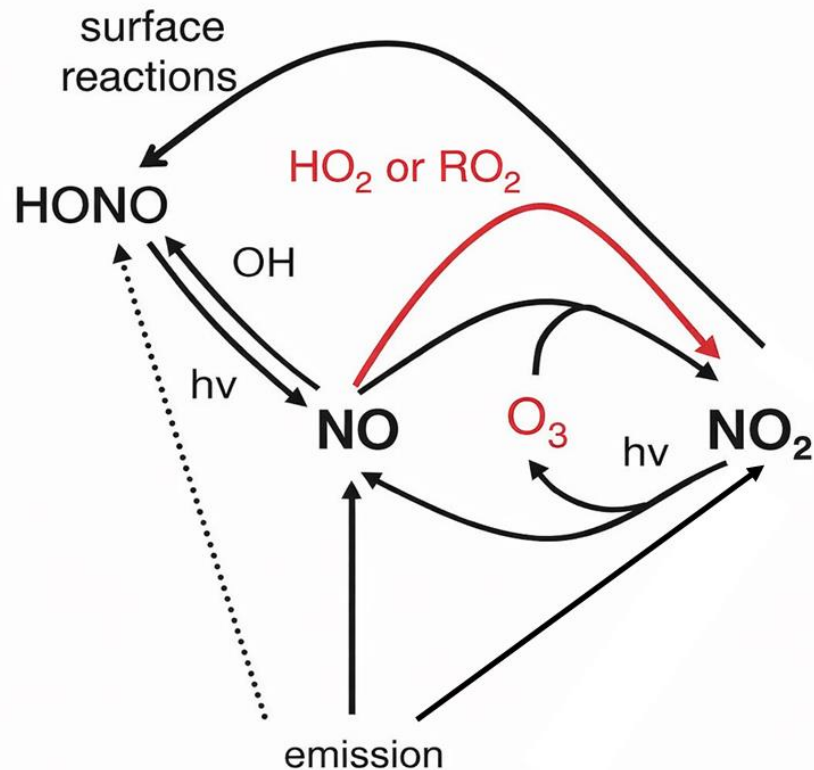
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# Reactive nitrogen has profound effects on atmospheric chemistry

## Coupling between atmospheric reactive nitrogen cycling and radical cycling

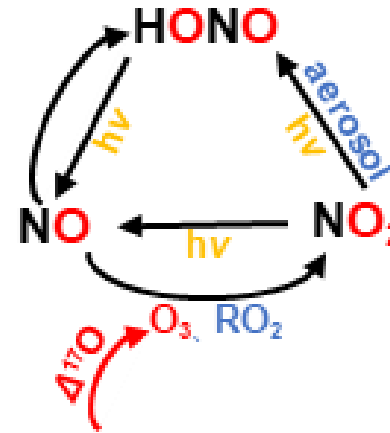
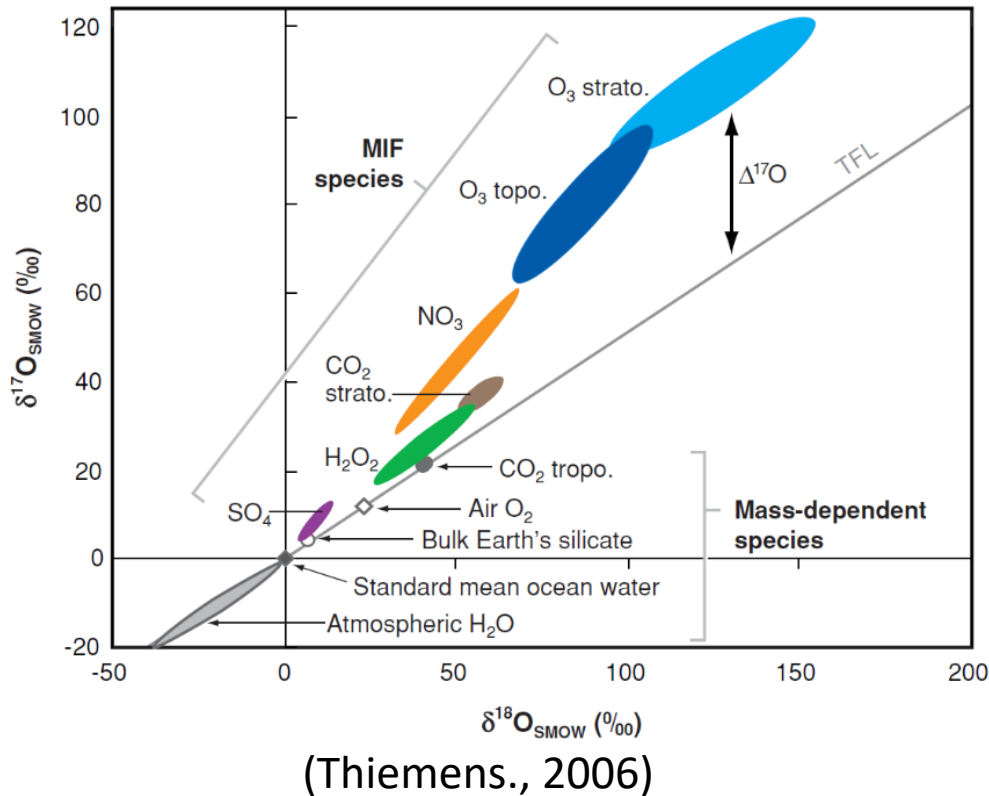


➤ Atmospheric oxidation capacity (AOC) controls the oxidation of key greenhouse gases (e.g., CH<sub>4</sub>) and the formation of secondary pollutants (O<sub>3</sub> and PM<sub>2.5</sub>), regulating their transformation and removal.

➤ The photochemical cycling of reactive nitrogen (NO<sub>x</sub>, HONO) influences O<sub>3</sub>, OH, and peroxy radicals, thereby shaping AOC.

**Understanding the sources and transformation mechanisms of reactive nitrogen is essential for exploring changes in atmospheric oxidative capacity.**

# $\Delta^{17}\text{O}$ is a powerful tracer of reactive nitrogen oxidation pathways



- **O<sub>3</sub> dominated reactions give higher  $\Delta^{17}\text{O}$**
- **RO<sub>2</sub> dominated reactions give lower  $\Delta^{17}\text{O}$**

Oxidant	$\Delta^{17}\text{O}$	reference
O <sub>3</sub>	~26.2 ‰	(Vicars et al., 2014)
RO <sub>2</sub> , OH	~0	(Michalski et al., 2006)

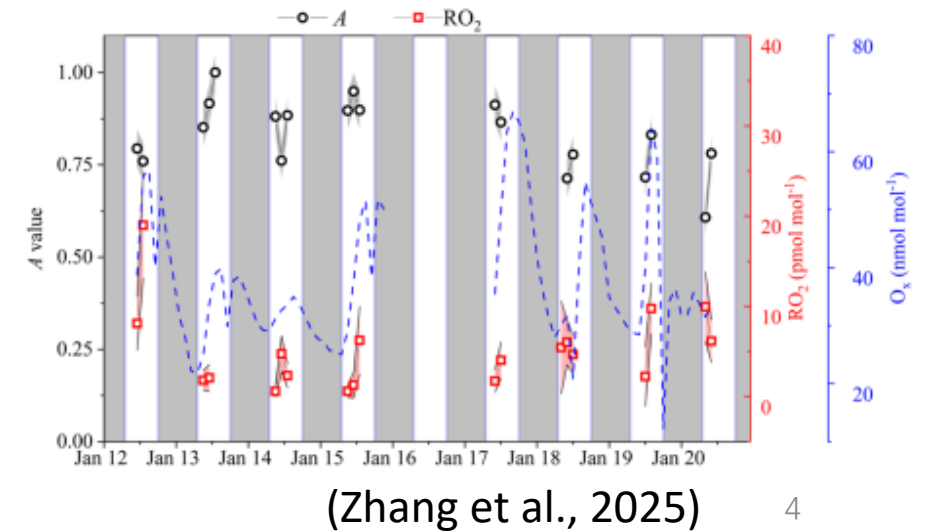
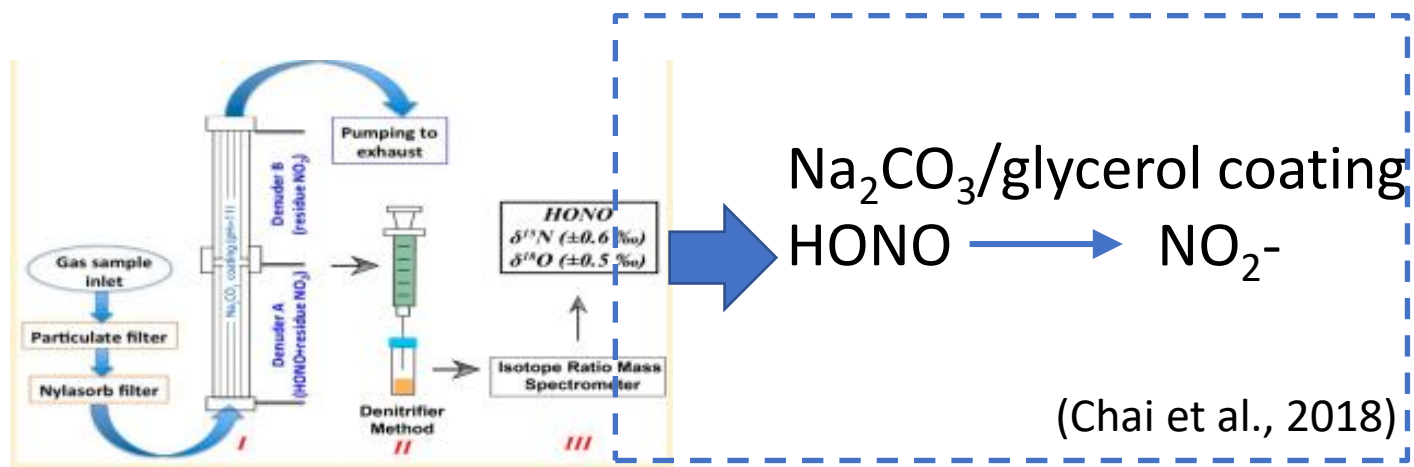
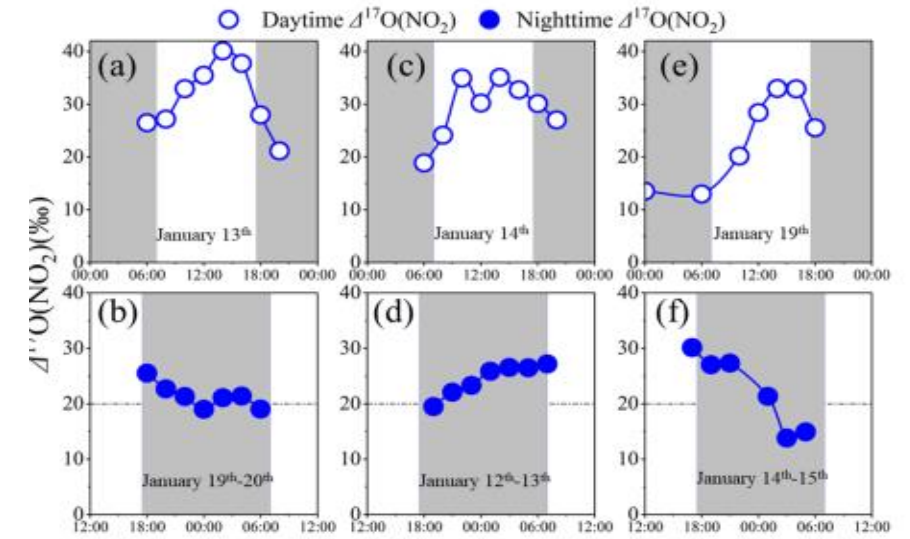
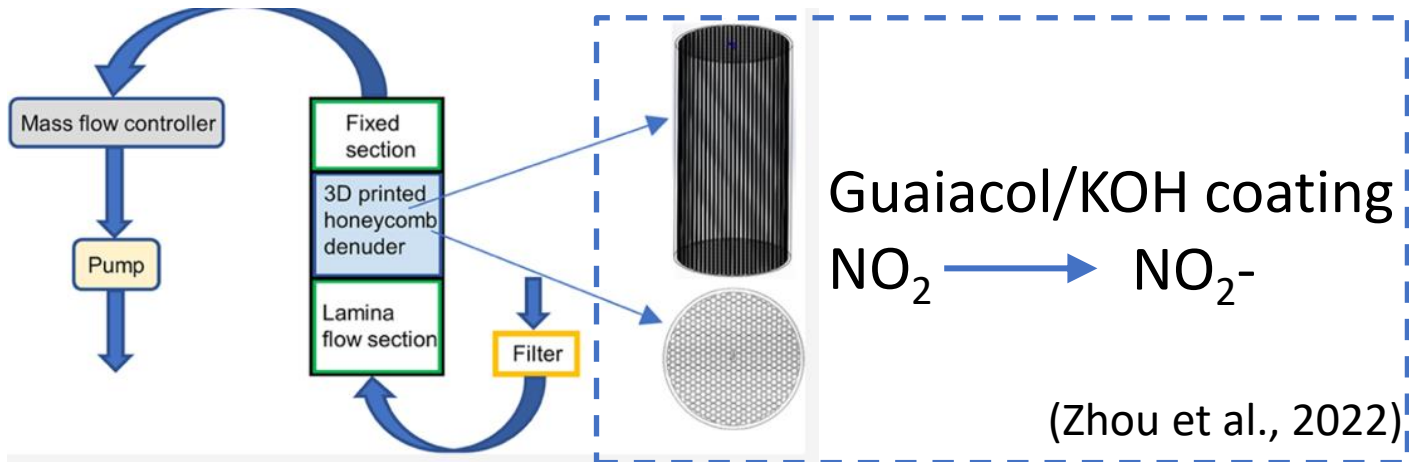
The definition of the oxygen mass independent signal (O-MIF):  $\Delta^{17}\text{O} = \delta^{17}\text{O} - 0.52 \times \delta^{18}\text{O}$

**$\Delta^{17}\text{O}$  of atmospheric NO<sub>2</sub> and HONO:**

- **The formation mechanisms of NO<sub>2</sub> and HONO;**
- **Oxidants concentration;**
- **Atmospheric oxidation capacity**

# The principle of the denuder sampling methods

The denuder sampling devices were used to collect atmospheric  $\text{NO}_2$  and HONO for isotope analysis



# The key bottleneck is the lack of reliable nitrite $\Delta^{17}\text{O}$ standards

Chemical coating convert  $\text{NO}_2/\text{HONO}$  to  $\text{NO}_2^-$



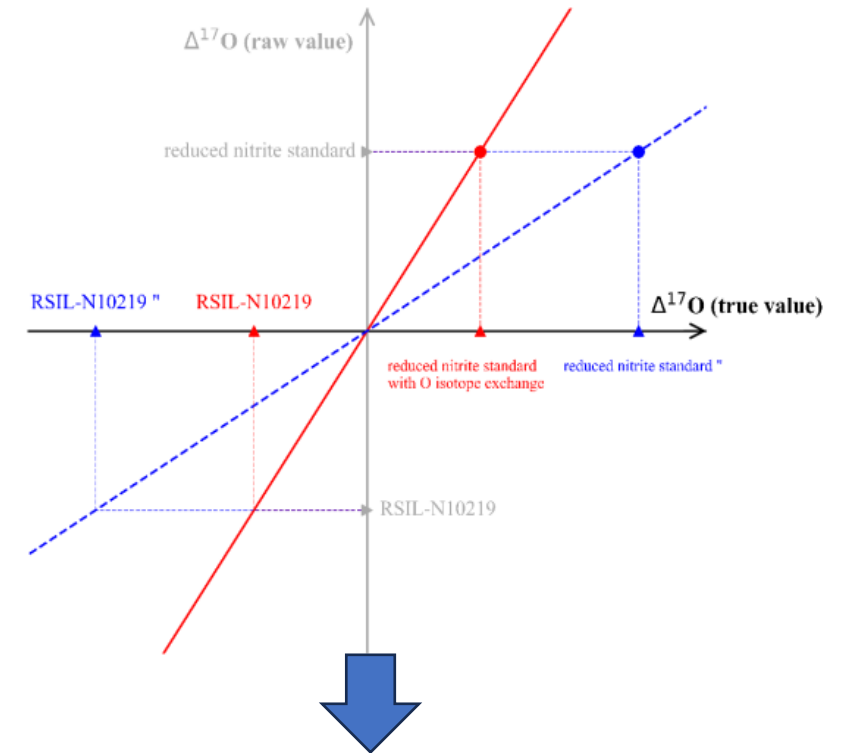
$\text{NO}_2^-$  sample and isotope standard identical treatment  
(using the azide method convert  $\text{NO}_2^-$  to  $\text{NO}_2$ )



Acquire the  $\Delta^{17}\text{O}$  value of  $\text{NO}_2^-$  sample based on the isotope calibration curve

## Three international nitrite isotope standards

Nitrite standard	$\delta^{15}\text{N}$ (‰)	$\delta^{18}\text{O}$ (‰)	$\Delta^{17}\text{O}$ (‰)
RSIL-N10219	2.8	88.5	non-zero
RSIL-N7373	-79.6	4.5	zero
RSIL-N23	3.7	11.4	zero



- The adopted true  $\Delta^{17}\text{O}$  value of RSIL-N10219 ranges from  $-7\text{‰} \sim -12.9\text{‰}$ , which will introduce significant uncertainty in calibration
- Atmospheric  $\text{NO}_2$  has positive  $\Delta^{17}\text{O}$  signals

# Objectives of this study

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- Prepare nitrite isotope standards with positive  $\Delta^{17}\text{O}$  signals
- Developed a new calibration approach suitable for accurate  $\Delta^{17}\text{O}$  signal determination of the RSIL-N10219 and the newly homemade nitrite standards



Precise evaluation of regional atmospheric  $\Delta^{17}\text{O}(\text{NO}_y)$  variation

# $^{17}\text{O}$ -enriched nitrite standards were prepared by oxygen exchange with water

## Step 1: Prepare $^{17}\text{O}$ -enriched Water

400  $\mu\text{L}$   $^{17}\text{O}$ -enriched mother solution + 1 L Mili-Q water ( $\Delta^{17}\text{O} = 76.5\text{‰}$ )

## Step 2: Isotope exchange

- Take 10 mL of prepared water dissolve 2 g ultrapure  $\text{NaNO}_2$
- Transfer to 100 mL tinted Schott bottle
  - Incubate at  $85^\circ\text{C}$  for 2 months

## Step 3: Drying and collection

Pour into 40 mL beaker || Dry at  $85^\circ\text{C}$  for 24 hours || Collect powder with scalpel || Homogenize with mortar || Dry at  $105^\circ\text{C}$  for 5 hours

## Step 4: Create standards

- Yield: 1.8 g pure nitrite
- Store in dark/dry conditions (RT)
  - Label:  **$\text{N-}\Delta^{17}\text{O-1}$**

Mix 0.1 g  $\text{N-}\Delta^{17}\text{O-1}$  with 0.1 g  $\text{NaNO}_2$  (1:1 mix)

Mix 0.1 g  $\text{N-}\Delta^{17}\text{O-1}$  with 0.9 g  $\text{NaNO}_2$  (1:9 mix)



**$\text{N-}\Delta^{17}\text{O-1}$**



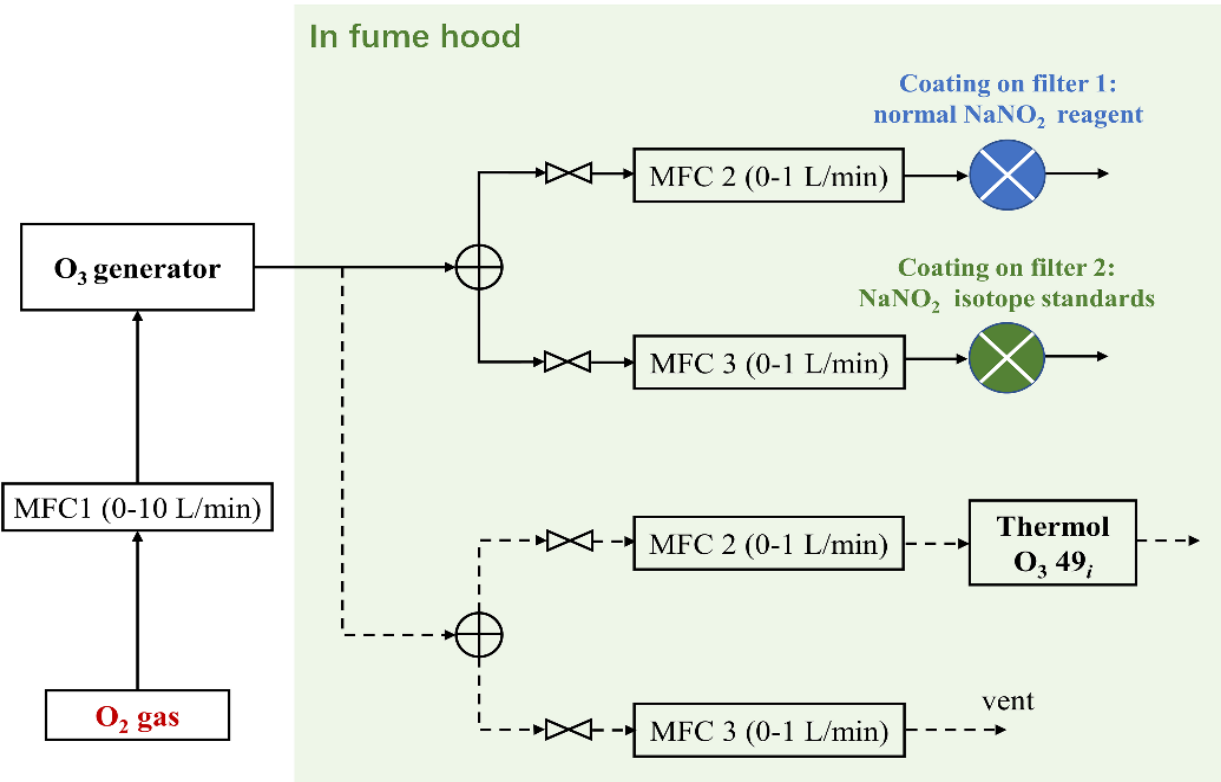
**$\text{N-}\Delta^{17}\text{O-2}$**



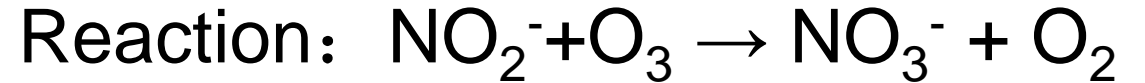
**$\text{N-}\Delta^{17}\text{O-3}$**

# Parallel O<sub>3</sub> oxidation method enables the calibration of nitrite Δ<sup>17</sup>O

## Schematic diagram of the O<sub>3</sub> oxidation experimental setup



solid line: ozone oxidation streams  
dotted line: O<sub>3</sub> concentration measurement



$$\Delta^{17}\text{O}(\text{NO}_3^-)_1 = \frac{2}{3} \times \Delta^{17}\text{O}(\text{NO}_2^-)_{\text{normal}} + \frac{1}{3} \times \Delta^{17}\text{O}(\text{O}_3^*)$$

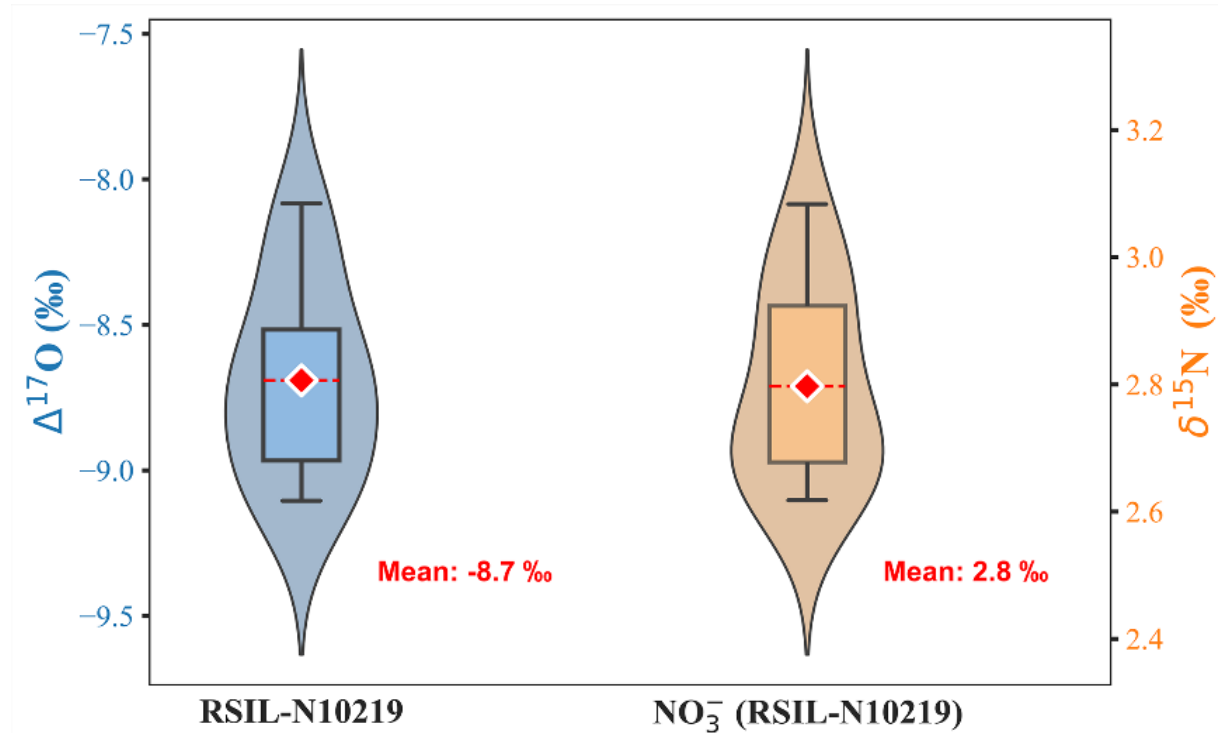
$$\Delta^{17}\text{O}(\text{NO}_3^-)_2 = \frac{2}{3} \times \Delta^{17}\text{O}(\text{NO}_2^-)_{\text{std}} + \frac{1}{3} \times \Delta^{17}\text{O}(\text{O}_3^*)$$

$$\Delta^{17}\text{O}(\text{NO}_2^-)_{\text{std}} = \frac{3}{2} \times (\Delta^{17}\text{O}(\text{NO}_3^-)_2 - \Delta^{17}\text{O}(\text{NO}_3^-)_1)$$

The  $\Delta^{17}\text{O}(\text{NO}_3^-)_1$  and  $\Delta^{17}\text{O}(\text{NO}_3^-)_2$  represent the  $\Delta^{17}\text{O}$  values of nitrate produced from the ozone oxidation of either normal nitrite reagent or nitrite standard, respectively .

# The $\Delta^{17}\text{O}$ calibration results of RSIL-N10219

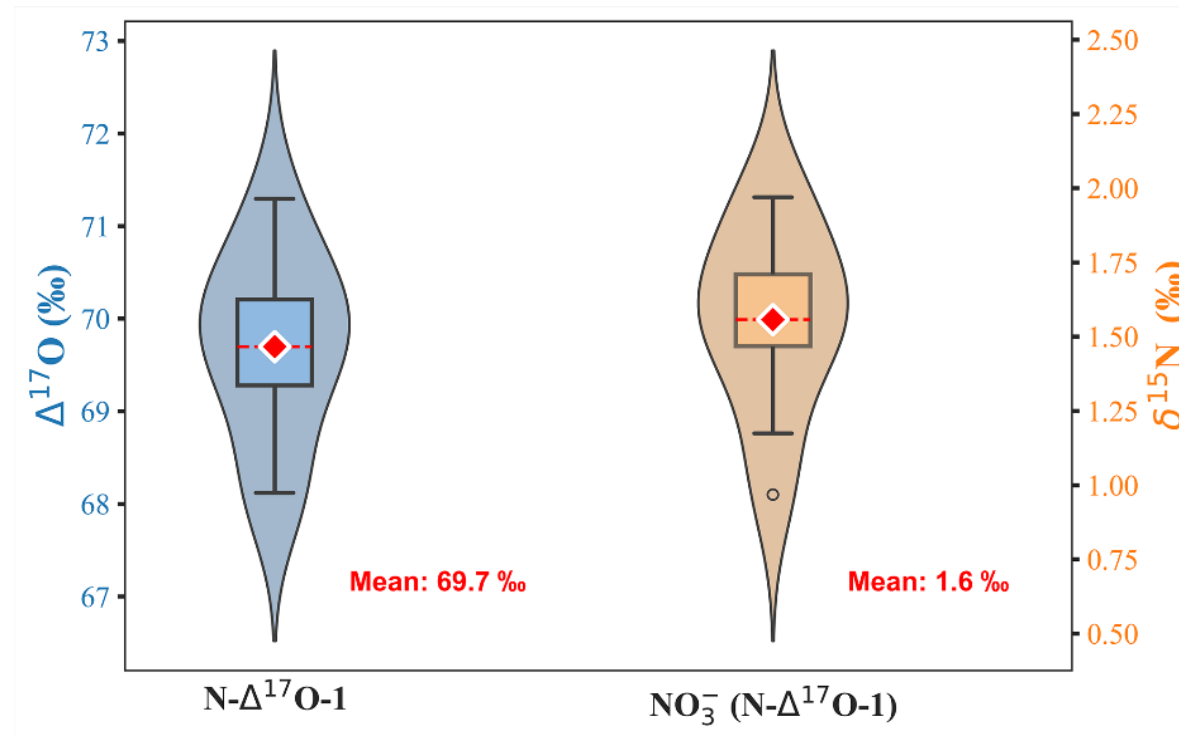
The  $\Delta^{17}\text{O}$  calibration results of RSIL-N10219 obtained from 11 different groups of  $\text{O}_3$  oxidation experiments



- The mean  $\Delta^{17}\text{O}$  value of RSIL-N12019 is  $(-8.7 \pm 0.3) \text{‰}$  ( $1\sigma$ ,  $n=11$ ).
- The measured  $\delta^{15}\text{N}$  value of  $\text{NO}_3^-$  produced by the  $\text{O}_3$  oxidation of RSIL-N10219 is  $(2.80 \pm 0.15) \text{‰}$  ( $\pm 1\sigma$ ,  $n=11$ ), in good agreement with the recommended  $\delta^{15}\text{N}$  value of RSIL-N10219 (2.80 ‰)
- The  $\Delta^{17}\text{O}$  of the generated ozone vary in a wide range from 26.1 ‰ to 32.5 ‰

# The $\Delta^{17}\text{O}$ calibration results of N- $\Delta^{17}\text{O}$ -1

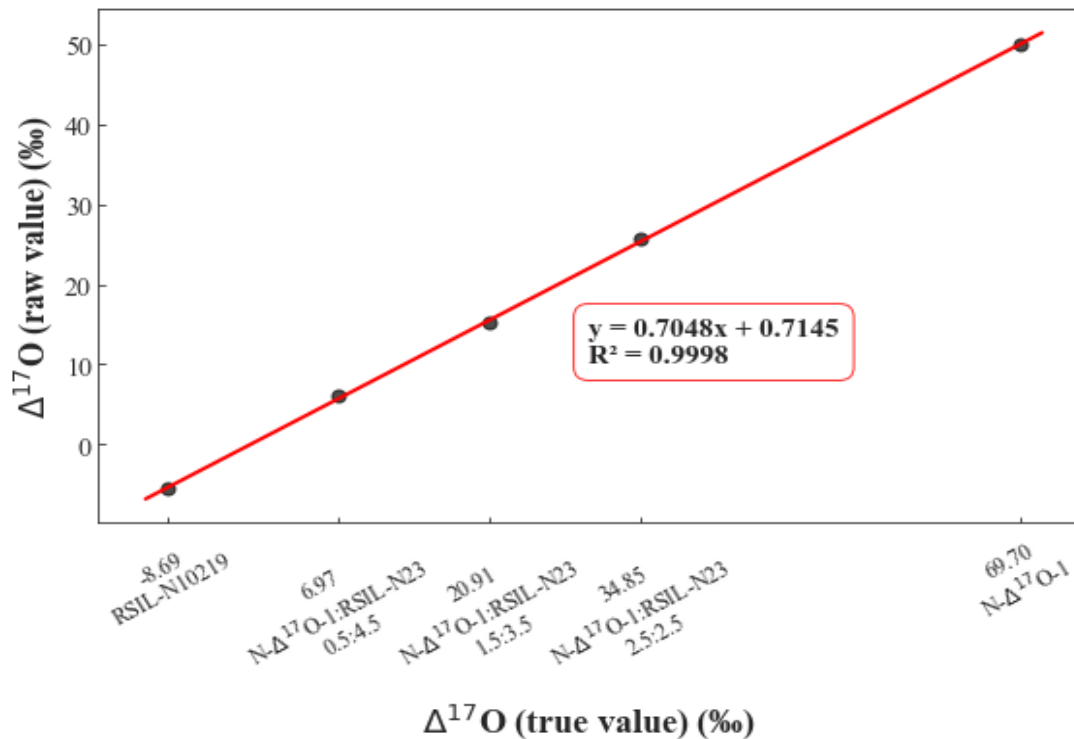
The  $\Delta^{17}\text{O}$  calibration results of N- $\Delta^{17}\text{O}$ -1 obtained from 10 different groups of  $\text{O}_3$  oxidation experiments



- The  $\Delta^{17}\text{O}$  of N- $\Delta^{17}\text{O}$ -1 was determined as  $69.7 \pm 1.0\text{‰}$  ( $n = 10$ ), showing good reproducibility.
- The measured value is lower than the  $\Delta^{17}\text{O}$  of  $^{17}\text{O}$ -enriched water ( $76.5\text{‰}$ ), suggesting incomplete isotope exchange between  $\text{NO}_2^-$  and water during the standard preparation process.

# The $\Delta^{17}\text{O}$ calibration results of N- $\Delta^{17}\text{O}$ -2 and N- $\Delta^{17}\text{O}$ -3

$\Delta^{17}\text{O}$  values of N- $\Delta^{17}\text{O}$ -2 and N- $\Delta^{17}\text{O}$ -3 were determined using RSIL-N10219 and N- $\Delta^{17}\text{O}$ -1 as references.



$\delta^{15}\text{N}$  and  $\delta^{18}\text{O}$  measurements of the homemade  $^{17}\text{O}$ -enriched nitrite samples

type	standard name	$\delta^{15}\text{N}$ (‰)	$\delta^{18}\text{O}$ (‰)	$\Delta^{17}\text{O}$ (‰)	n
reference	RSIL-N10219	2.8	88.5	-8.7	/
	RSIL-N7373	-79.6	4.5	0	/
	RSIL-N23	3.7	11.4	0	/
	RSIL-N1:1	-38.4	46.5	-4.35	/
sample	N- $\Delta^{17}\text{O}$ -1	$1.0 \pm 0.1$	$60.9 \pm 0.4$	$69.5 \pm 0.5$	10
	N- $\Delta^{17}\text{O}$ -2	$0.8 \pm 0.3$	$38.0 \pm 0.1$	$34.1 \pm 0.1$	6
	N- $\Delta^{17}\text{O}$ -3	$0.5 \pm 0.1$	$20.1 \pm 0.3$	$6.6 \pm 0.1$	8

- Calibration curves show excellent linearity ( $R^2 = 0.9998$ ), confirming method reliability.
- The  $\Delta^{17}\text{O}$  values are  $34.5 \pm 0.3\text{‰}$  (N- $\Delta^{17}\text{O}$ -2) and  $6.4 \pm 0.1\text{‰}$  (N- $\Delta^{17}\text{O}$ -3).
- These agree well with expected values from 1:1 and 1:9 mixing ratios ( $34.9\text{‰}$  and  $6.97\text{‰}$ ).

# Conclusions

- Reliable  $^{17}\text{O}$ -enriched nitrite standards were previously lacking.
- Here, we developed new nitrite standards with positive  $\Delta^{17}\text{O}$  values and established the  $\text{O}_3$  oxidation method for  $\Delta^{17}\text{O}$  signal calibration.
- This enables accurate  $\Delta^{17}\text{O}$  measurements of atmospheric  $\text{NO}_2$  and HONO, and provides a powerful tool to better constrain atmospheric oxidation processes.

*T Zhou, S Albertin, Z Jiang, J Savarino, J Duan, Z Yu, Y Yu, Z Zhang, Yichao Wu, Shihao Liu, Zeqian Liu, Slimane Bekki, Nicolas Caillon, Lei Geng\*, Preparation and Calibration of  $^{17}\text{O}$ -Enriched Nitrite Isotope Standards, ACS Earth and Space Chemistry, <https://doi.org/10.1021/acsearthspacechem.5c00399> (2026)*