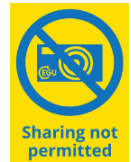


# Pollution dynamics in the lower Bengal Delta of Bangladesh



Image source: Sentinel-2 cloudless 2022 by EOX IT Services GmbH



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## Introduction

- The Bengal Delta, formed by the Ganges-Brahmaputra-Meghna river system, is located in the southwest coastal region of Bangladesh.
- It represents one of the most dynamic and ecologically sensitive deltaic environments in the world.
- Within this delta, tidal fluctuations under high and low tide conditions play a key role in altering river water chemistry, particularly through the mobilization of major ions such as  $\text{Na}^+$ ,  $\text{Cl}^-$ , and  $\text{SO}_4^{2-}$ .
- Compounding these natural processes, the delta is inhabited by approximately 170 million people who directly or indirectly depend on river and groundwater resources, making anthropogenic activities a significant driver of ion composition changes in coastal water.
- Furthermore, climate shifts observed over recent decades, including rising temperatures and increasingly erratic rainfall patterns, have further intensified ion compositional changes in coastal water chemistry.
- As a consequence of these interacting natural and anthropogenic pressures, the rivers and groundwater of the Bengal Delta face severe seawater intrusion and pollution arising from the complex interplay of terrestrial, mangrove, and marine processes.



## Study Area

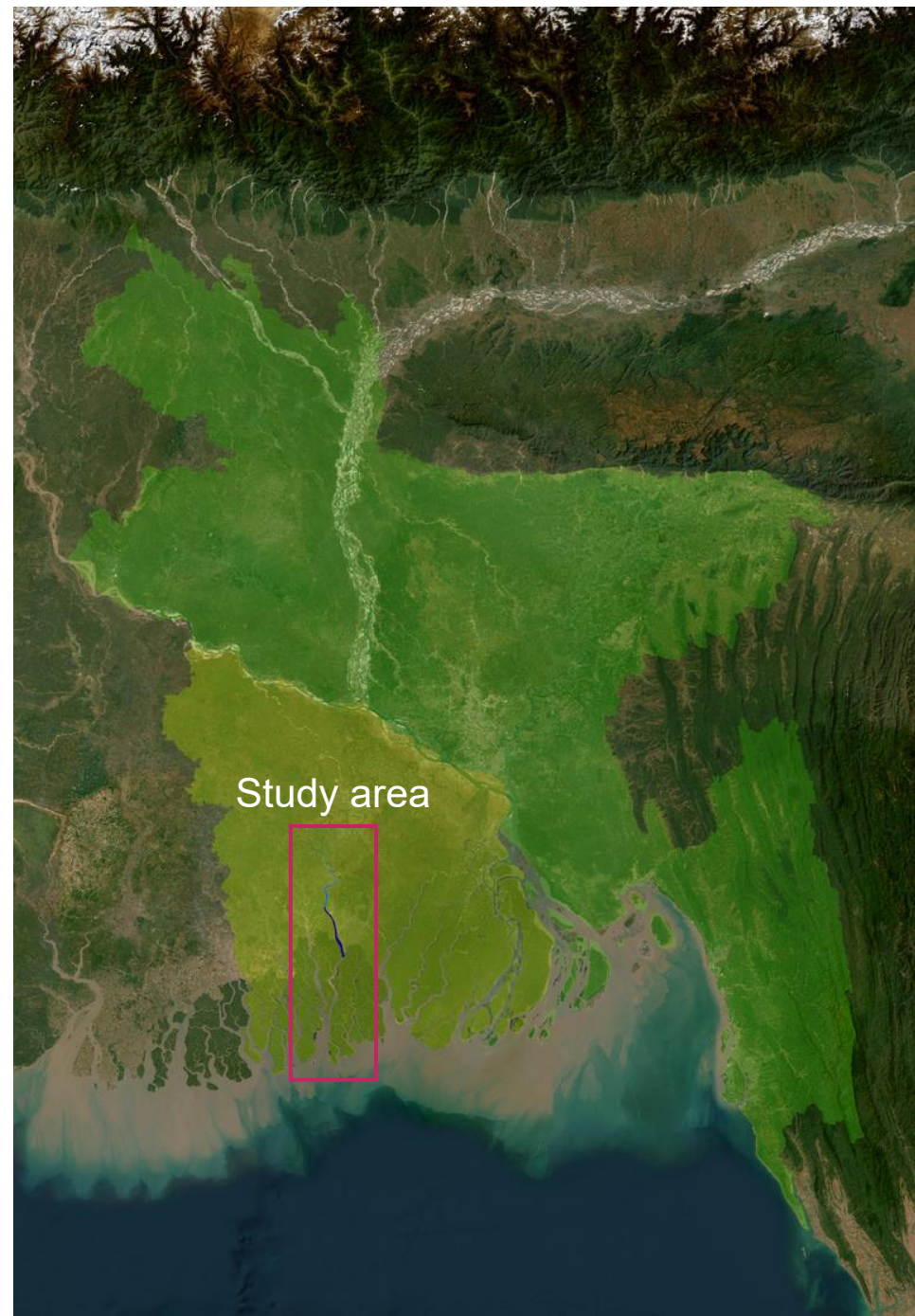


Fig.1 Study area

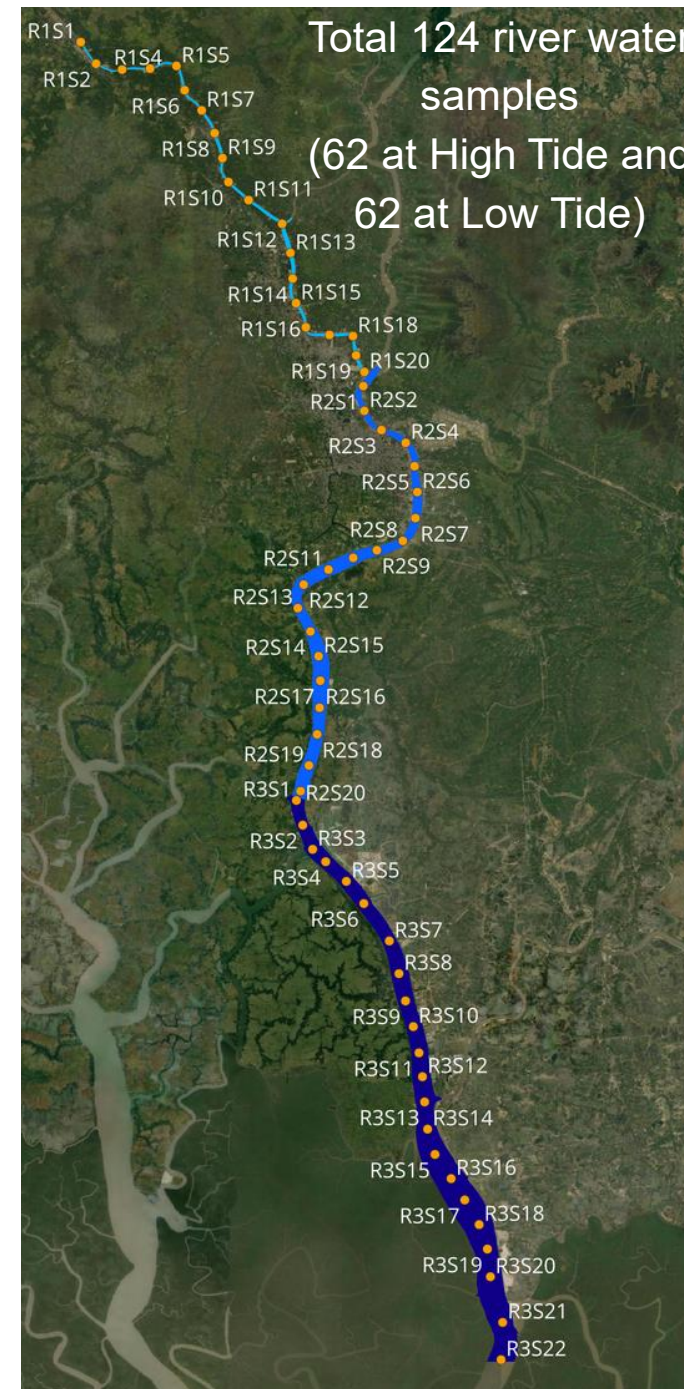
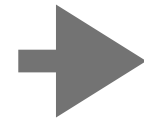


Fig. 1a river water (R1S1, R2S1, R3S1) sample locations

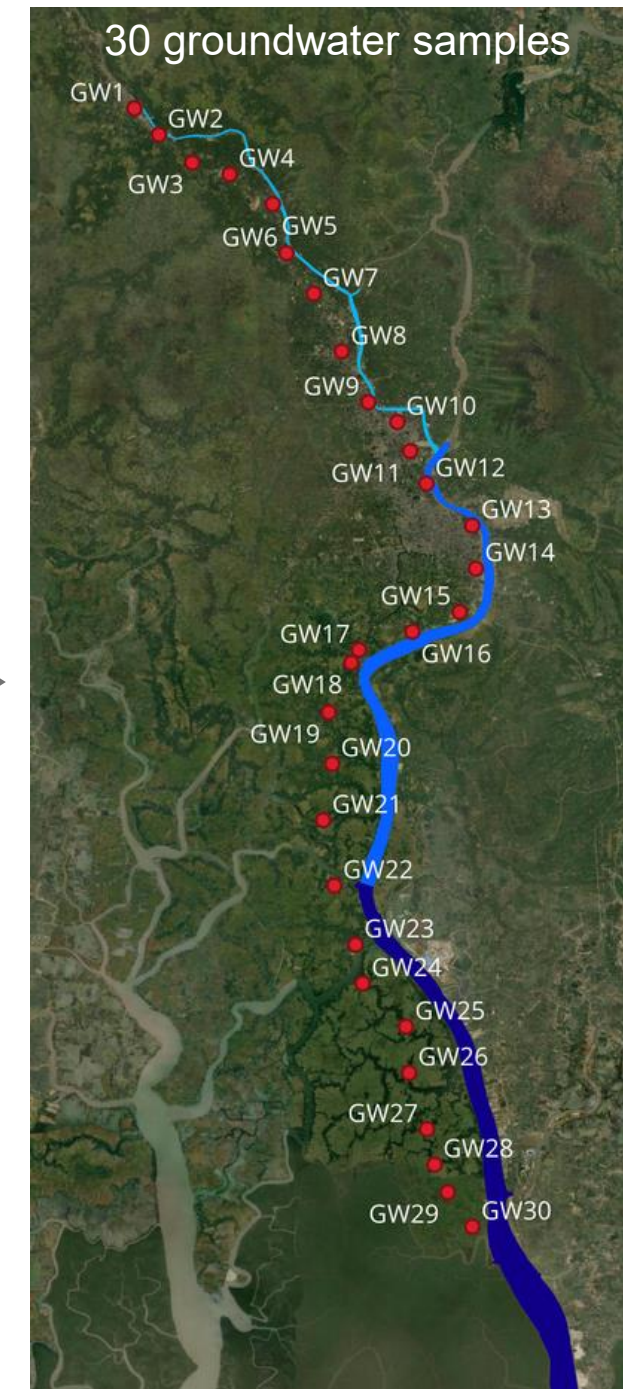
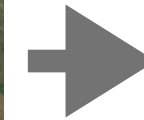


Fig. 1b groundwater (GW) sample locations

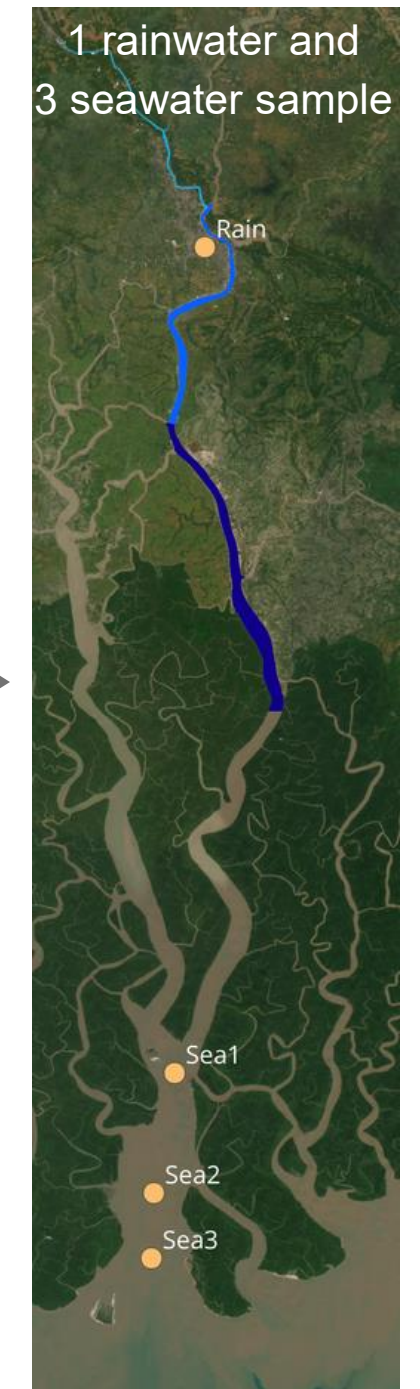
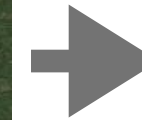


Fig. 1c rain and seawater sample locations

## Materials and Methods | Field Sampling

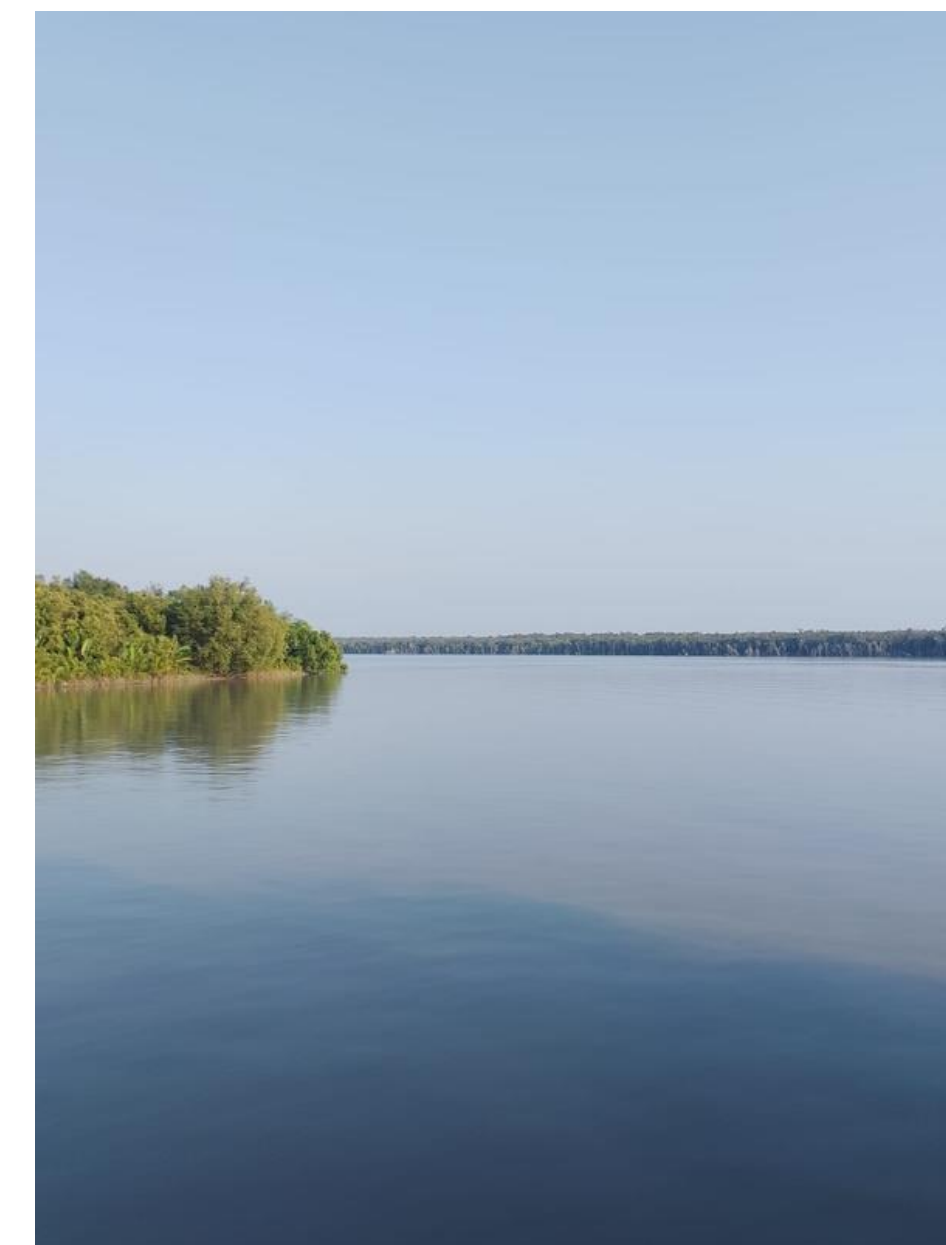


**Fig.2a** Sample collection and analysis in coastal rivers

**Analyzed parameters:** Water temperature, pH, Electrical Conductivity (EC), Total Dissolved Solids (TDS), Salinity, and Dissolved Oxygen (DO)



**Fig.2b** River and river-bank industry



**Fig.2c** River and Sundarbans mangrove forest



## Materials and Methods | Sampling Procedure

### Sampling technique

Composite Sampling

### Sampling period

Dry season, April to May 2025

### Sample collection

Total 158 water samples from (see location Fig.1)-

- Rivers
  - 3 coastal rivers
    - 62 water samples at high tide
    - 62 water samples at low tide
- Groundwater
  - 30 groundwater samples
- Sea
  - 3 seawater samples
- Rain
  - 1 rainwater sample

### In-situ (field) based water chemical analysis

- pH, water temperature, EC, TDS, Salinity, and DO

### Ex-situ (laboratory) based water chemical analysis

- BOD<sub>5</sub>, COD, and CaCO<sub>3</sub>
- major cations - Na<sup>+</sup>, K<sup>+</sup>, Ca<sup>2+</sup>, and Mg<sup>2+</sup>
- major anions - HCO<sub>3</sub><sup>-</sup>, Cl<sup>-</sup>, and SO<sub>4</sub><sup>2-</sup>
- selected nutrients parameters – o-PO<sub>4</sub><sup>3-</sup>, NO<sub>2</sub><sup>-</sup>-N, NO<sub>3</sub><sup>-</sup>-N, TP, and NH<sub>4</sub>-N
- selected heavy metals - Cr, Cu, Cd, Mn, Pb, Zn, and Fe and
- stable water isotopes - δ<sup>2</sup>H and δ<sup>18</sup>O

### Hydrochemical analysis

- Box-Whisker plot
- Scatter plot
- Piper and HFE-D diagram

### Isotopic analysis

- Stable water isotope analysis

### Geospatial analysis

- Land use/land cover map

## Materials and Methods | Pollution Status Assessment

Two equations were applied to assess the pollution status of three river catchments and groundwater quality status :

$$(\% ) \text{ of pollution index} = \frac{\text{Cl}^- + \text{SO}_4^{2-} + \text{NO}_3^- \text{-N}}{\text{Cl}^- + \text{SO}_4^{2-} + \text{NO}_3^- \text{-N} + \text{HCO}_3^-} \times 100$$

*Source: Pacheco and Van der Weijden, 1996*

$$\text{Water quality index (WQI)} = \frac{\sum_{i=1}^n W_i Q_i}{\sum_{i=1}^n W_i}$$

*Source: Horton, 1965; Brown et al., 1970; Cude, 2001, Tripaty and Sahu, 2005; Chowdhury et al., 2012*

Classification of water quality status:

0-25 = Excellent

26-50 = Good

51-75 = Poor

76-100 = Very poor

>100 = Unsuitable

*Source: Shweta et al., 2013; Chaterjee and Raziuddin, 2002.*



## Results | Data Analysis | River Water

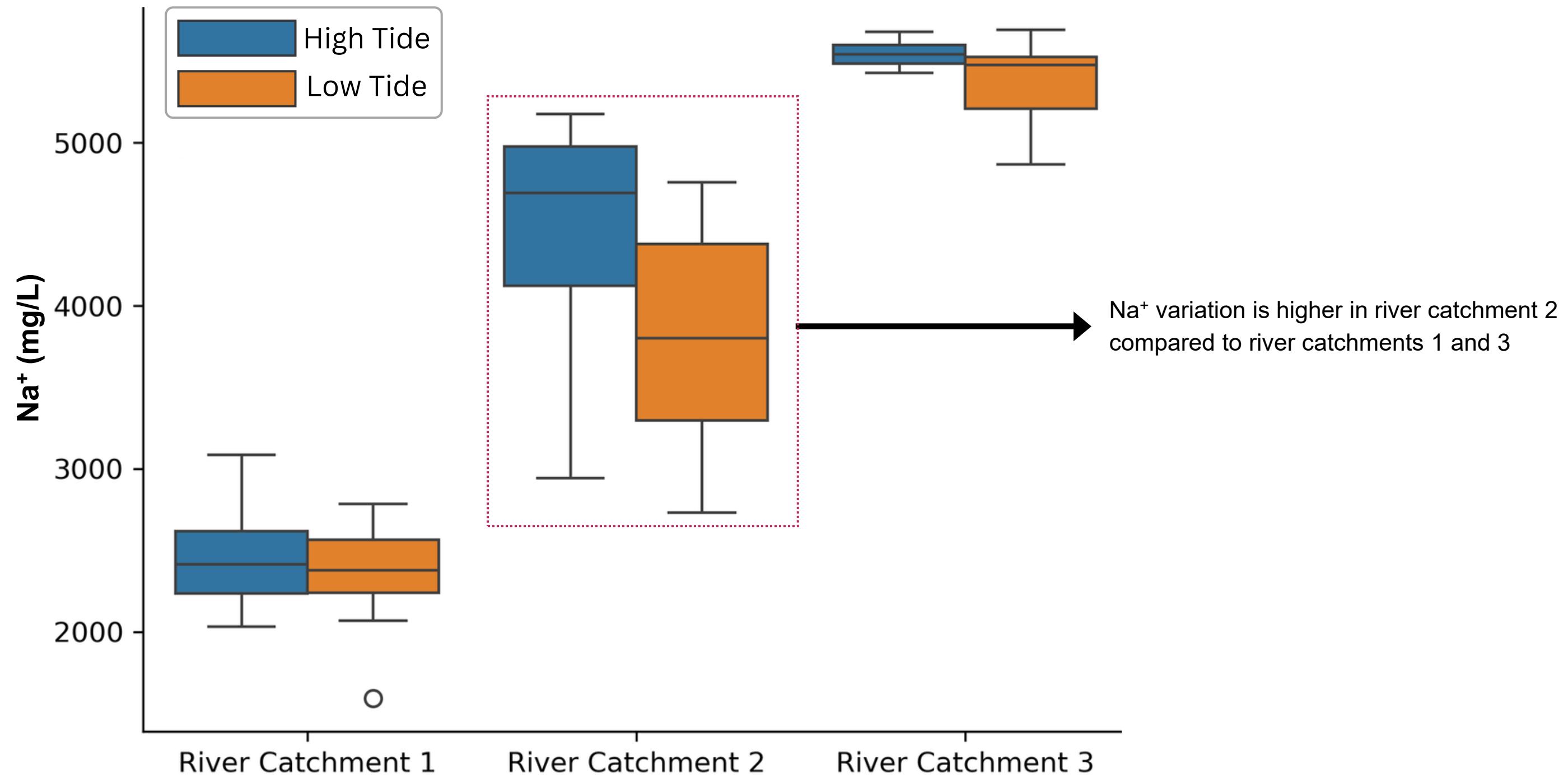


Fig. 3 Na<sup>+</sup>(mg/L) vs. three river catchments



Results | Data Analysis | River Water

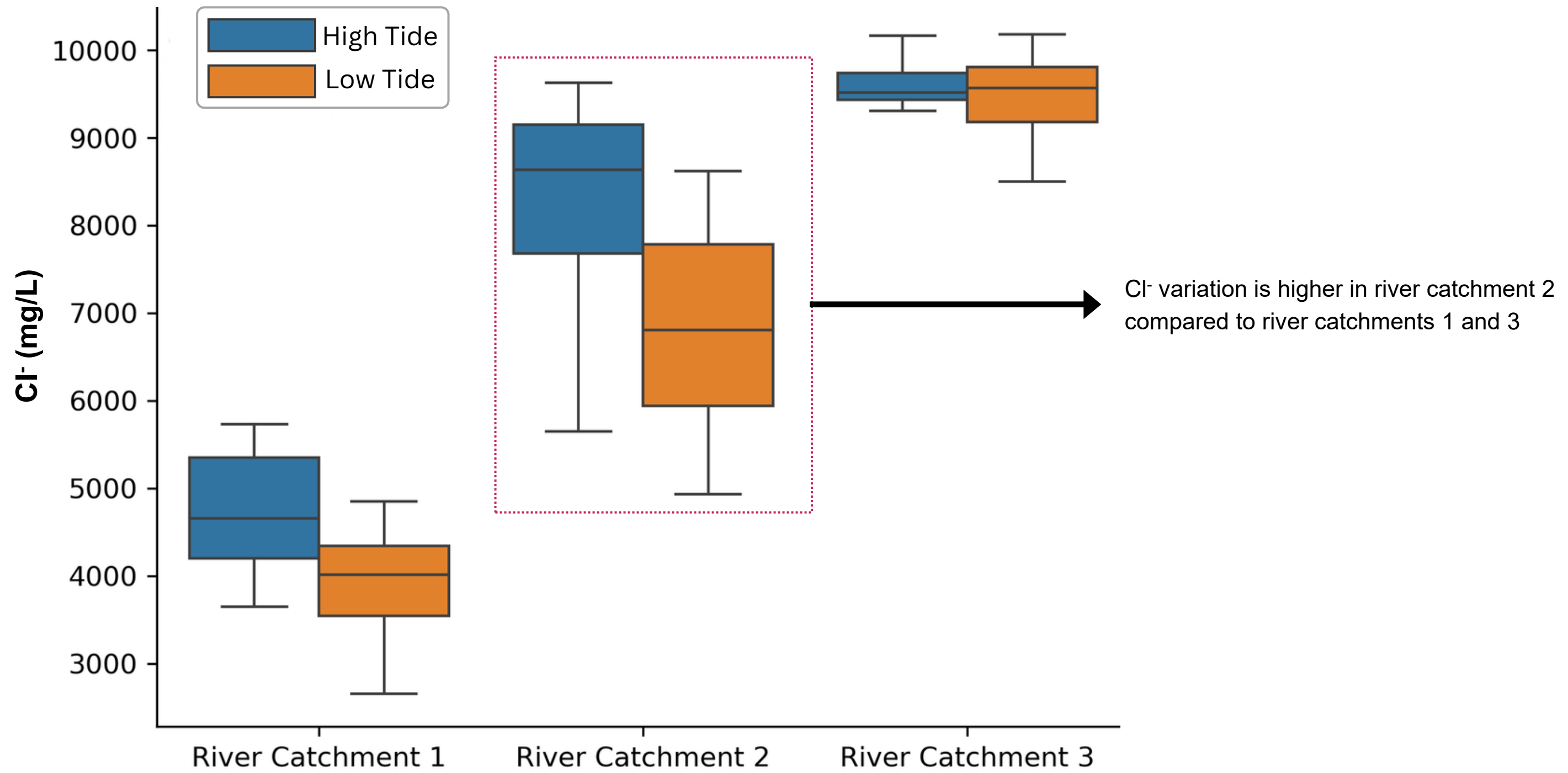


Fig. 4 Cl<sup>-</sup> (mg/L) vs. three river catchments



Results | Data Analysis | River Water

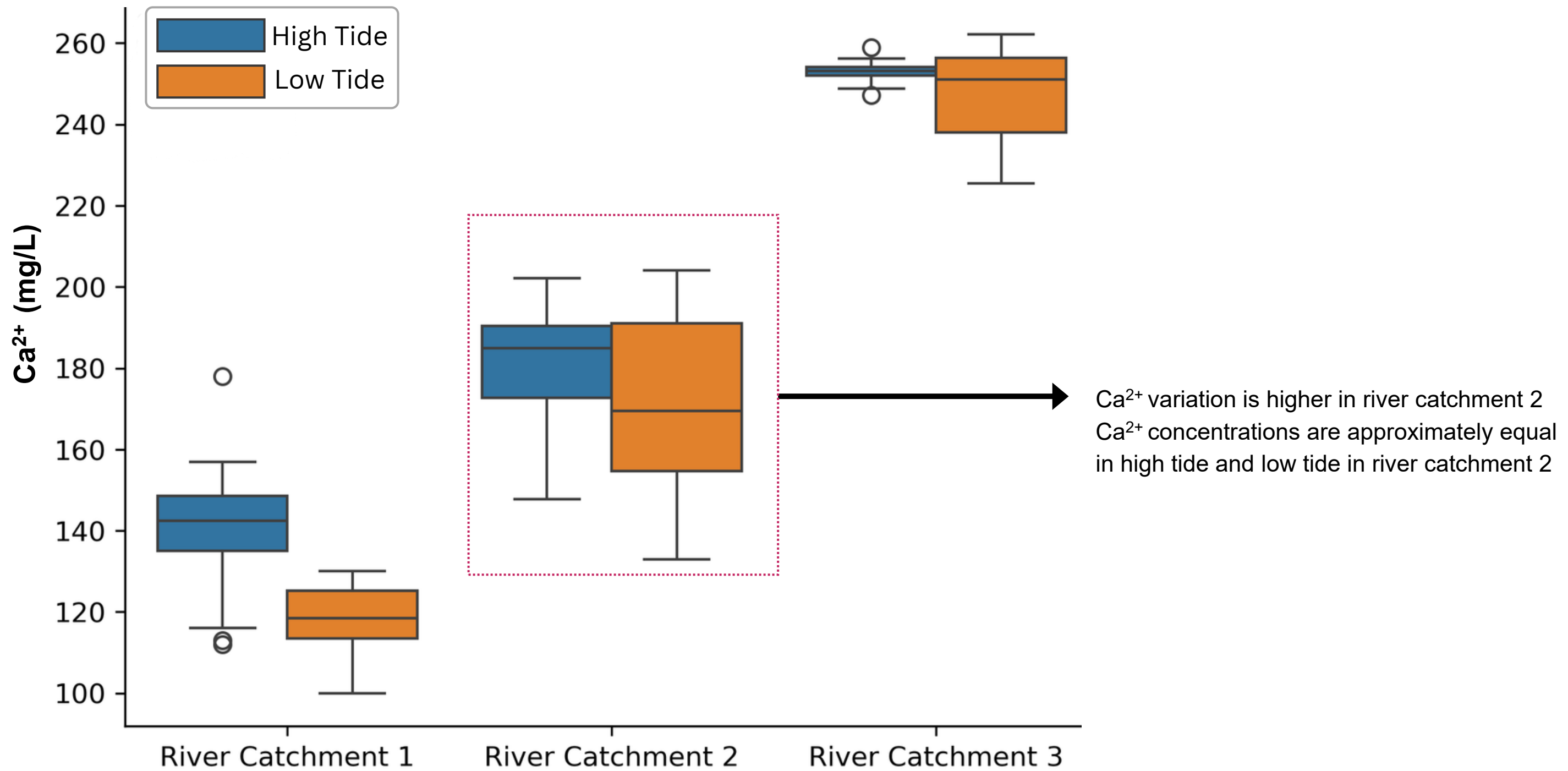


Fig. 5  $Ca^{2+}$  (mg/L) vs. three river catchments



Results | Data Analysis | River Water

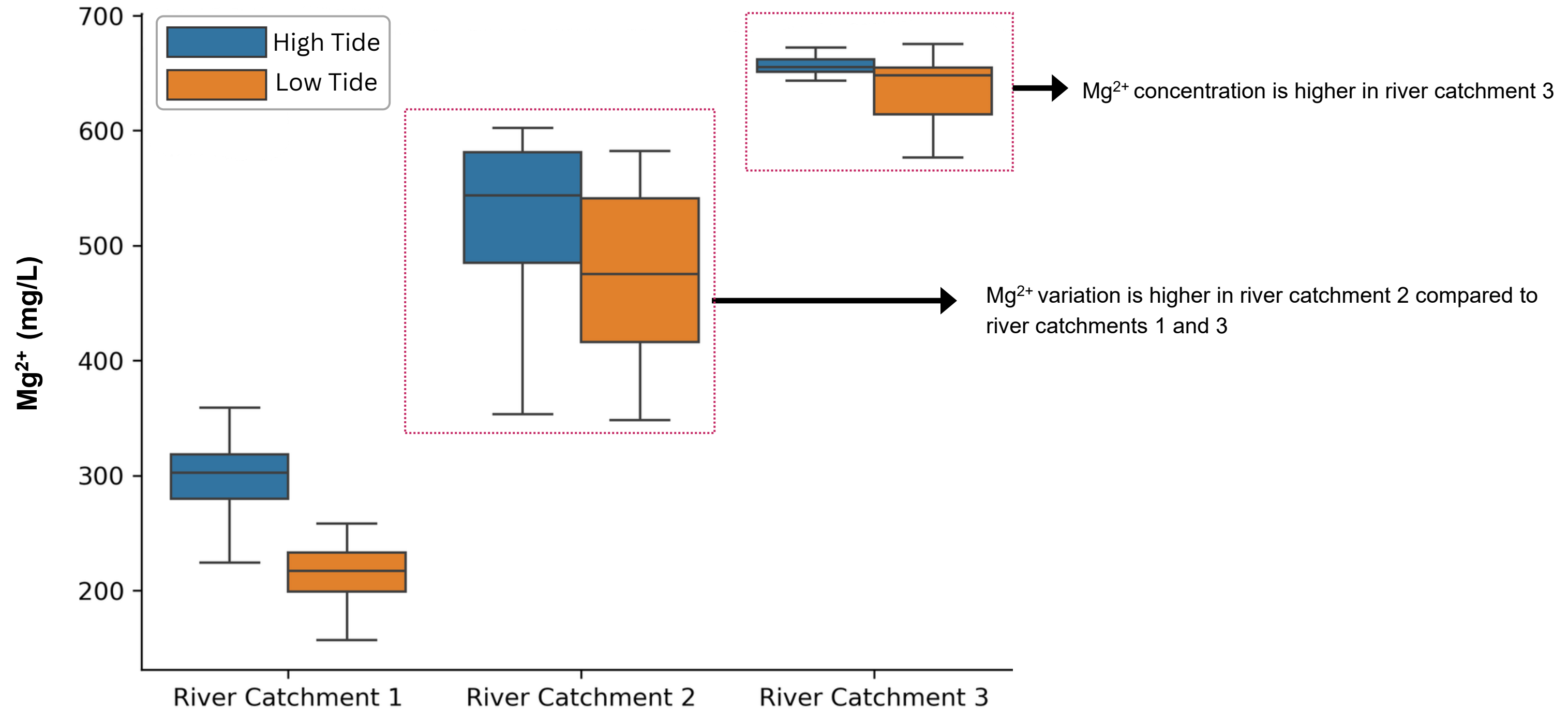


Fig. 6  $Mg^{2+}$  (mg/L) vs. three river catchments



## Results | Data Analysis | River Water

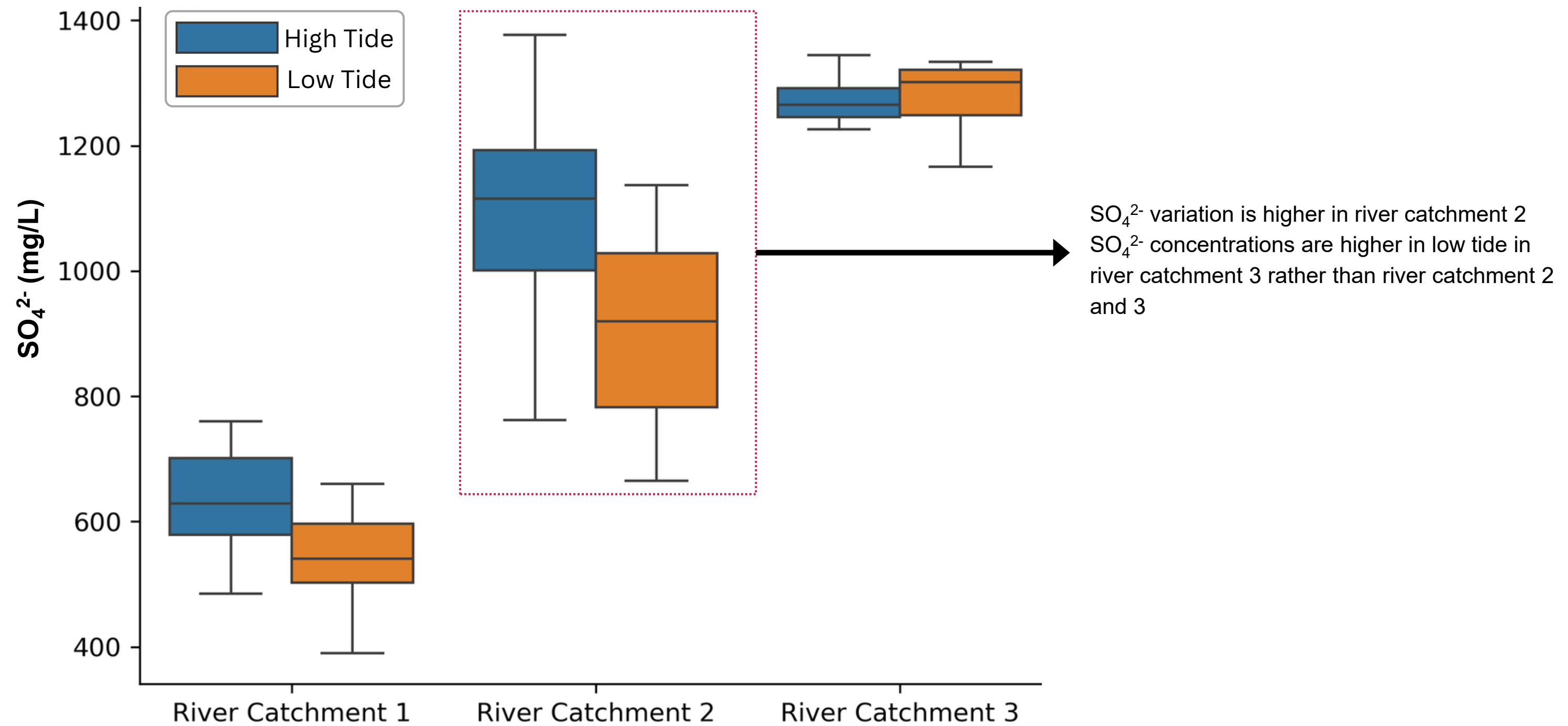


Fig. 7 SO<sub>4</sub><sup>2-</sup> (mg/L) vs. three river catchments



Results | Data Analysis | River Water

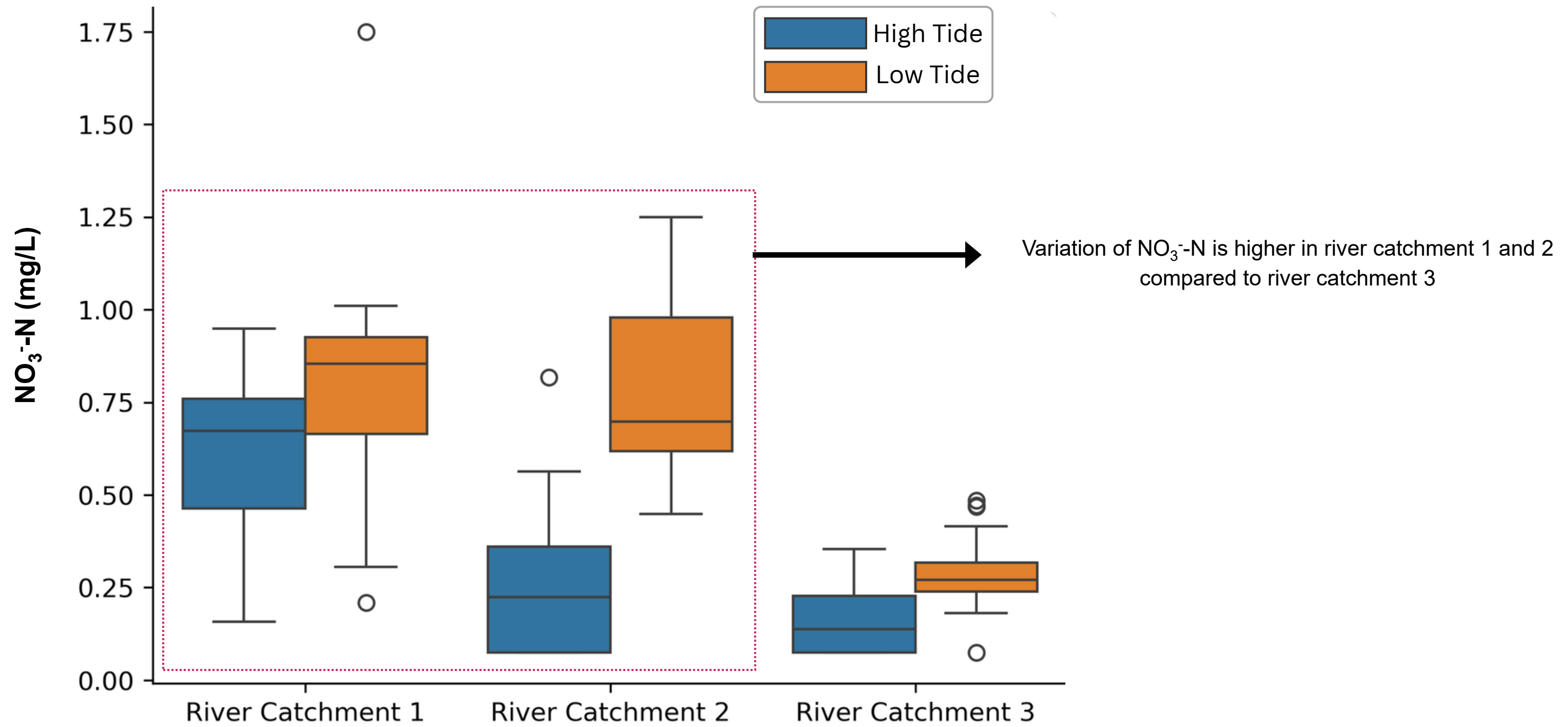
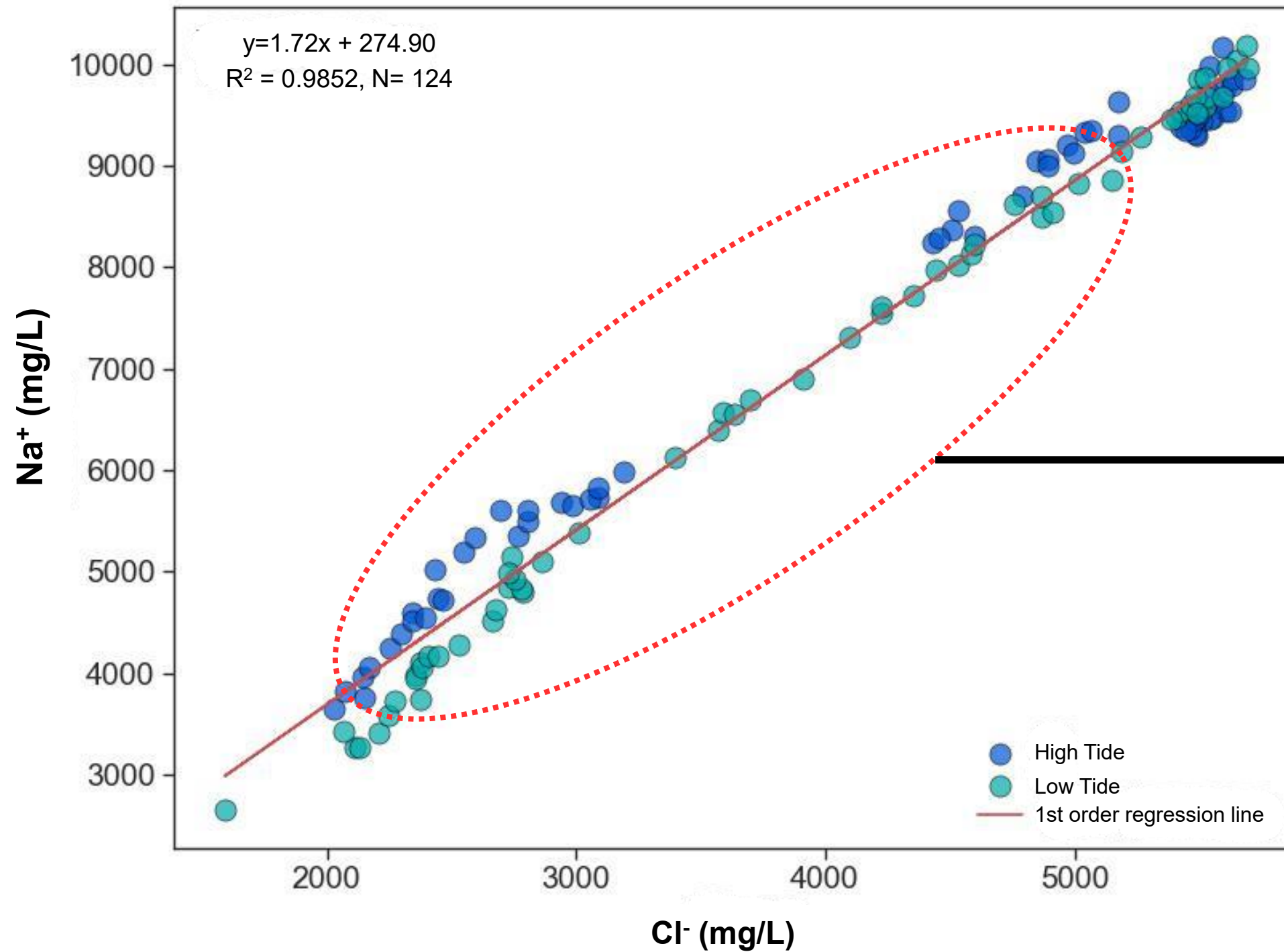


Fig. 8  $\text{NO}_3\text{-N}$  (mg/L) vs. three river catchments



## Results | Data Analysis | River Water

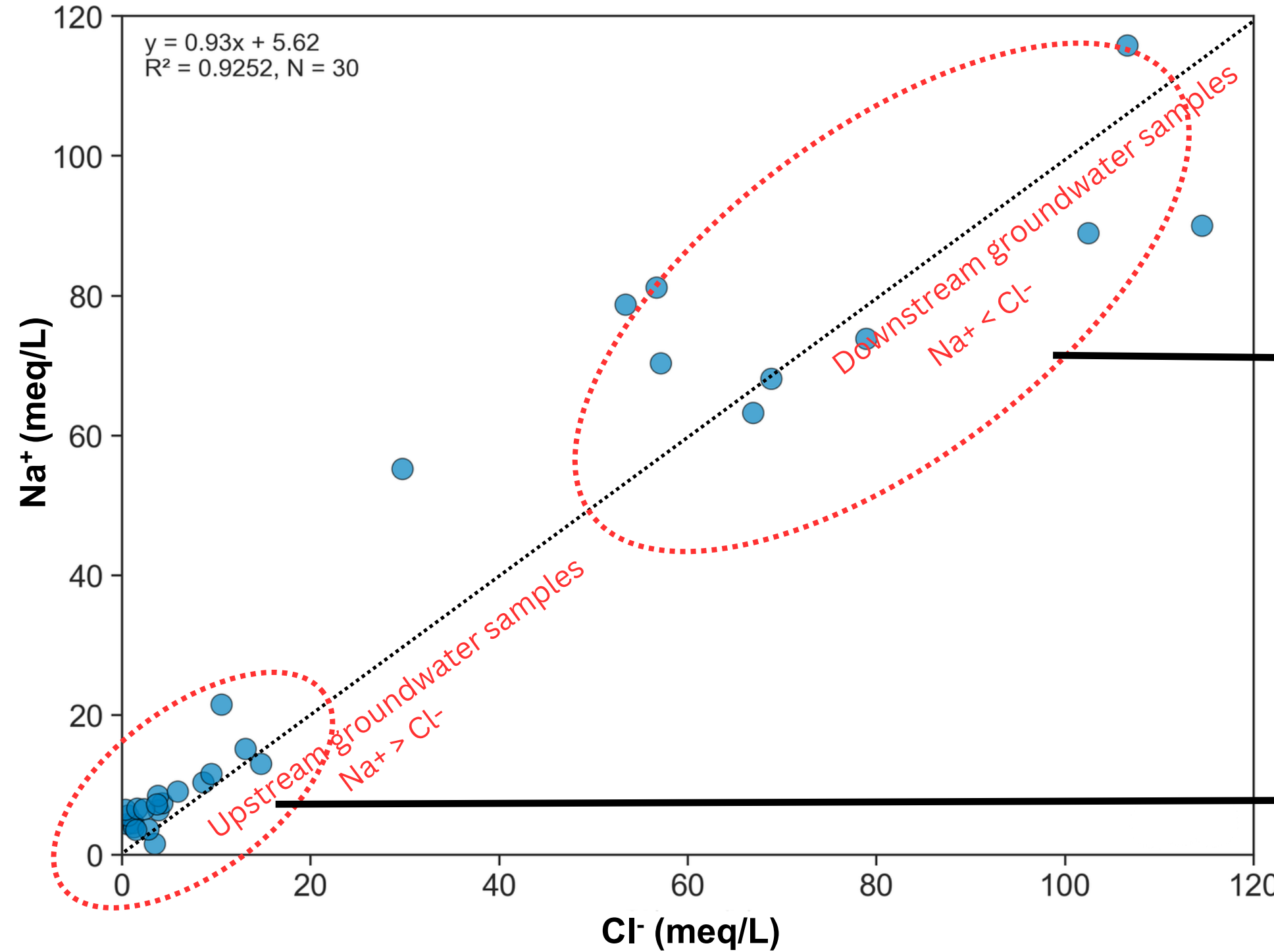


Na<sup>+</sup> and Cl<sup>-</sup> dominance in coastal river water

Fig. 9 Na<sup>+</sup>(mg/L) vs. Cl<sup>-</sup> (mg/L)



Results | Data Analysis | Groundwater



Na<sup>+</sup> < Cl<sup>-</sup> indicates reverse softening, seawater influence in downstream groundwater

Na<sup>+</sup> > Cl<sup>-</sup> indicates natural softening, dissolution of albite-plagioclase and ion exchange signature in upstream groundwater

Fig. 10 Na<sup>+</sup> (meq/L) vs. Cl<sup>-</sup> (meq/L)



Results | Data Analysis | Groundwater

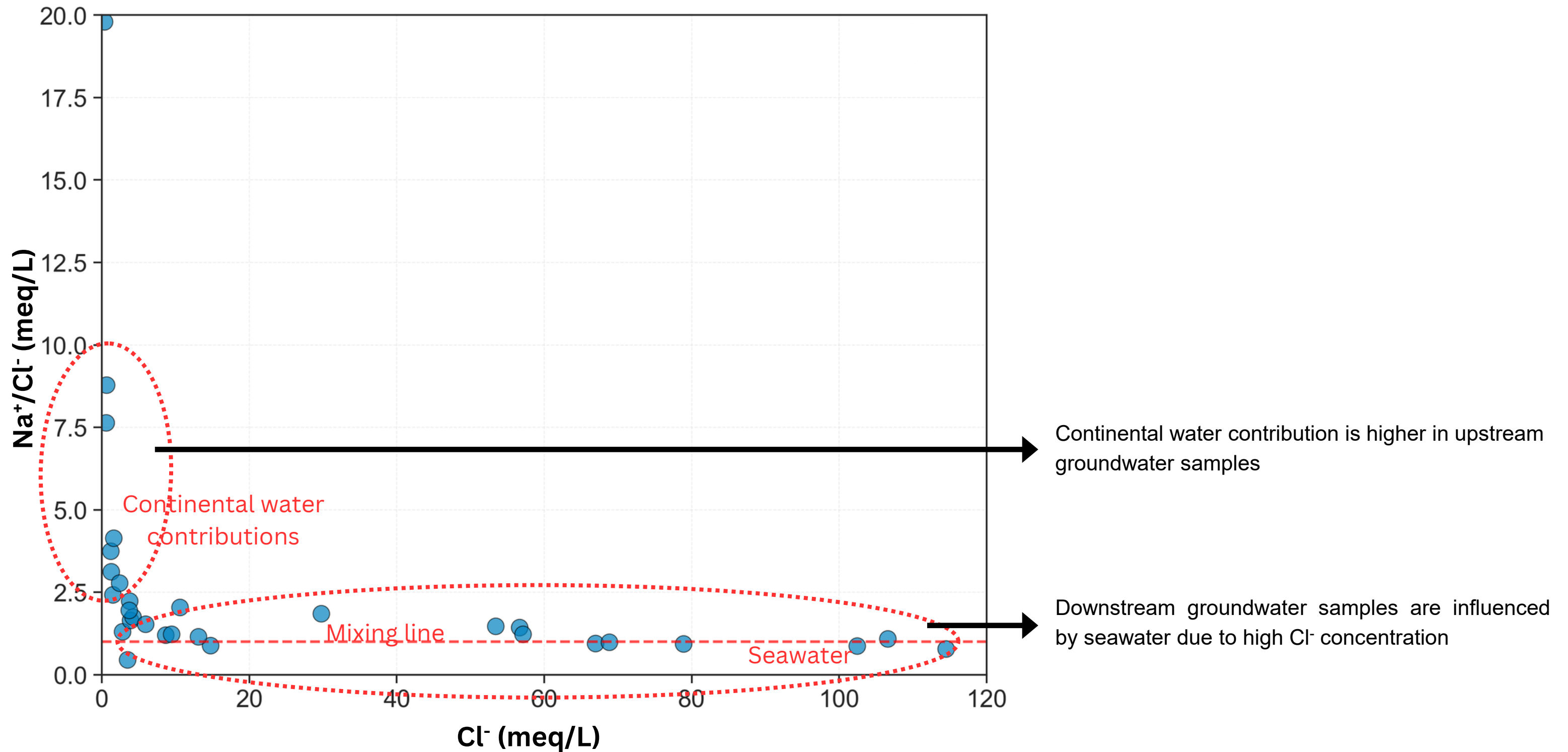


Fig. 11  $\text{Na}^+/\text{Cl}^-$  (meq/L) vs.  $\text{Cl}^-$  (meq/L)



Results | Data Analysis | River and Groundwater

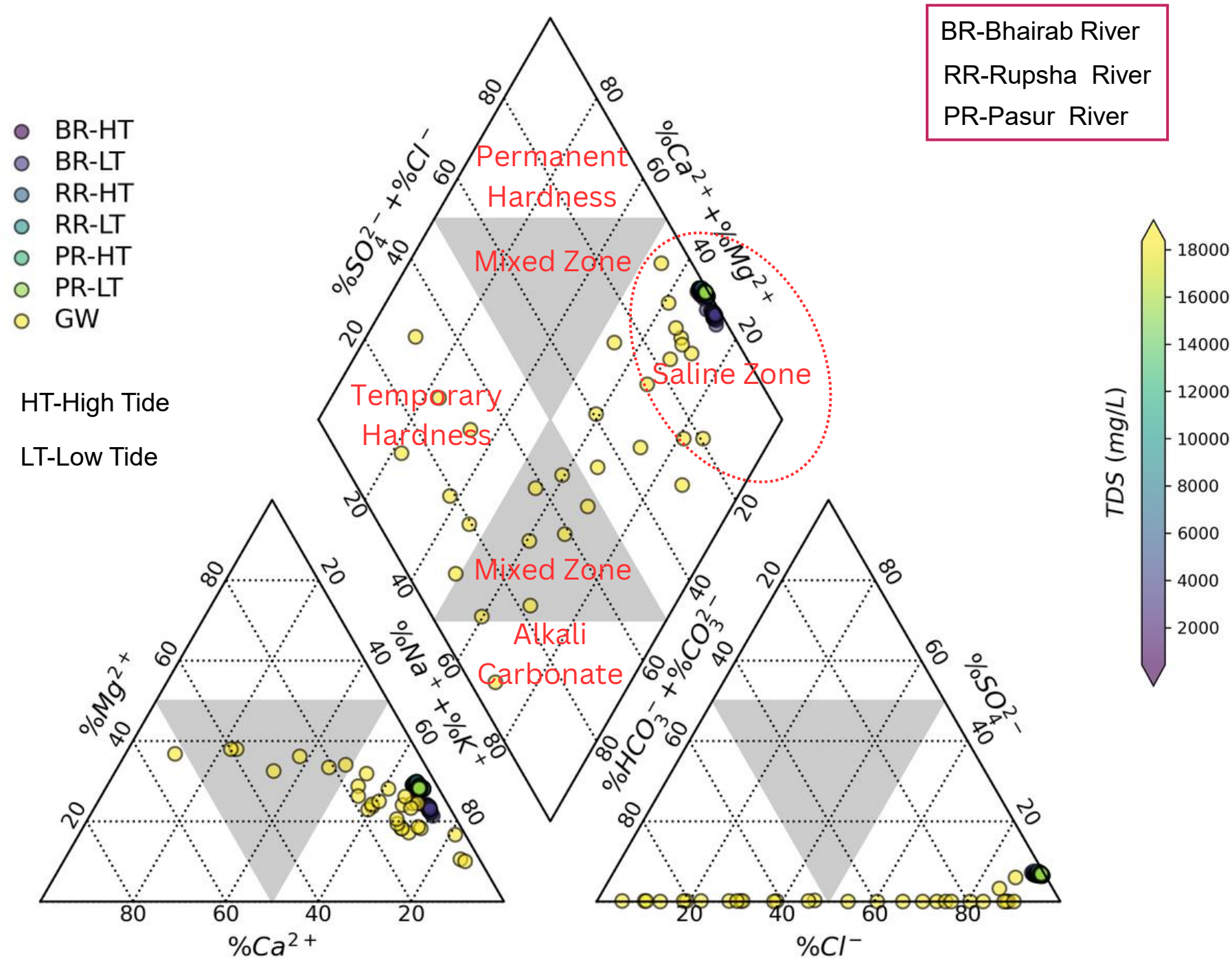


Fig.12a Piper diagram

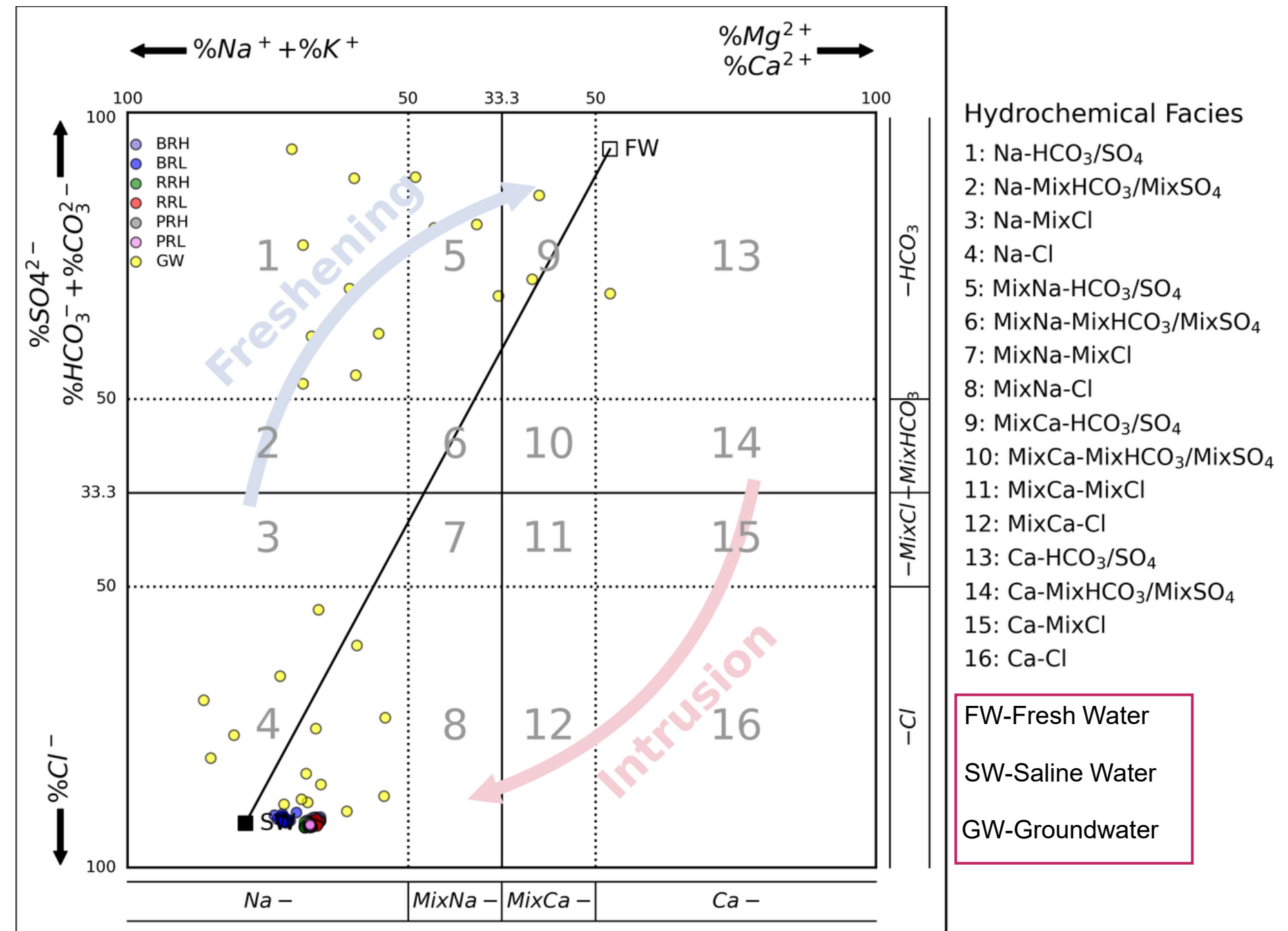


Fig.12b HFE-D diagram

Results | Data Analysis | River and Groundwater

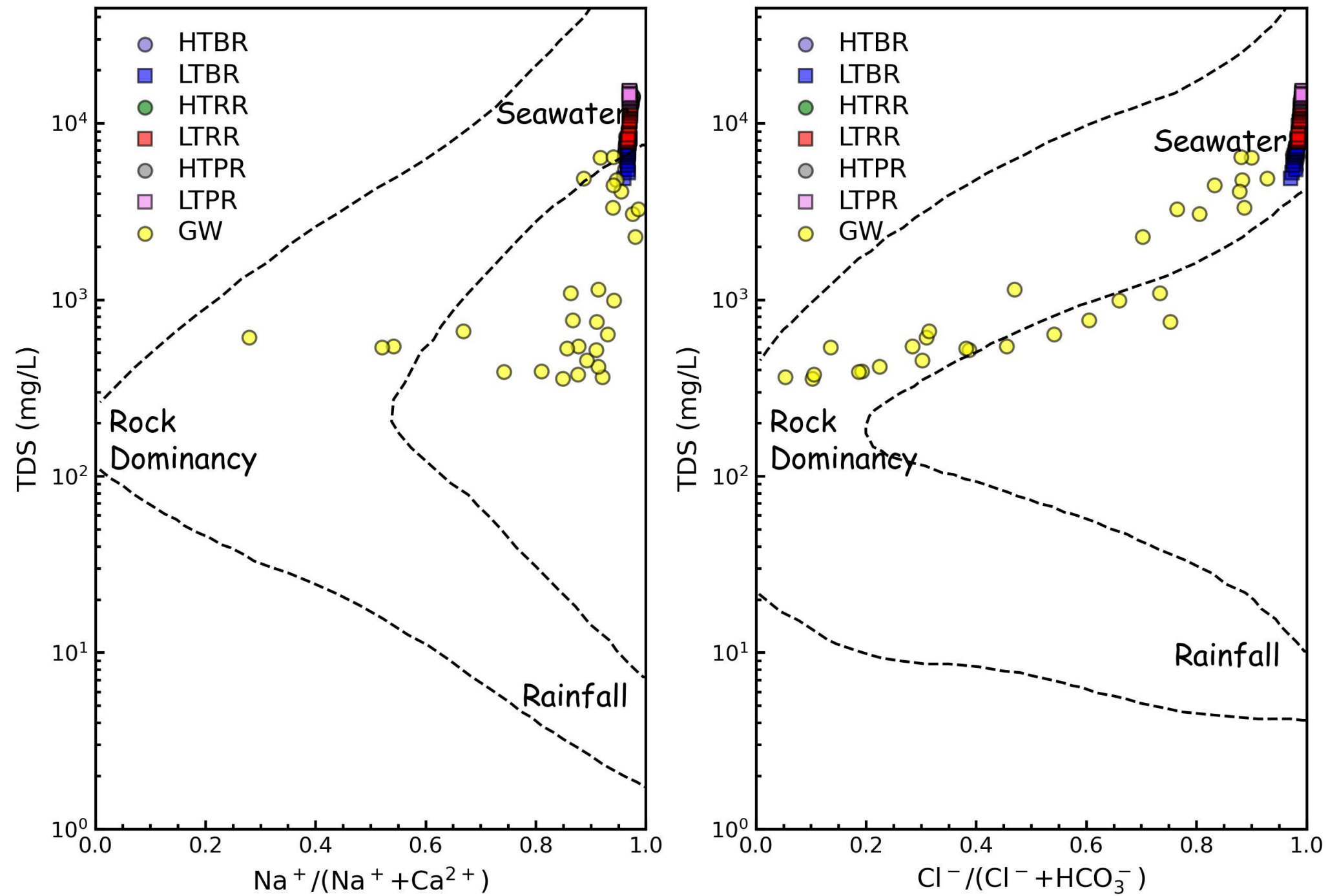
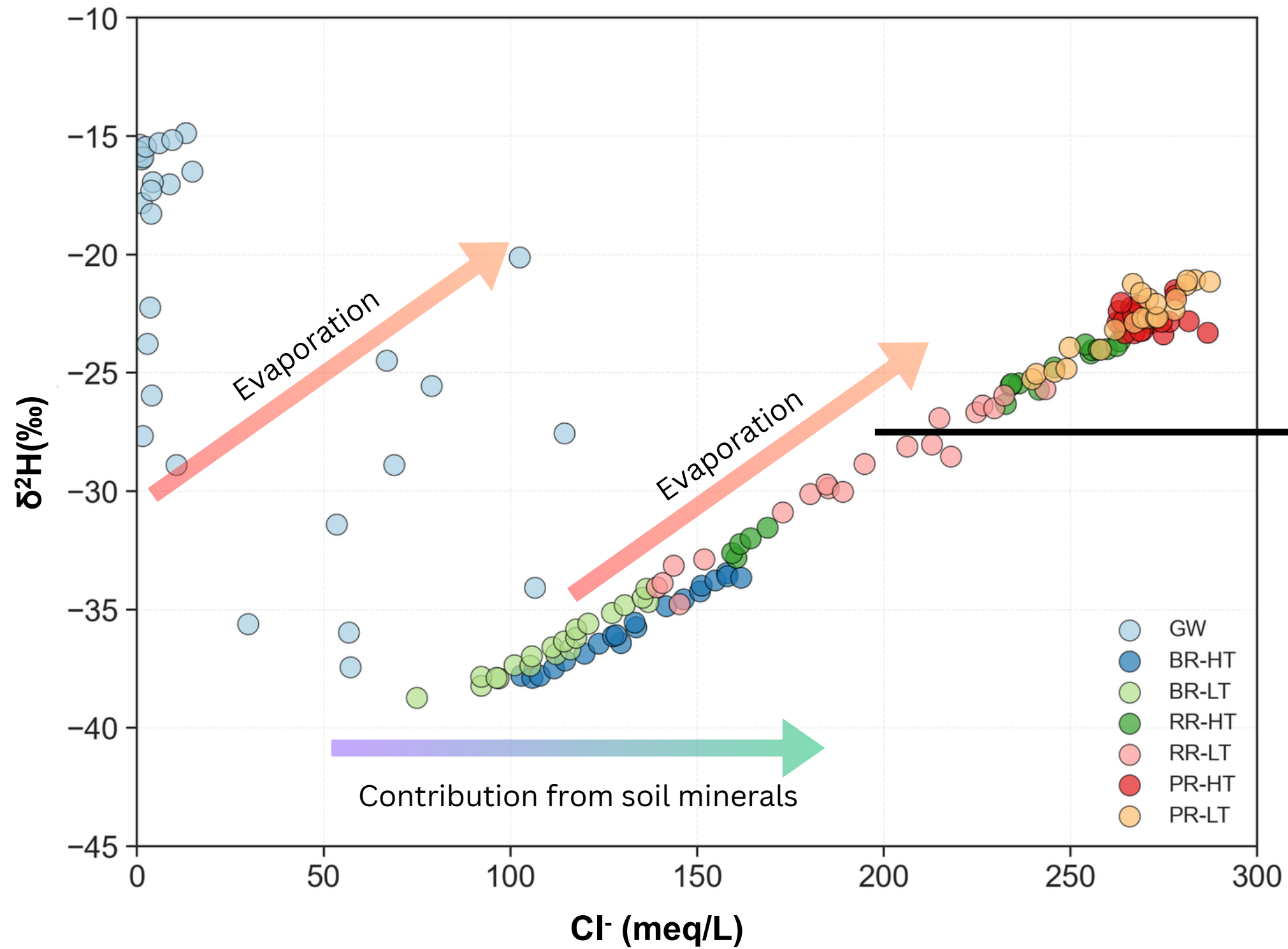


Fig.13 Gibbs diagram



Results | Data Analysis | River and Groundwater

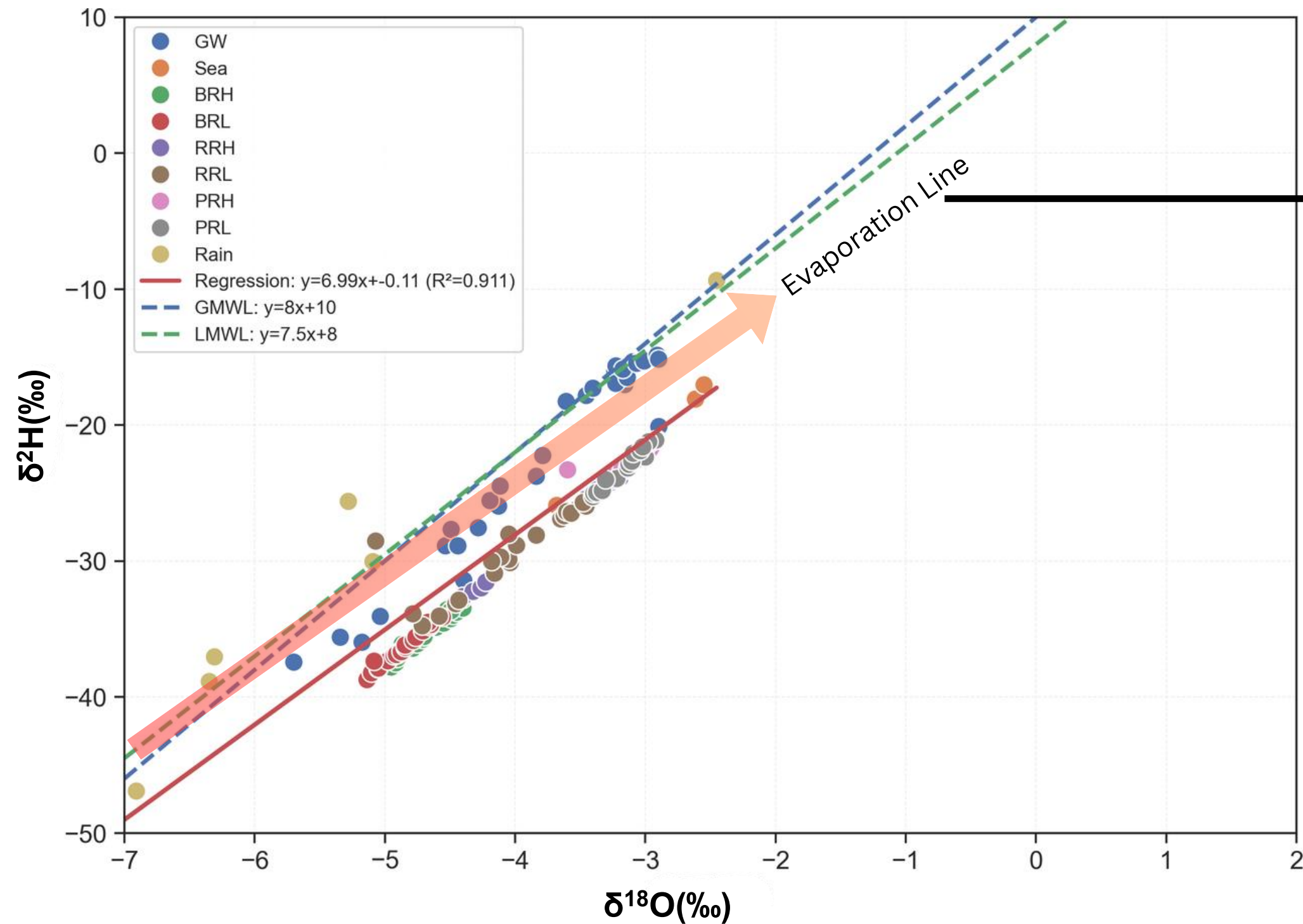


Increasing water salinity in hydrogeological processes indicates evaporation influence in water

Fig. 14 Hydrogeological processes of water salinity



Results | Data Analysis | River and Groundwater

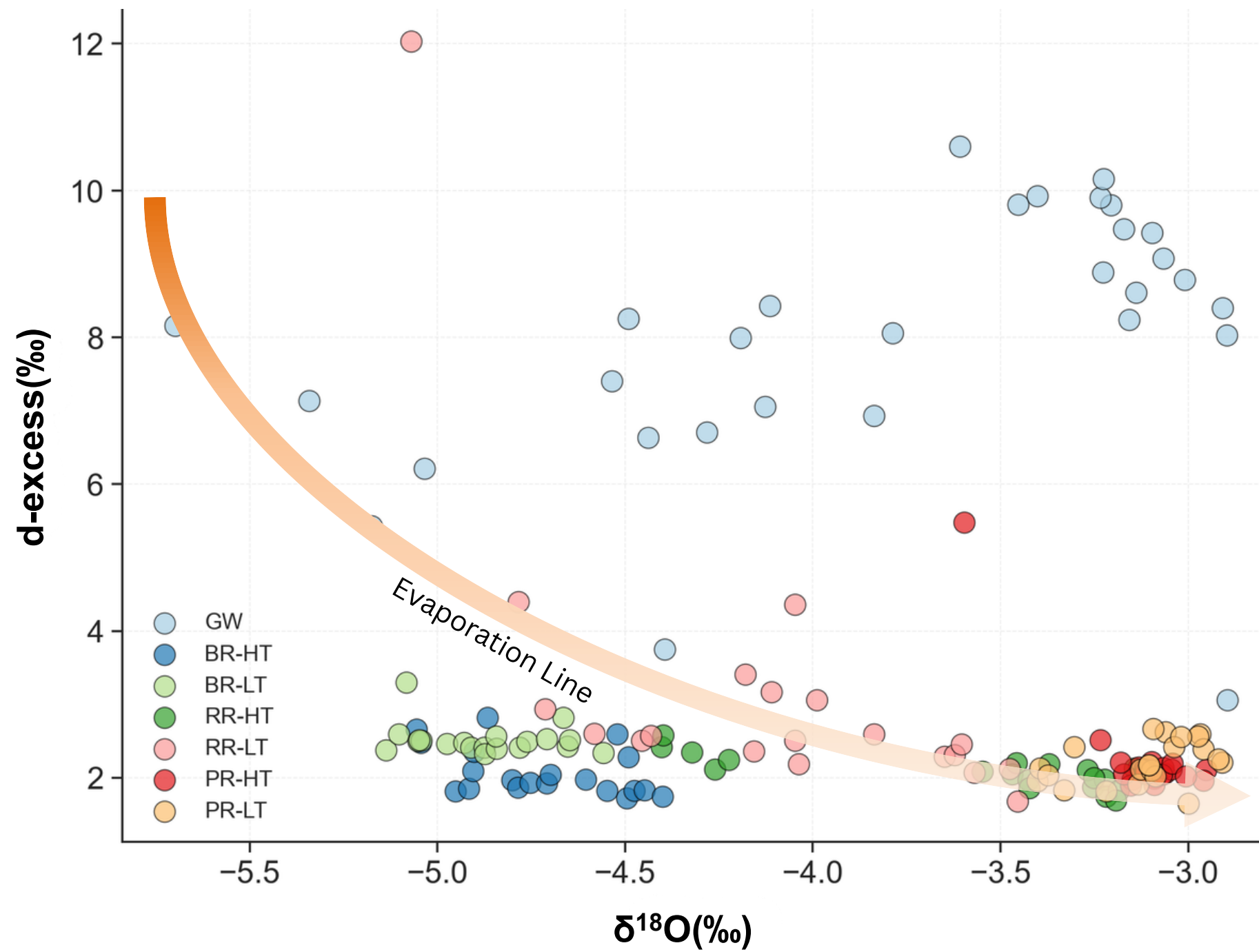


Evaporation influence in river and groundwater

Fig. 15 Evaporation trends of water salinity



## Results | Data Analysis | River and Groundwater



Lower deuterium (d) excess values indicate higher oxygen and evaporation influence in water

Fig. 16 Evaporation trends of water salinity

## Results | Data Analysis | Pollution Status | All River Catchments

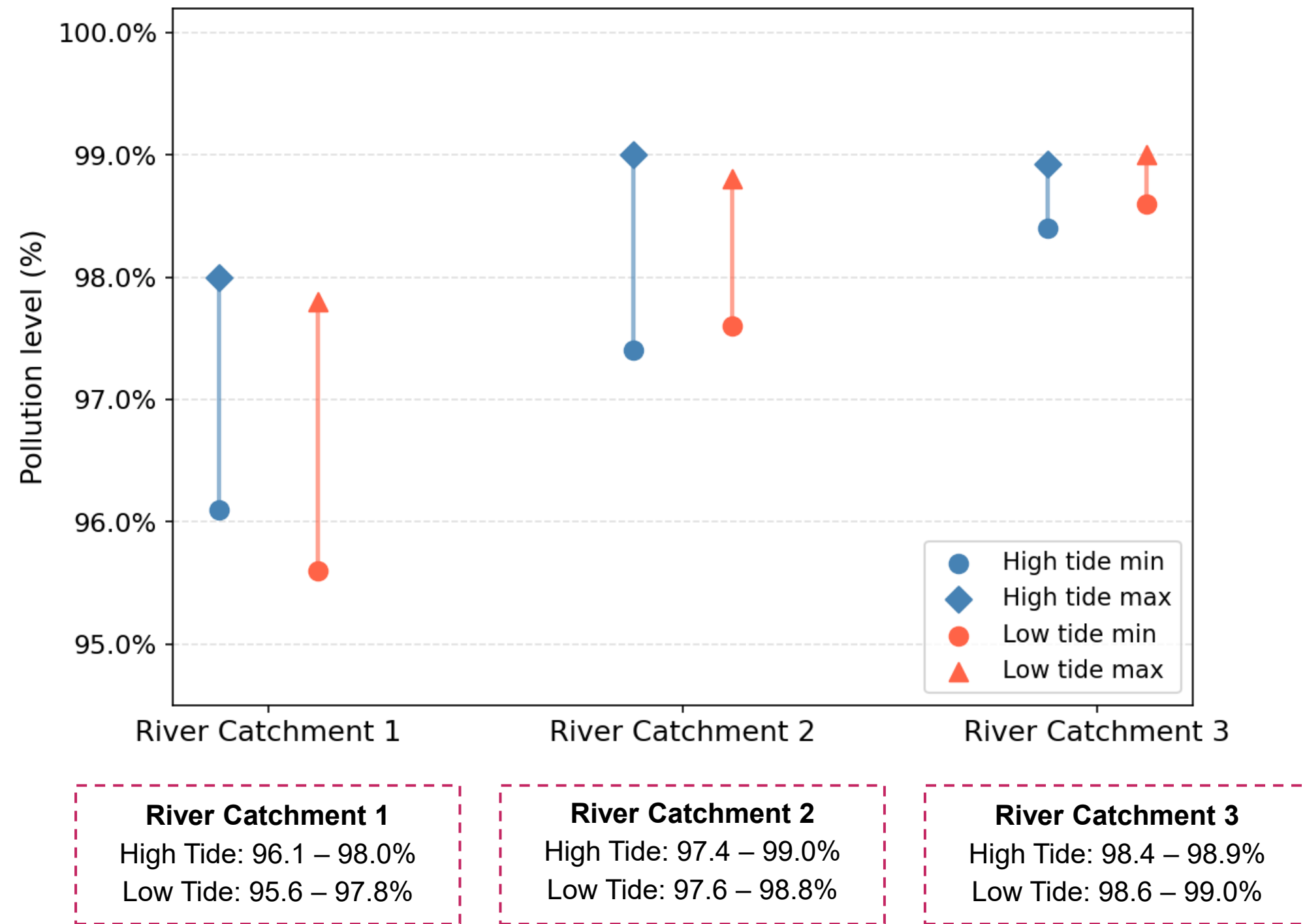


Fig. 17 Pollution status of three river catchments



Results | Data Analysis | Water Quality Status | Groundwater

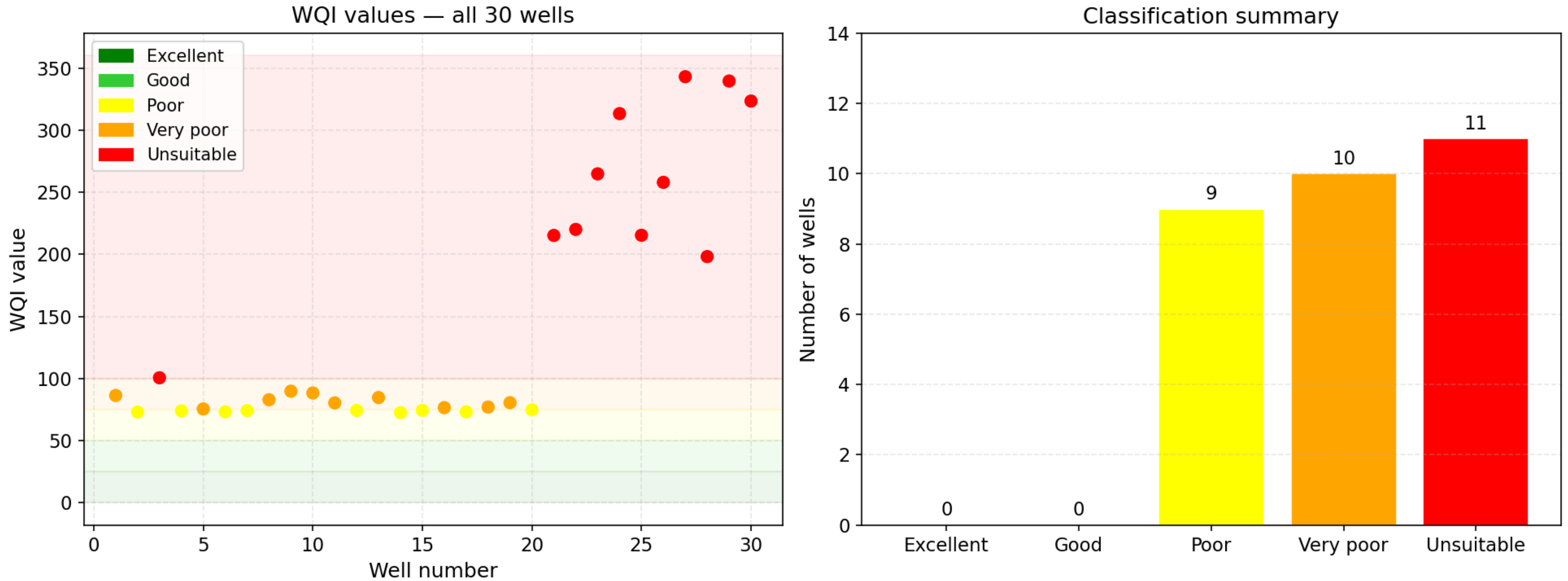


Fig. 18 Water quality status of coastal groundwater



## Results | Data Analysis | Land Use/Land Cover Maps

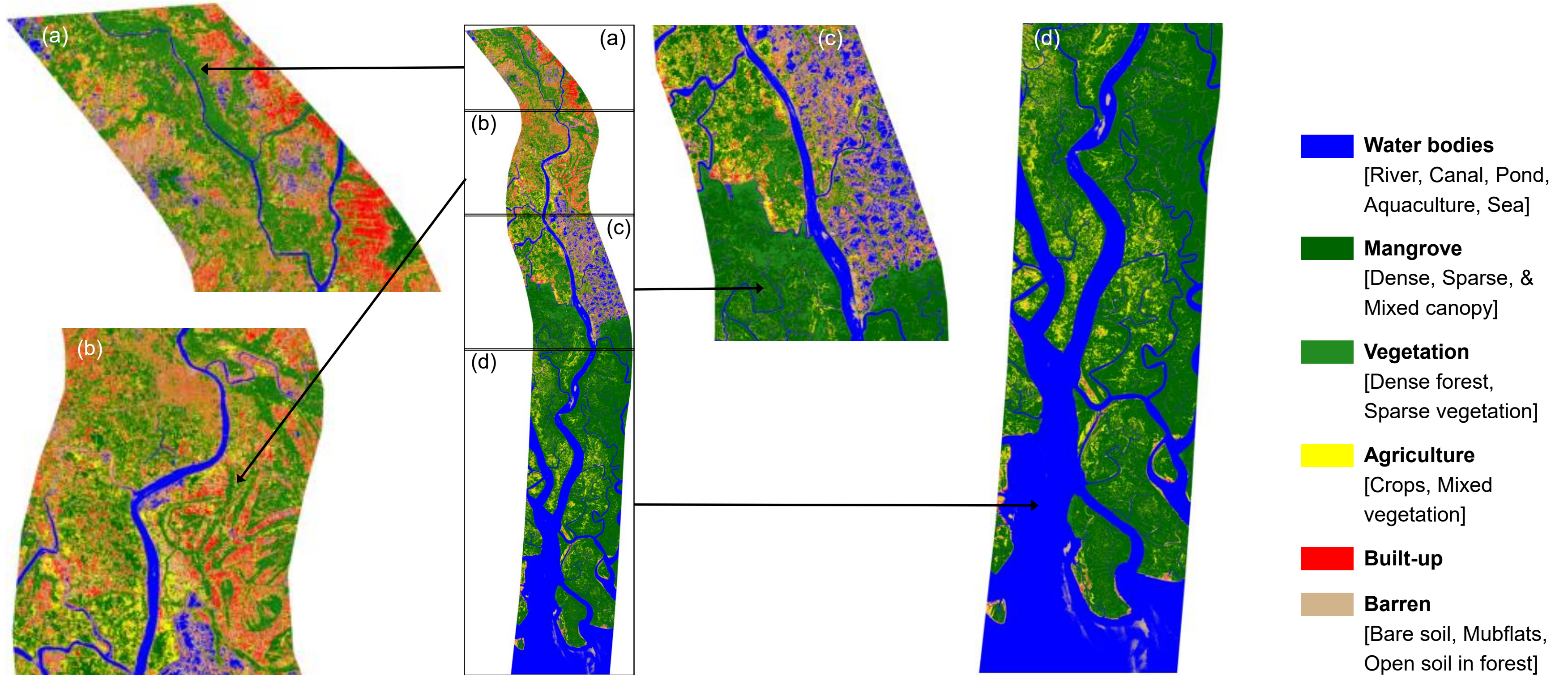


Fig. 19 Land use/Land cover maps (a) River catchment 1, (b) River catchment 2, (c) River catchment 3, and (d) Mangrove forest areas.



## Results | Data Analysis | Climate | River Catchment 1



Fig. 20 Annual rainfall (mm) and annual average temperature (deg.C) in river catchment 1



## Results | Data Analysis | Climate | River Catchment 2



Fig. 21 Annual rainfall (mm) and annual average temperature (deg.C) in river catchment 2



## Results | Data Analysis | Climate | River Catchment 3



Fig. 22 Annual rainfall (mm) and annual average temperature (deg.C) in river catchment 3



## Conclusion

- Variation of selected ion concentrations is higher in river catchment 2, whereas  $\text{NO}_3^-$ -N variation is higher in river catchments 1 and 2.
- River water in both high and low tide, and groundwater chemistry is dominated by  $\text{Na}^+$  and  $\text{Cl}^-$ .
- Regression analysis illustrated significant positive relationship ( $r^2$ ) between  $\text{Na}^+$  and  $\text{Cl}^-$  in river ( $r^2 = 0.99$ ) and groundwater ( $r^2 = 0.93$ ).
- Groundwater salinity in the downstream is higher, a number of samples have a  $\text{Na}^+/\text{Cl}^-$  ratio lower than or close to 1, indicating mixing and seawater intrusion.
- A number of groundwater samples in the upstream area have a  $\text{Na}^+/\text{Cl}^-$  ratio greater than 1, indicating lesser influence of halite dissolution with lower  $\text{Cl}^-$  concentrations, and freshwater conditions.
- Increasing river and groundwater salinity in hydrogeological processes is mostly influenced by evaporation.
- Pollution levels are significantly increased across all river catchments, ranging from 96% to 99%.
- Groundwater quality status is predominantly classified as poor, very poor, and unsuitable for drinking and domestic purposes.